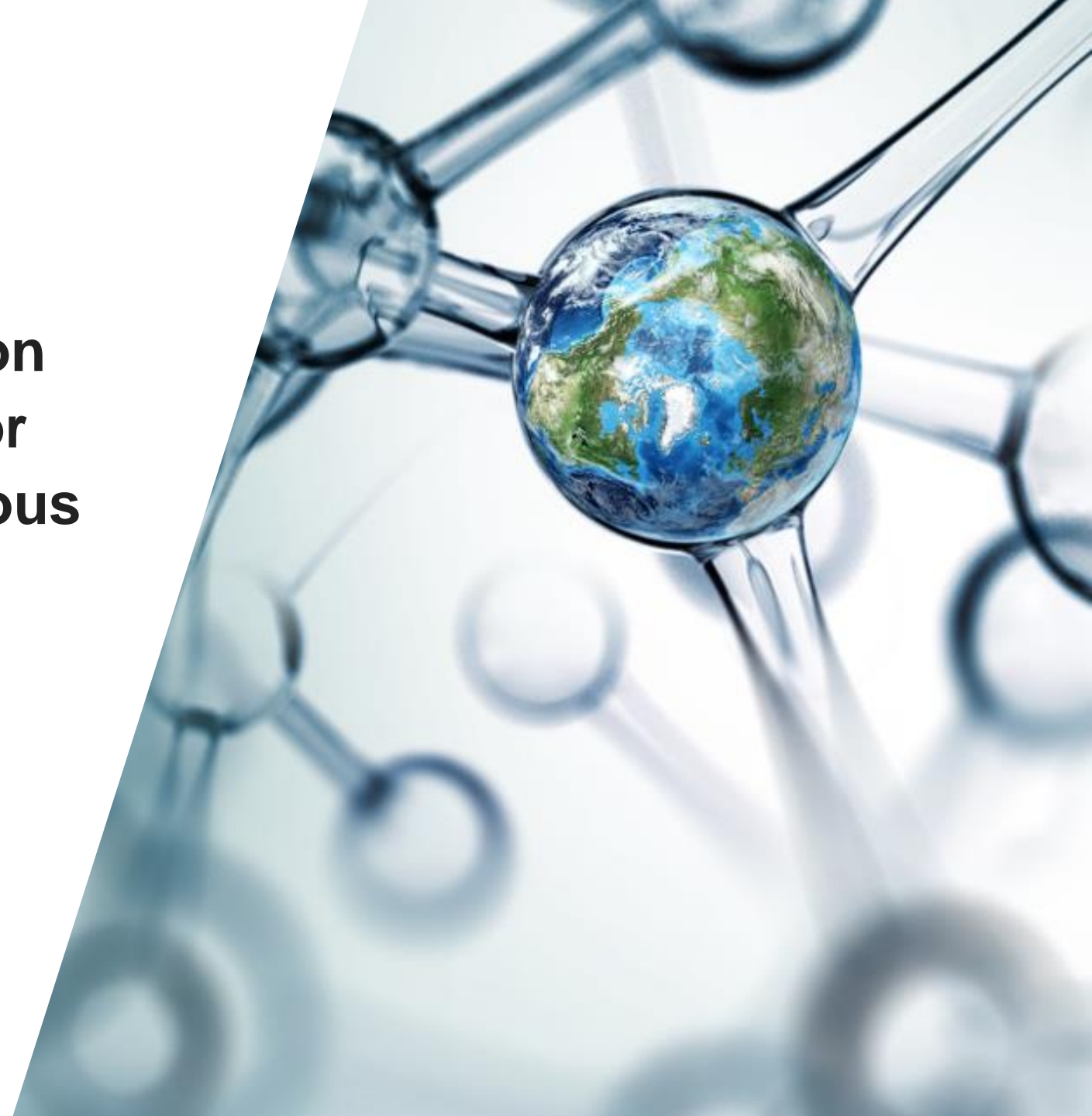


Advances in the design of anion exchange stationary phases for Ion Chromatography with various bonding chemistries

Alexandra Zatirakha, Christopher Pohl

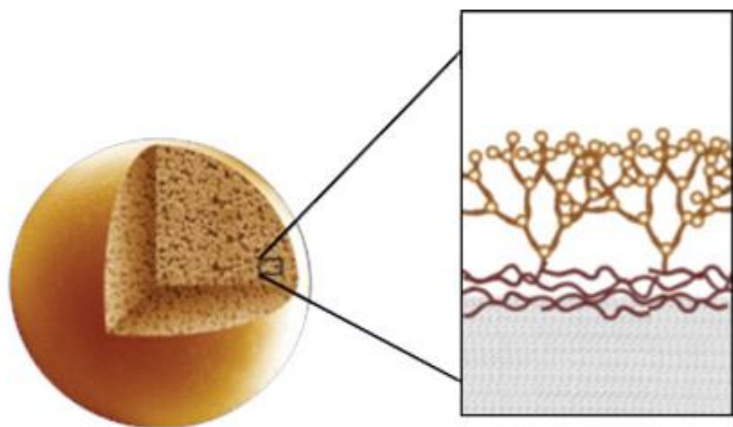
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- Introduction
- Electrostatically bonded hyperbranched anion exchangers
- Covalently bonded hyperbranched anion exchangers
 - Preparation of basement coatings
 - Effect of reaction cycles number
 - Effect of bonding chemistry on selectivity of hyperbranched phases
 - Effect of amine in hyperbranched layer
 - Effect of grafted monomer
- Conclusions

Hyperbranched anion exchangers

- Electrostatically bonded anion exchange materials
- Covered by original patent US 7,291,395
- Represent major portion of ICSP column revenues (at least 15 products)
- Simplified manufacturing process (automated in-column synthesis)
- High performance due to high surface hydrophilicity which minimizes hydrophobic interactions with analytes
- Many possibilities for selectivity variations



(12) **United States Patent**
Pohl et al.

(10) **Patent No.:** **US 7,291,395 B2**
(45) **Date of Patent:** **Nov. 6, 2007**

(54) **COATED ION EXCHANGED SUBSTRATE AND METHOD OF FORMING**

5,532,279 A 7/1996 Barretto et al.
5,865,994 A 2/1999 Riviello et al.
6,074,541 A 6/2000 Srinivasan et al.
6,867,295 B2* 3/2005 Woodruff et al. 536/103

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(73) Assignee: **Dionex Corporation**, Sunnyvale, CA (US)

(*) Notice: Subject to any disclaimer, the term of this patent is extended or adjusted under 35 U.S.C. 154(b) by 536 days.

(21) Appl. No.: **10/782,366**

(22) Filed: **Feb. 18, 2004**

(65) **Prior Publication Data**

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(51) **Int. Cl.**
B32B 5/16 (2006.01)
B05D 7/00 (2006.01)

(52) **U.S. Cl.** **428/407**; 427/221; 427/222

(58) **Field of Classification Search** 428/403, 428/407; 427/212, 221, 222
See application file for complete search history.

(56) **References Cited**

OTHER PUBLICATIONS

Alpert, A., et al., "Preparation of a porous microparticulate anion-exchange chromatography support for proteins," *J. Chromatogr.* 185:375-392 (1979).
Kopaciewicz, W., et al., "Stationary phase contributions to retention in high-performance anion-exchange protein chromatography: ligand density and mixed mode effects," *J. Chromatogr.* 318:157-172 (1985).

