

### ASMS 2017 MP 263

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Estimation of Elemental Compositions for Additives in Polymers Using Newly Developed EI/CI Ion Source Without Venting MS

### Introduction

• We have developed a new combined EI/CI ion source that removes the need of physical exchange between an electron ionization (EI) source and chemical ionization (CI) source.

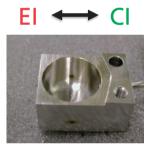
This reduces system downtime and allows uninterrupted analysis using both El and Cl.

### El Method

Qualitative analysis based on mass pattern information.

### CI Method

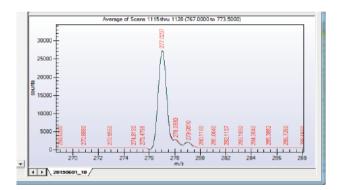
Obtains information about the molecular weights.



 MassWorks software (Cerno Biosciences) allows accurate mass information to be obtained from mass profile data using quadrupole MS. Elemental compositions can be estimated based on molecular-related ions.



	Formula	Mono Isotope	Mass Error (mDa)	Mass Error (PPM)	Spectral Accuracy
1	C11H2N8P	277.0135	10.2450	36.9823	98.6127
2	C13H12OP3	277.0096	14.1490	51.0752	98.6066



# **Experimental Procedure**

- 1. Evaluating the New El/Cl Ion Source Performance We analyzed a mixed standard solution of phosphate flame-retardant plasticizing agents (10 ppm, solvent: acetone) by liquid injection-GC/MS using three different ion sources (El-specific ion source, Cl-specific ion source, and the new combined El/Cl ion source).
- 2. Evaluating the Elemental Composition of a Standard Sample Estimated by MassWorks A sample of approximately 0.5 mg was cut from a polymer standard sample containing seven phthalate esters (DIBP, DBP, BBP, DEHP, DNOP, DINP, and DIDP, each at 1,000 mg/kg, PE, Shimadzu). The sample was analyzed by spot CI, using a Py-GC/MS system with the new El/CI ion source installed.

MassWorks was then used to estimate an elemental composition from the molecular-related ions (M+H) detected, and the accuracy of each estimated phthalate ester was evaluated.

3. Estimating Chemical Compounds of Unknown Peak Detected by Py-GC/MS Analysis of Polymers A sample of approximately 0.5 mg was cut from a real-world polymer sample (cable sheath). The sample was analyzed using EI and CI by a Py-GC/MS system with the new El/CI ion source installed. Chemical compound in unknown peak was also estimated using MassWorks to perform elemental composition analysis.

## Analytical Conditions

Instrument			
Pyrolyzer	: EGA/PY-3030D multi-shot pyrolyzer		
	+ AS-1020E auto-shot sampler (Frontier Lab.)		
GCMS	: GCMS-QP2020 (Shimadzu)		
Pyrolysis			
Analysis Mode	: Heart-Cut EGA (thermal extraction)		
Thermal Decomposition Temp.	: 200 °C - 20 °C/min - 300 °C - 5 °C/min – 340 °C (1 min)		
ITF Temp.	: 300 °C (manual)		
GC-MS			
Column	: UA-PBDE (15 m × 0.25 mm l.D., df = 0.05 μm)		
Oven Temp.	: 80 °C -20 °C/min - 300 °C (5 min)		
Carrier Gas	: He (linear velocity = 52.1 cm/sec)		
Injection Method	: Split (1:50)		
Injection Port Temp.	: 280 °C		
ITF Temp.	: 320 °C		
Ion Source Temp.	: 230 °C		
Ionization	: El, Cl (reagent gas isobutane)		
Measurement mode	: Scan		

### Spot Cl

Spot CI is a method where the filament is only turned ON while detecting the target peak (the peak to be analyzed for molecular weight information), but is turned OFF at other times.

The merits of this method are a reduced loss of ion source sensitivity and reduced attrition of the filament.

#### EGA/PY-3030D + AS-1020E or AOC-20i+s

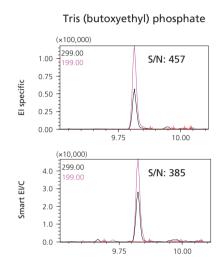


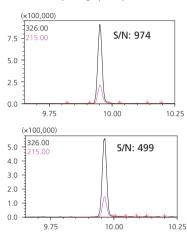
GCMS-QP2020



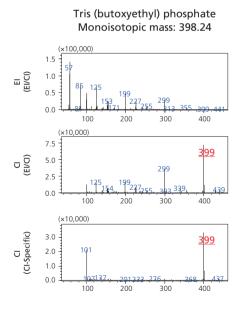
# Analysis Results

- 1. Evaluating the New El/Cl Ion Source Performance
- Comparison of SIM Sensitivity of EI Methods

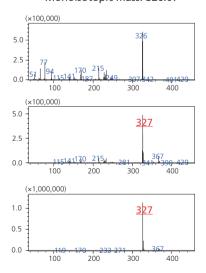




### Comparison of Mass Spectra of CI Methods



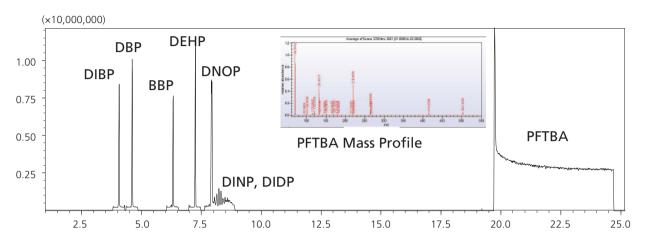
Triphenyl phosphate Monoisotopic mass: 326.07



Triphenyl phosphate

Venting MS

2. Evaluating the Elemental Composition of a Standard Sample Estimated by MassWorks

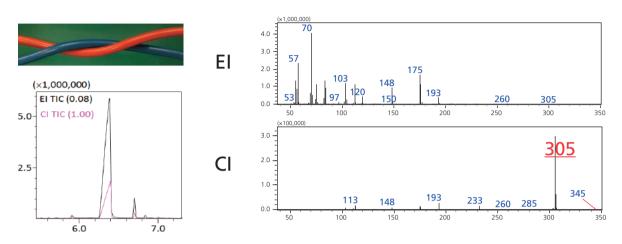


TIC of Polymer Standard Sample Containing Seven Phthalate Esters (1,000 mg/kg, Spot Cl)

### • Results of Composition Estimation

Without consideration of H desorption With consideration of H desorption Name of Spectral Mass Error Ranking Ranking Formula Monoisotope Compound Accuracy (mDa) 10 DIBP C16O4H23 279.1591 1 96.5385 -1.8856 5 DBP C16O4H23 279.1591 1 97.2037 -2.8856 C1904H21 313.1434 96.822 18.3644 6 BBP 1 1 6 DEHP C24O4H39 391.2843 95.4571 8.6138 5 DNOP C24O4H39 391.2843 1 95.6035 -0.7862 1 DINP C26O4H43 419.3156 1 94.7035 3.4137 1 1 DIDP C28O4H47 447.3