* cited by examiner

Primary Examiner—H. T Le
(74) *Attorney, Agent, or Firm*—David J. Brezner; Morgan, Lewis & Bockius LLP

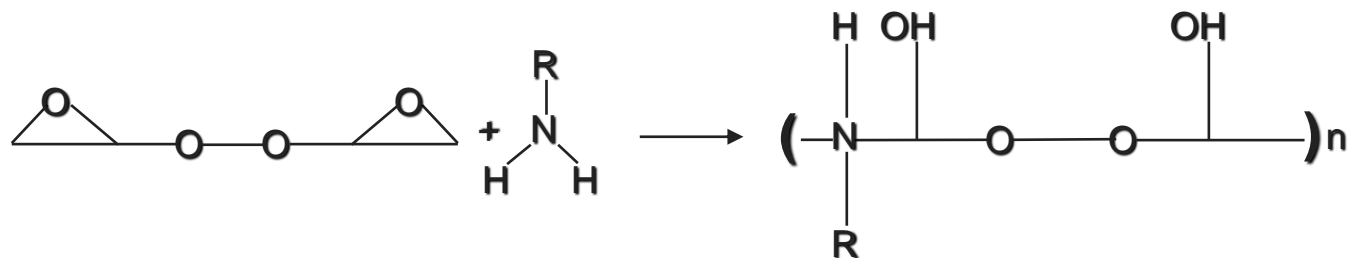
(57) **ABSTRACT**

A method for making an ion exchange coating (e.g., a chromatographic medium) on a substrate comprising (a) reacting at least a first amine compound comprising amino groups, with at least a first polyfunctional compound, in the presence of a substrate to form a first condensation polymer reaction product, with a first unreacted excess of either at least said first amino group or polyfunctional compound functional moieties, irreversibly attached to the substrate, and (b) reacting at least a second amine compound or at least

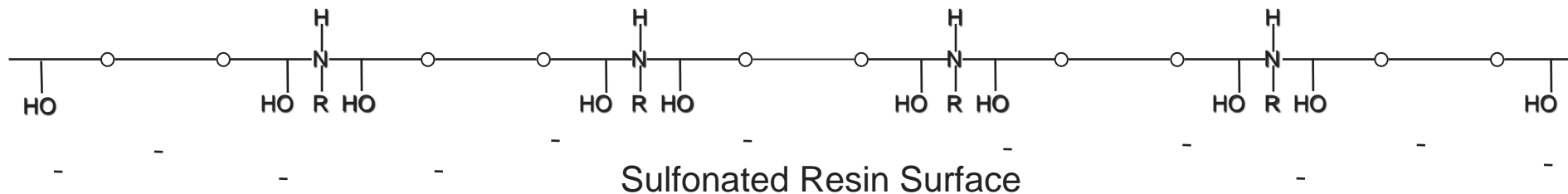
New selectivity → covalent attachment of hyperbranched layer → functionalized resin surface is required

Formation of electrostatically bonded basement coating

Hypothetical product of 1:1 ratio (diepoxide:amine)

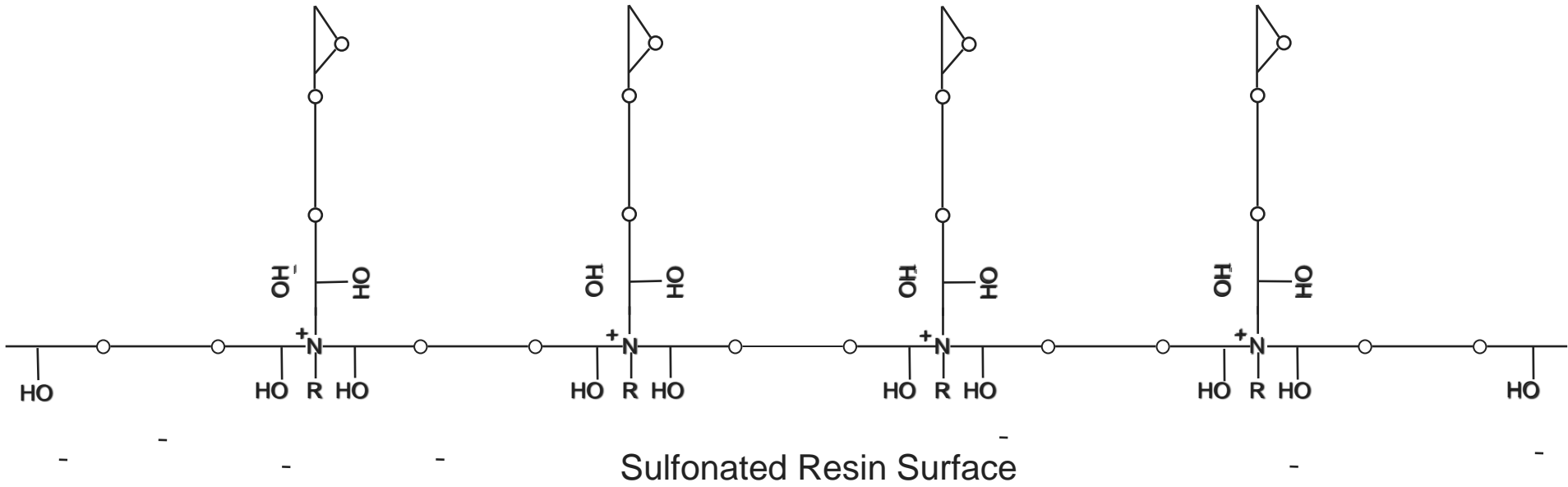


Basement coating [1:1 ratio (diepoxide:amine)]



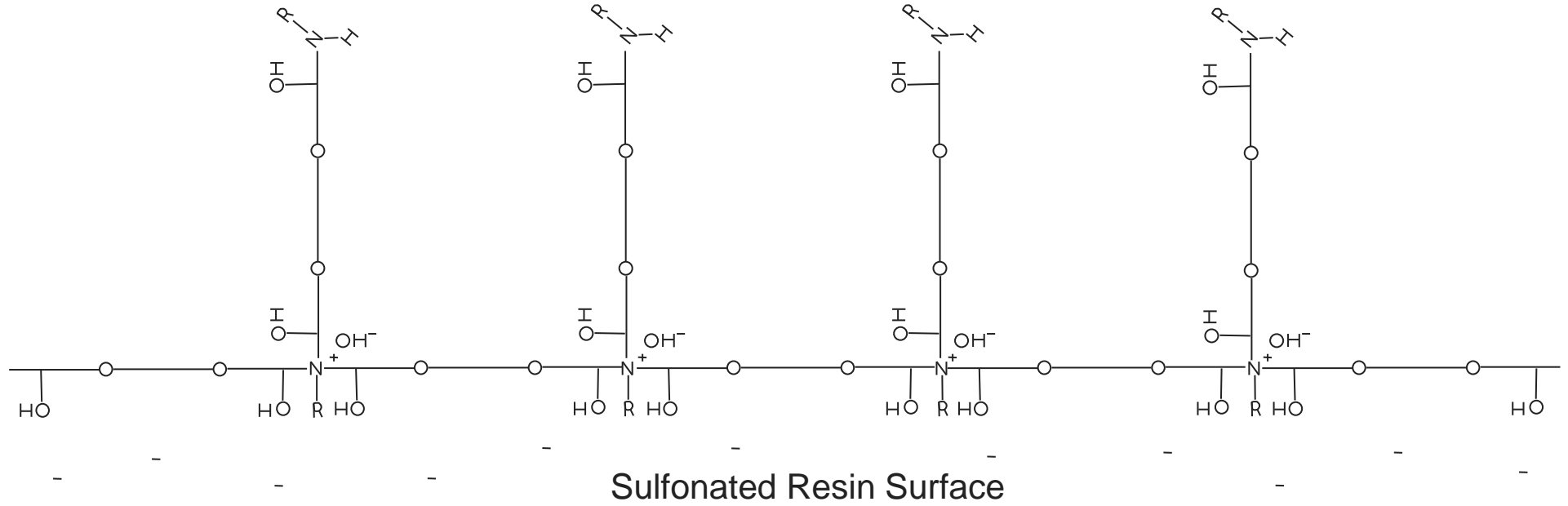
Step-growth polymer schematic

Layer 1 after diepoxide treatment



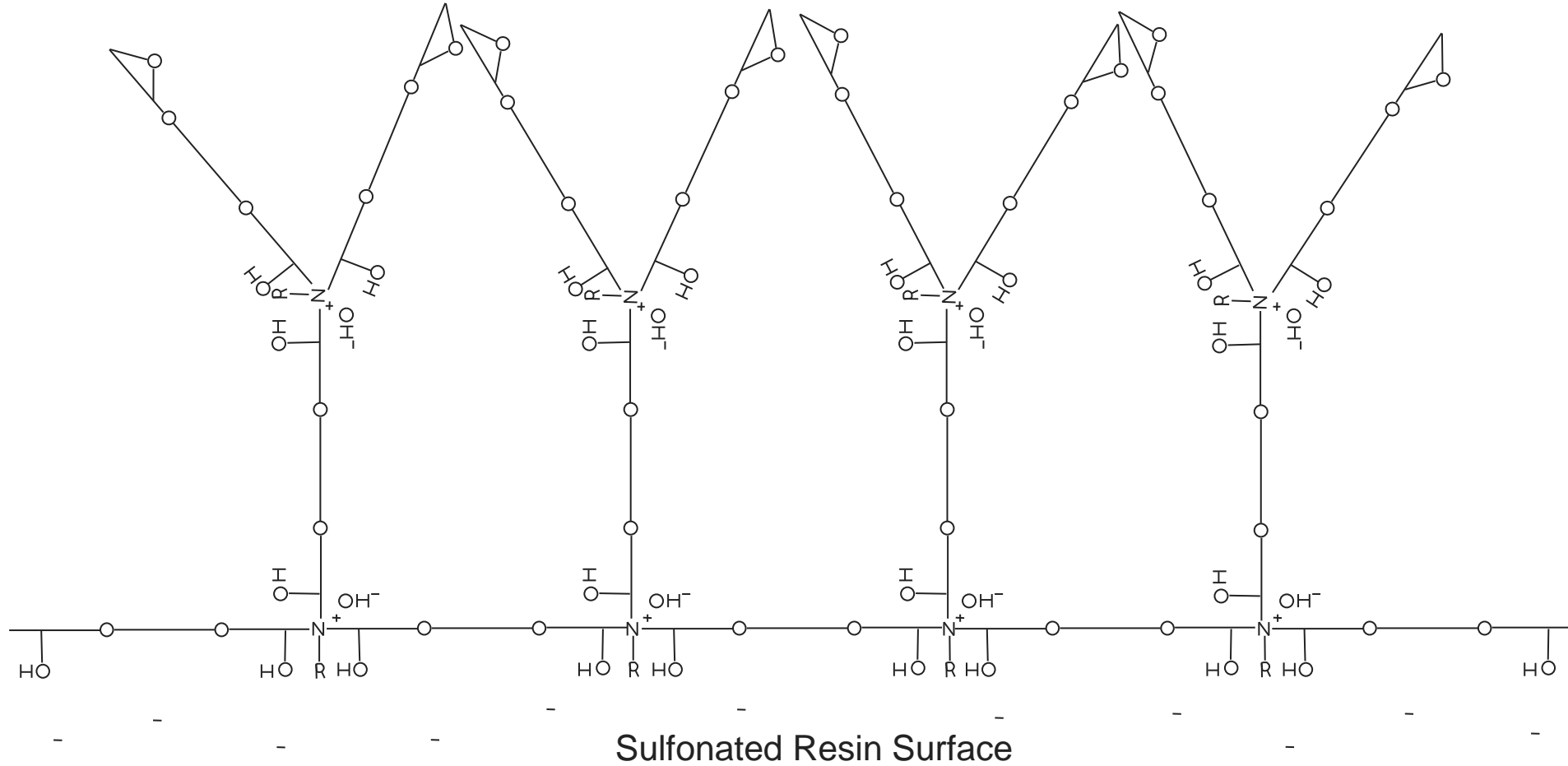
Step-growth polymer schematic

Layer 1 after diepoxide and amine treatment



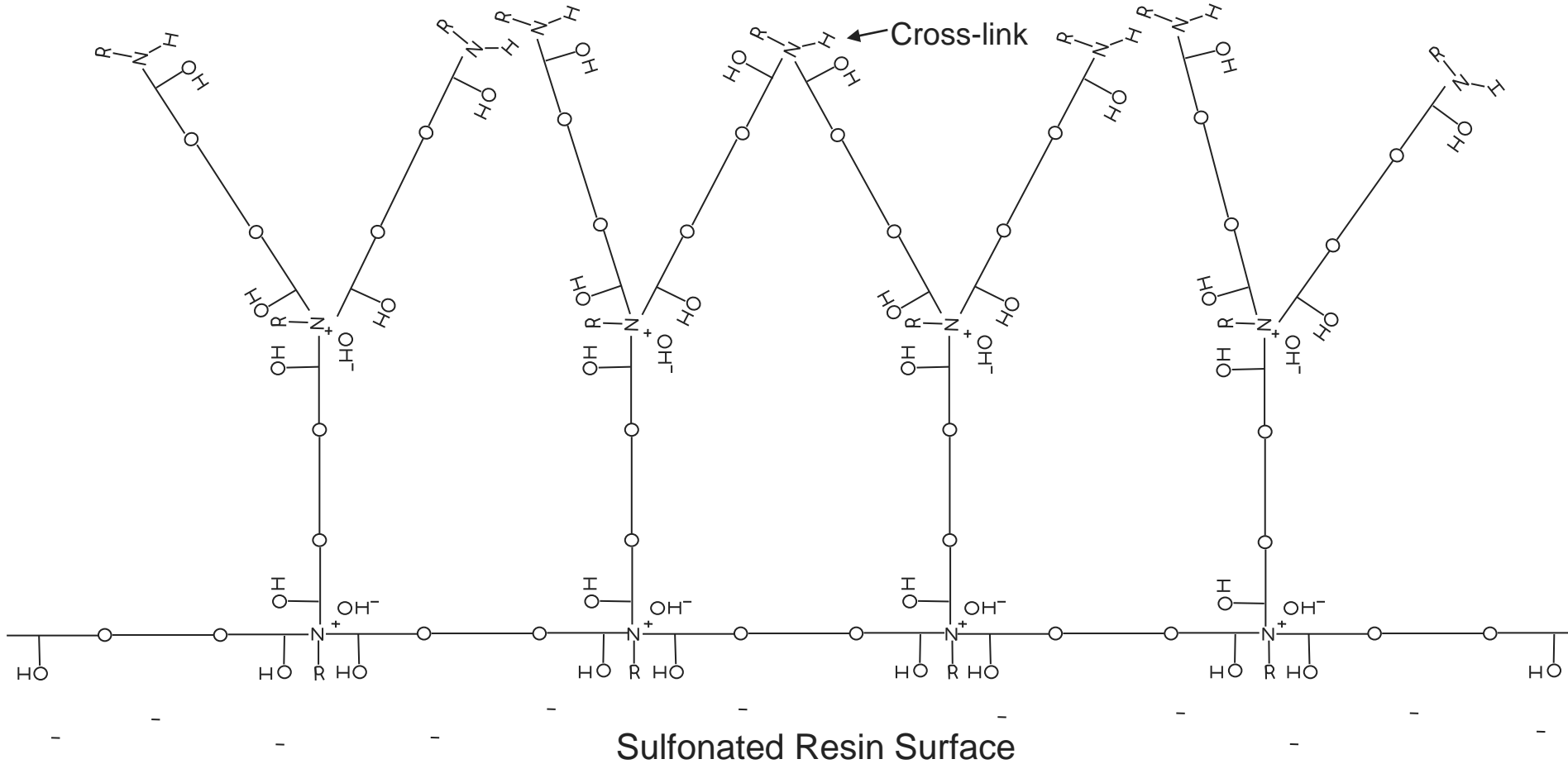
Step-growth polymer schematic

Layer 2 after diepoxide treatment

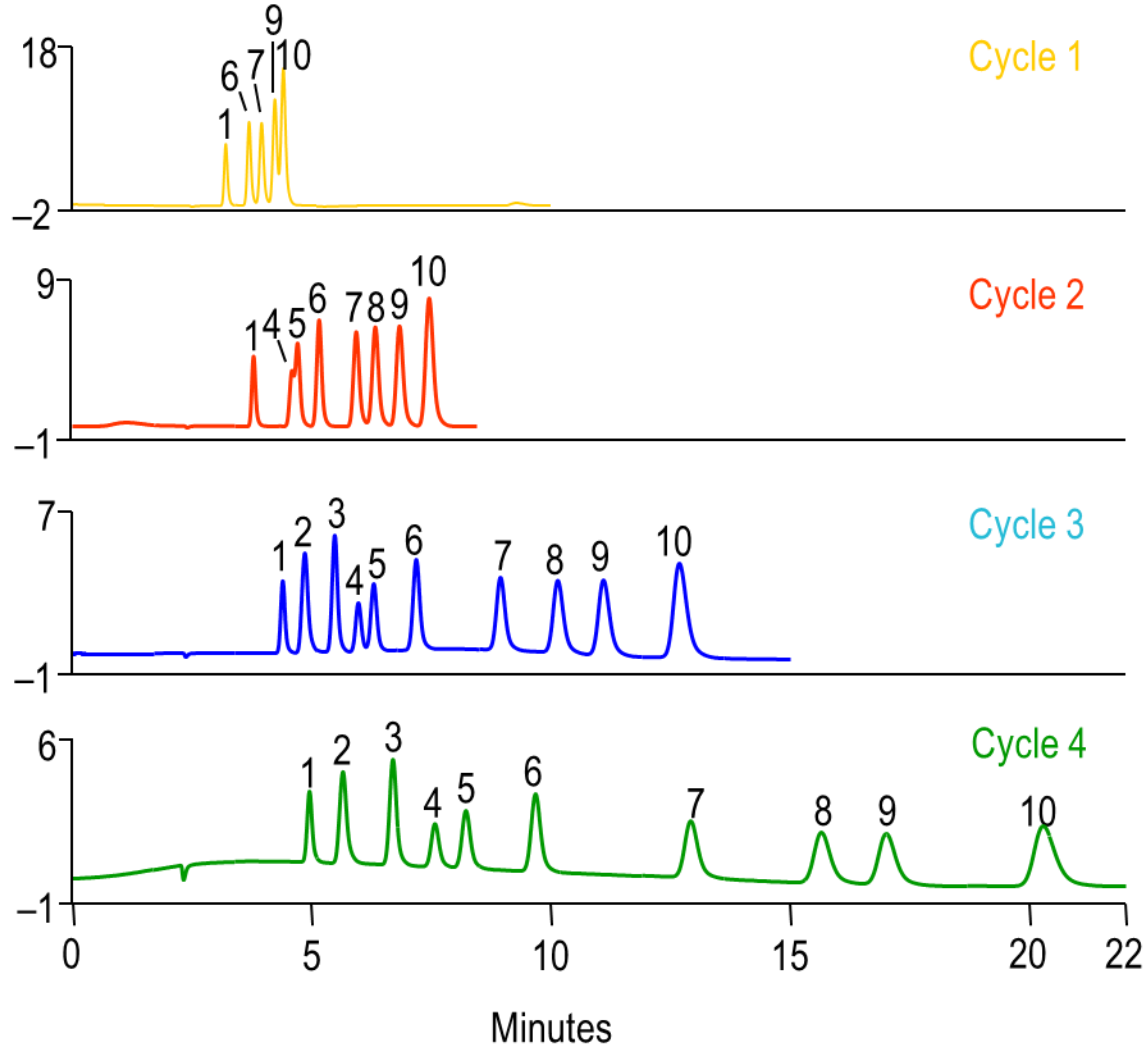


Step-growth polymer schematic

Layer 2 after diepoxide and amine treatment



In-column preparation of step-growth electrostatic graft



Column: Prototype prepared using MA and 1.4-BDDGE
Eluent: 5 mM KOH
Flow Rate: 1 mL/min
Inj. Volume: 25 μ L

Peaks:

1. Fluoride	1 ppm
2. Acetate	10
3. Formate	5
4. Chlorite	5
5. Bromate	10
6. Chloride	3
7. Nitrite	5
8. Chlorate	10
9. Bromide	10
10. Nitrate	10

Covalently bonded anion exchangers

Major requirements:

- Functionalization should be limited to the resin surface
- Functionalization method should provide good surface hydrophilization

Solutions for covalent attachment proposed in literature:

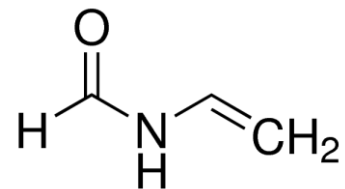
- Chemical derivatization of substrate - difficult to accomplish surface modification → poor stationary phase performance
- Incorporation of a reactive monomer as a comonomer in resin synthesis - uneven distribution of functional groups inside the particle → poor efficiency of the stationary phase.

Alternative solutions:

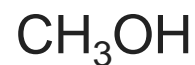
- To use reagents (monomers) that are not soluble in the resin
- To use the solvent that doesn't cause resin swelling (highly polar solvents for PS-DVB functionalization)

Formation of covalently bonded basement coating

Highly Polar Monomer



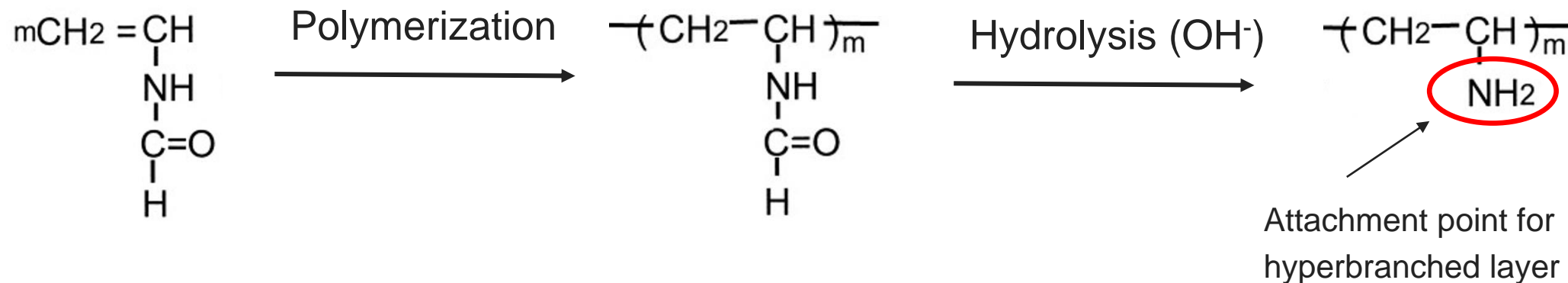
Polar Solvent



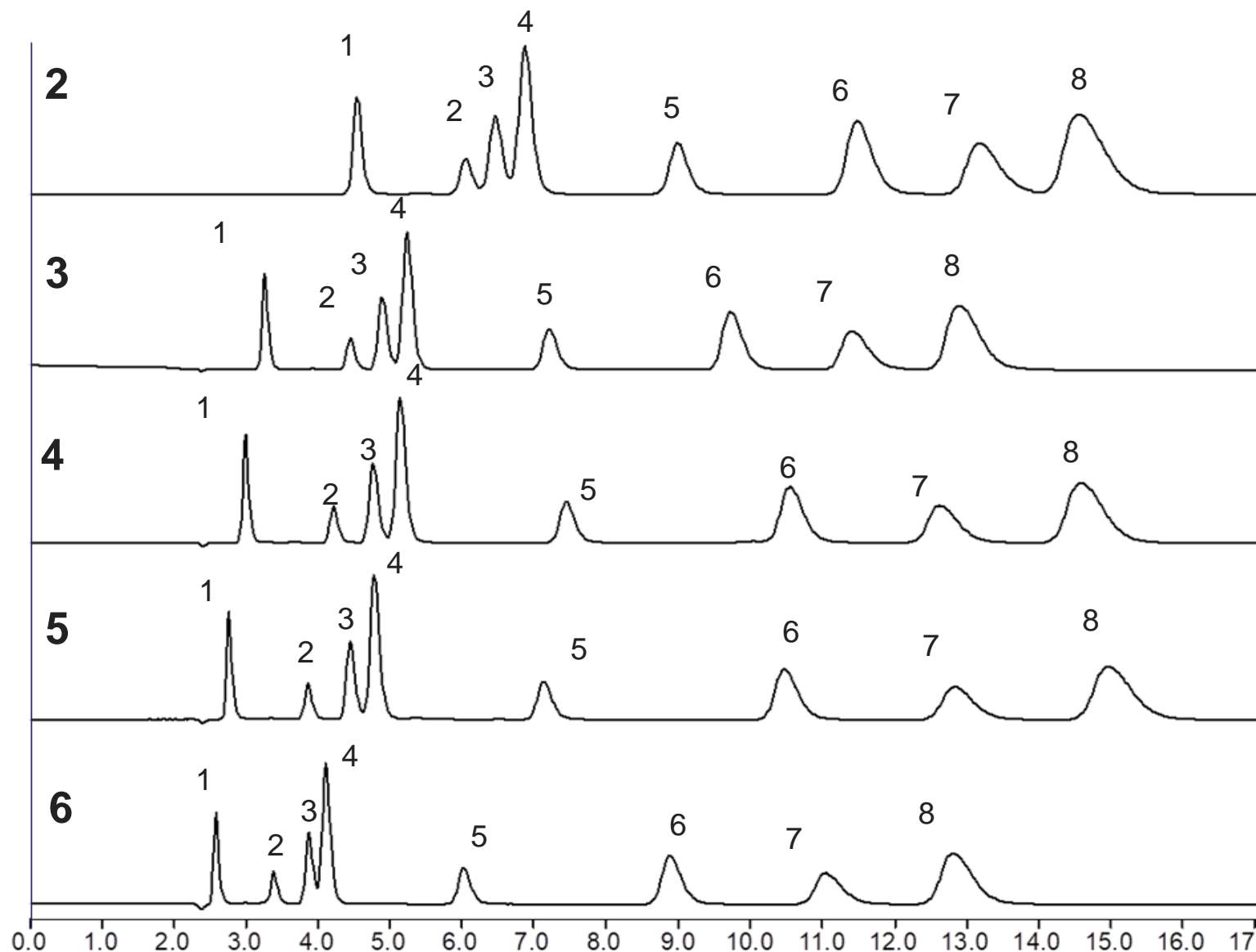
Initiator

AIBN or ACVA

N-Vinylformamide polymerization and hydrolysis



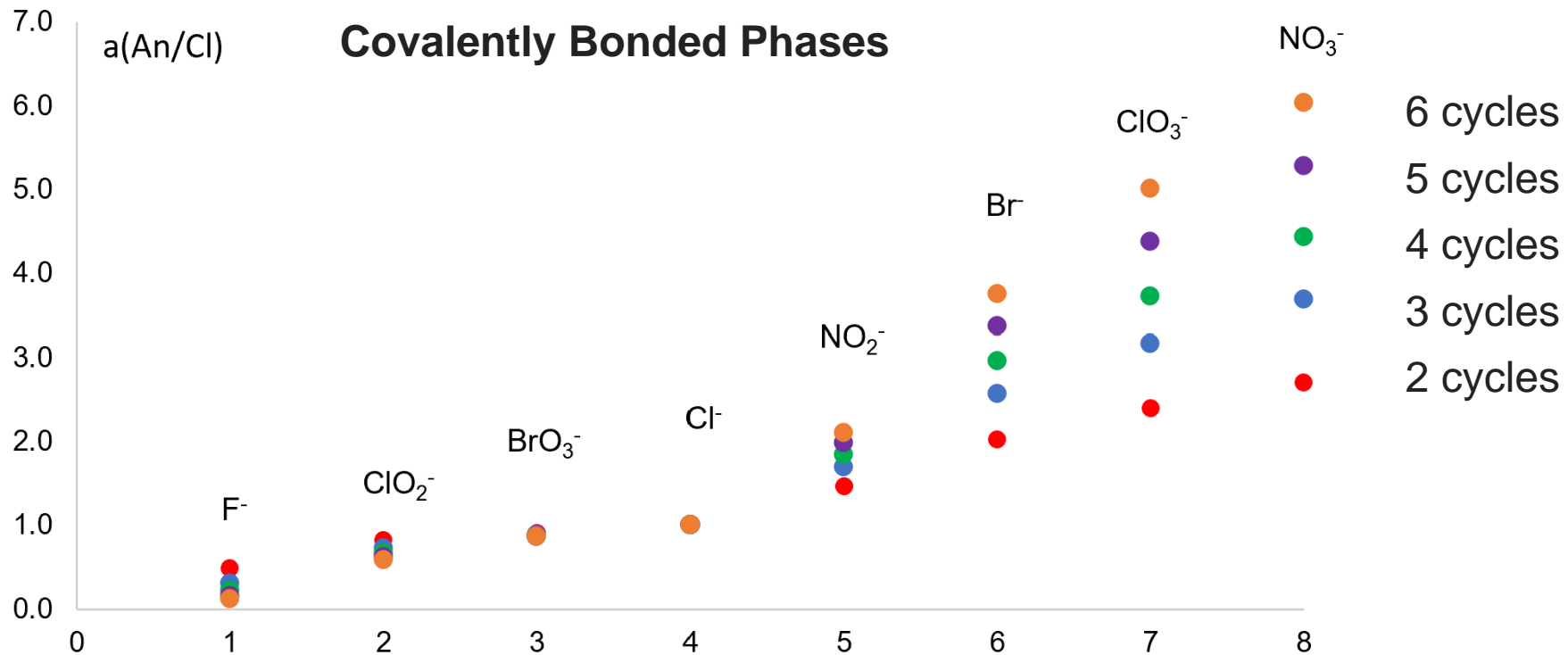
Effect of reaction cycles number on separation



Parameter	Analyte
N of cycles	2 – 6
Diepoxide	1,4-BDDGE
Amine	MA
Columns	250 mm x 4 mm i.d.
Flow	1.0 mL/min

Peak	Analyte
1	F ⁻
2	ClO ₂ ⁻
3	BrO ₃ ⁻
4	Cl ⁻
5	NO ₂ ⁻
6	Br ⁻
7	ClO ₃ ⁻
8	NO ₃ ⁻

Effect of reaction cycles number on selectivity

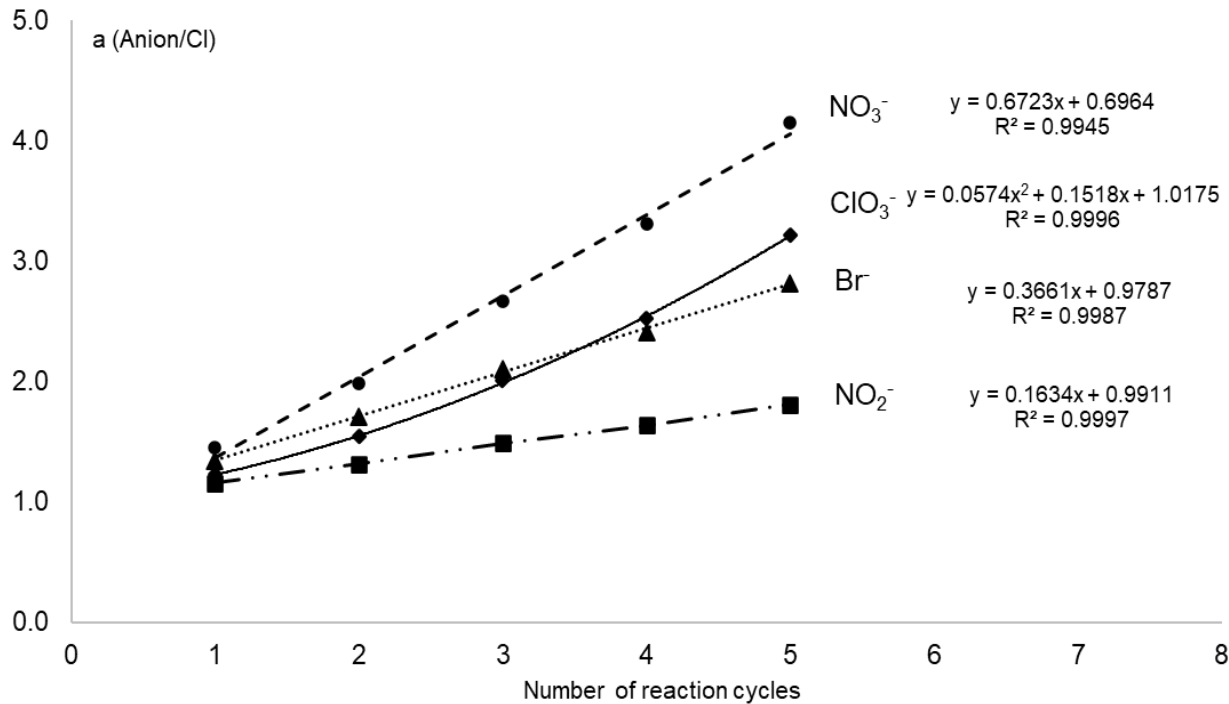


N of cycles	$a(\text{ClO}_3/\text{Br})$	$a(\text{NO}_3/\text{ClO}_3)$	$a(\text{NO}_3/\text{Br})$	$\frac{t_r(\text{NO}_3)-t_r(\text{ClO}_3)}{t_r(\text{ClO}_3)-t_r(\text{Br})}$
2	1.19	1.13	1.34	0.82
3	1.23	1.17	1.43	0.89
4	1.26	1.19	1.50	0.93
5	1.30	1.21	1.57	0.89
6	1.33	1.20	1.61	0.81

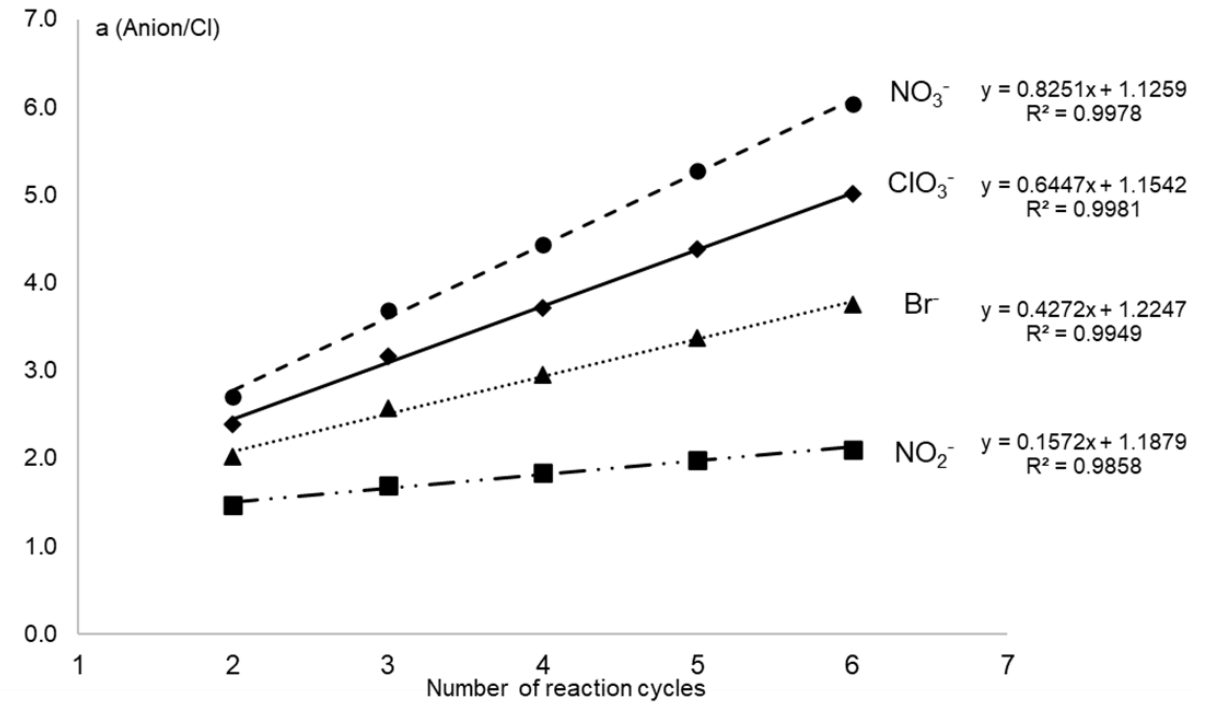
Effect of bonding chemistry on selectivity

Reaction cycles: 1,4-BDDGE + methylamine

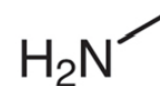
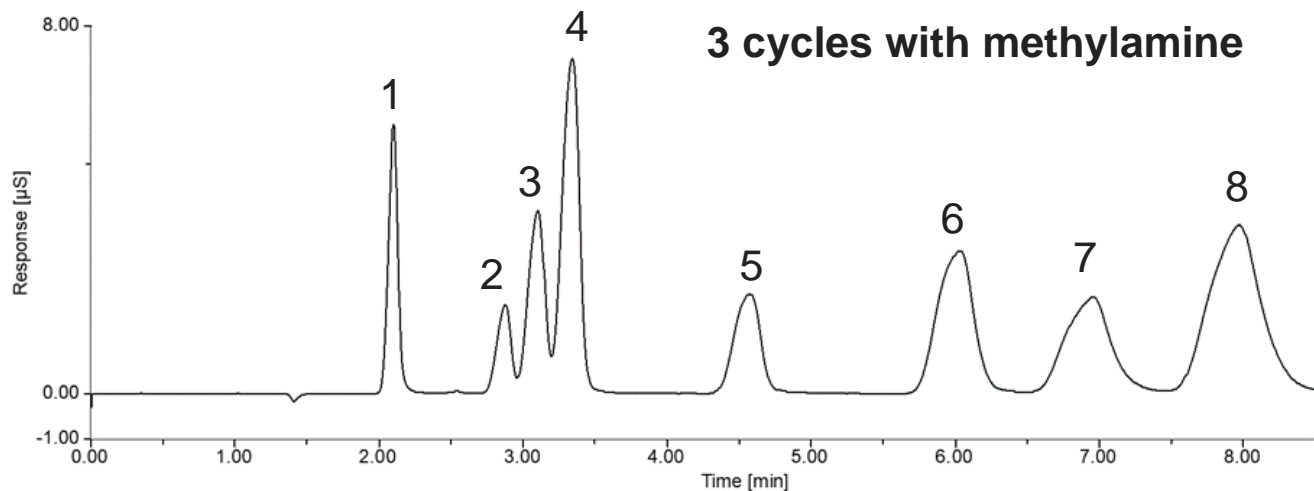
Electrostatically Bonded



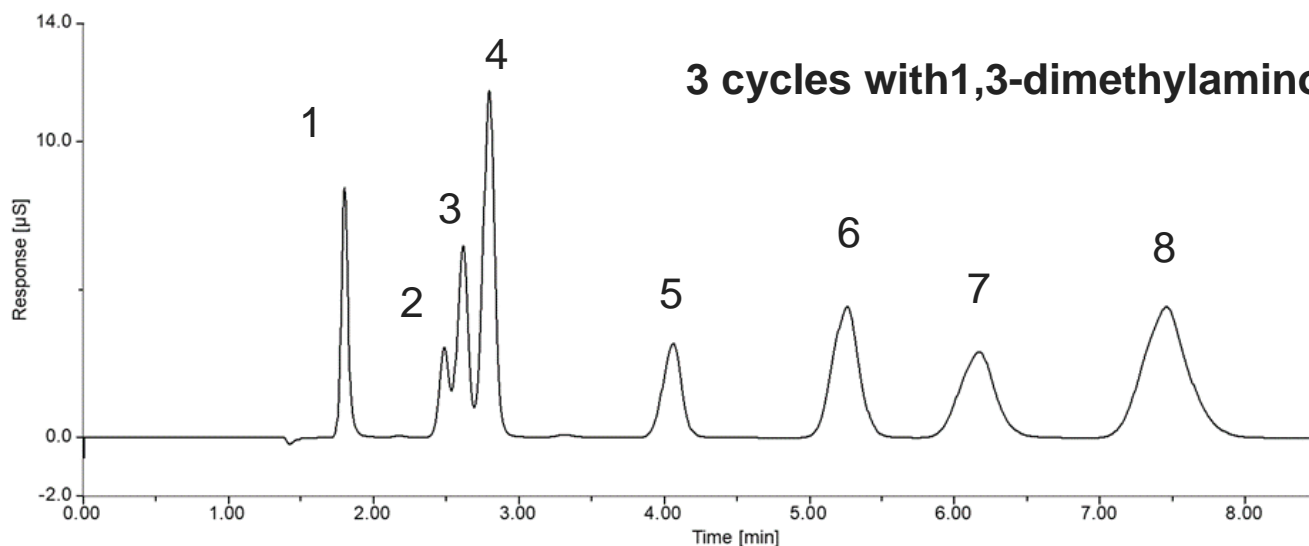
Covalently Bonded



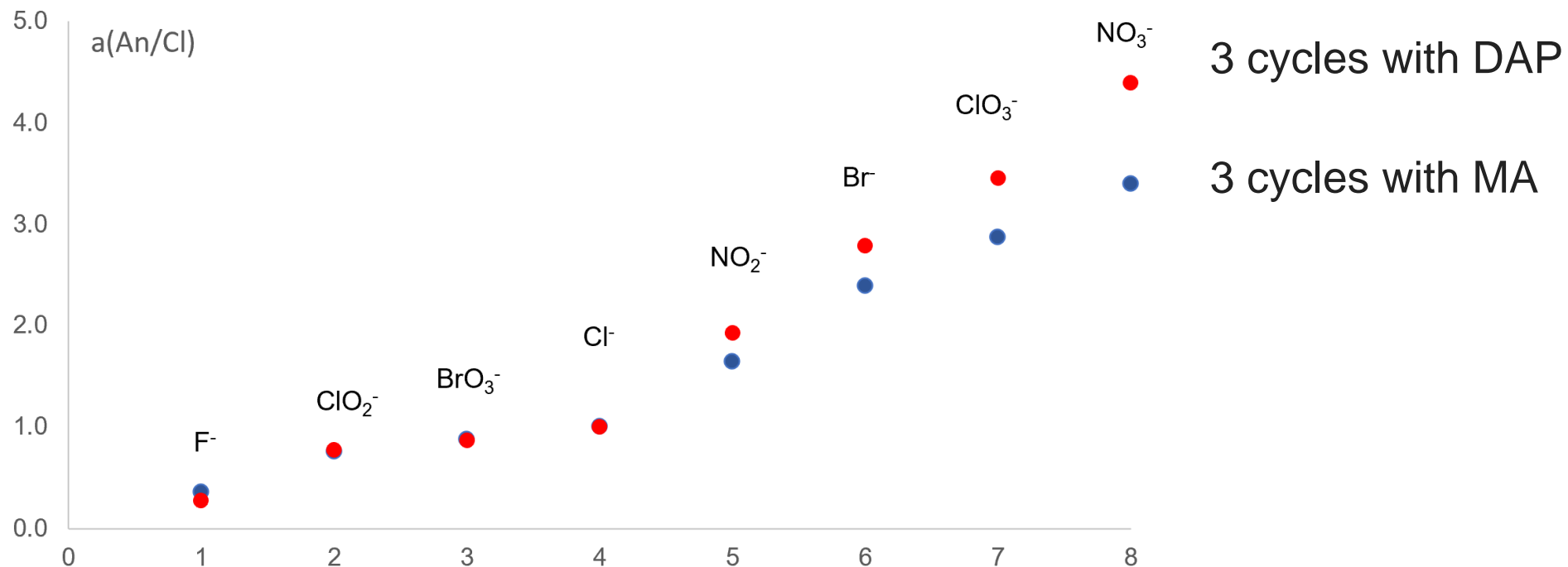
Effect of amine in reaction cycle



Parameter	Value
N of cycles	3
Diepoxide	1,4-BDDGE
Amine	MA
Columns	250 mm x 4 mm i.d.
Eluent	5 mM KOH
Flow	1.0 mL/min



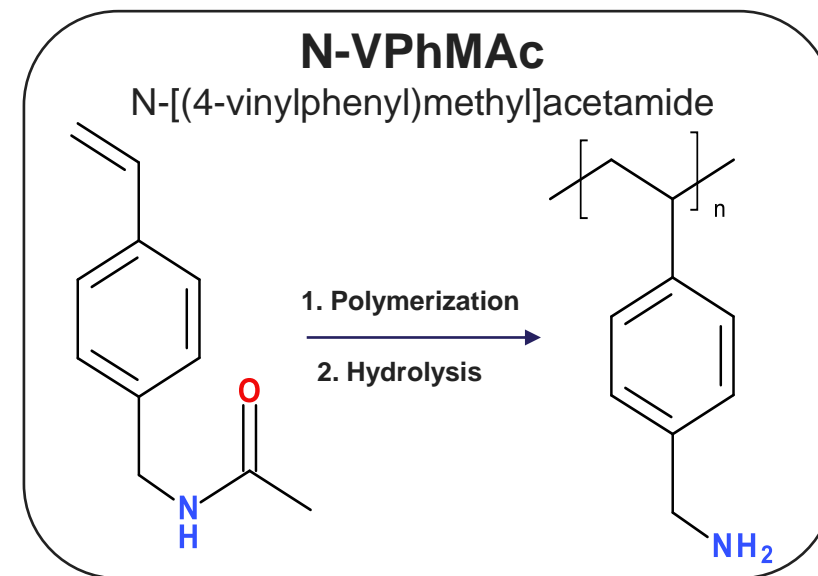
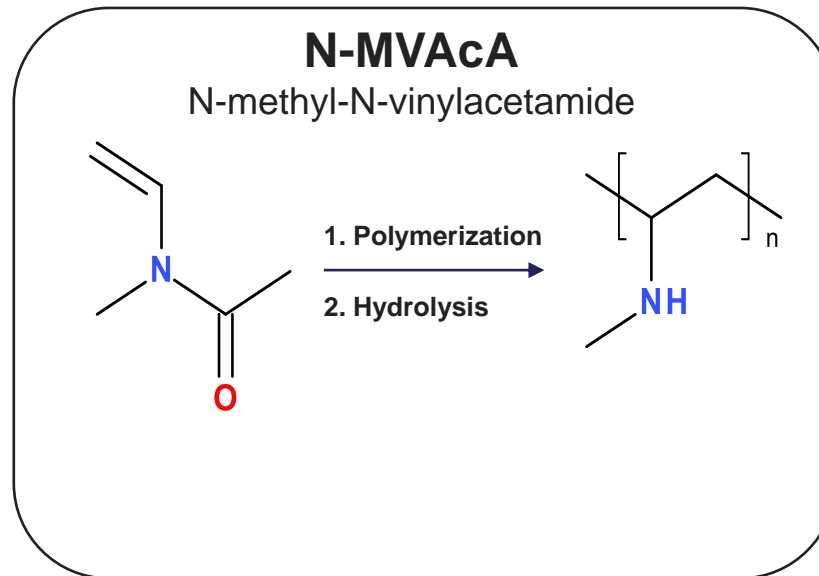
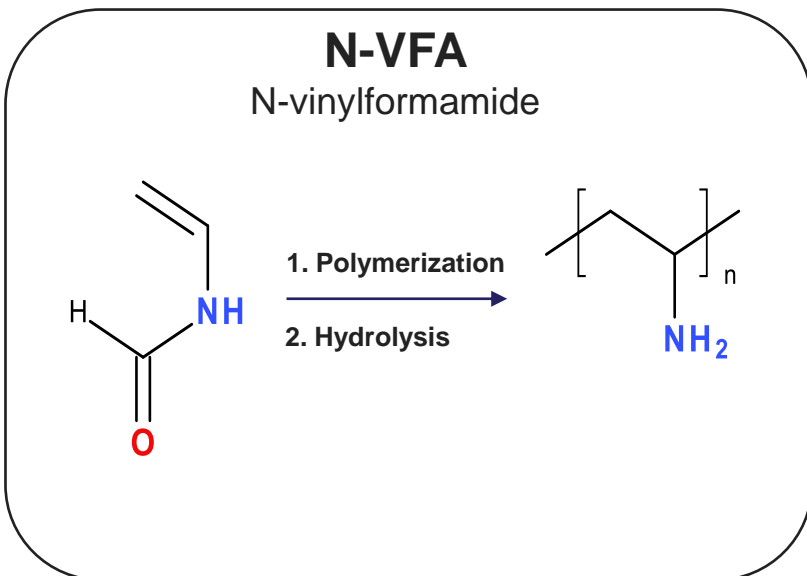
Effect of amine in reaction cycle on selectivity



N cycles/Amine	a(ClO ₃ /Br)		a(NO ₃ /ClO ₃)		a(NO ₃ /Br)	
	MA	DAP	MA	DAP	MA	DAP
3 cycles	1.20	1.24	1.18	1.27	1.42	1.57
4 cycles	1.26	1.26	1.19	1.28	1.50	1.62
5 cycles	1.30	1.30	1.21	1.29	1.57	1.67

Effect of monomer used for grafting

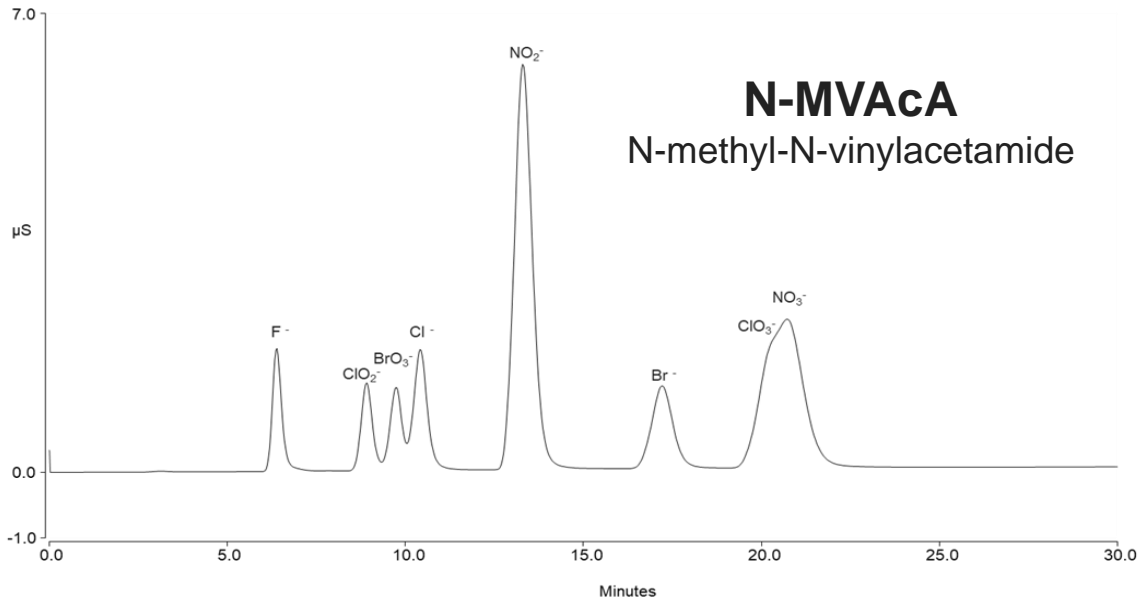
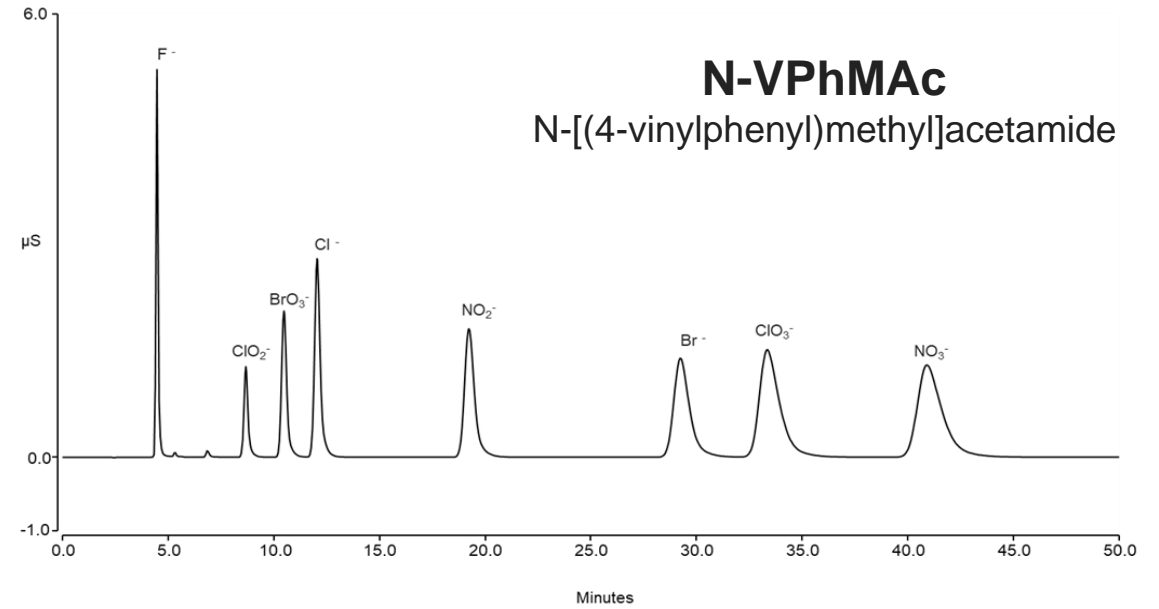
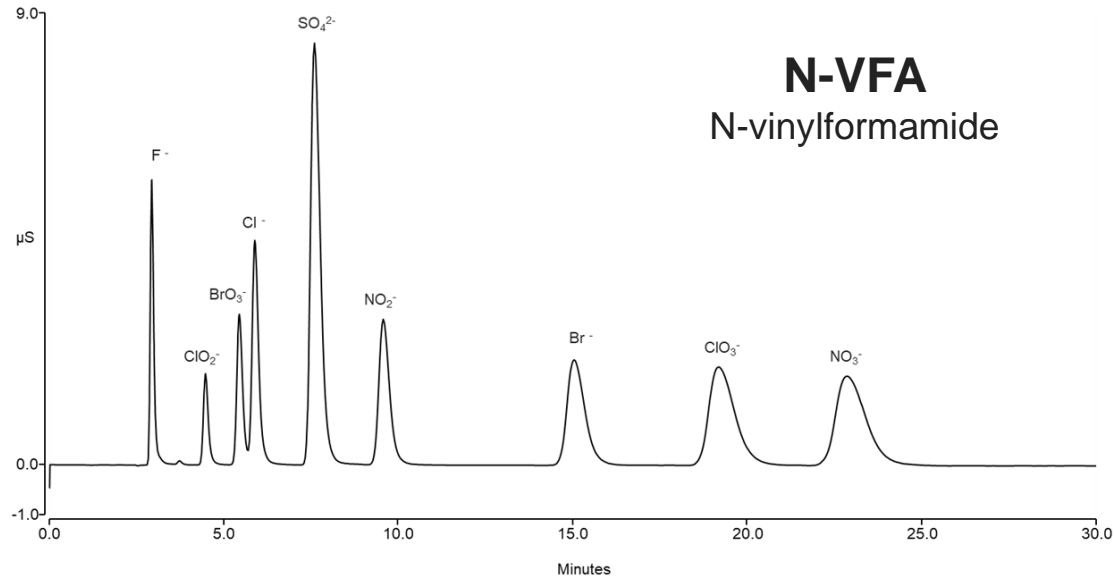
Monomers in the base layer



Monomer in the base layer	*Capacity after base layer (μEq)
N-vinylformamide	91.2
N-methyl-N-vinylacetamide	1.7
N-[(4-vinylphenyl)methyl]acetamide	22.9

*Per 250 mm x 4 mm i.d. column

Effect of monomer used for grafting



Monomer in the base layer	N of cycles/ capacity	Eluent
N-VFA	6 / 154 μeq	20 mM KOH
N-MVAcA	6 / 35 μeq	1 mM KOH
N-VPhMAc	5 / 127 μeq	5 mM KOH

Conclusions

- The proposed grafting approach allows one to limit the modification of the resin to the surface thus preparing covalently bonded hyperbranched phases with high chromatographic performance
- Proposed method of functional layer attachment allows for the preparation of hyperbranched anion exchangers with new selectivities and controlled crosslinks throughout the layer
- Elution order for anions on the covalently bonded hyperbranched anion exchangers is not dependent on the number of the reaction cycles and amine structure used for hyperbranching
- The most effective way to influence selectivity of covalently bonded phases is by changing the structure of grafted monomer and grafting conditions

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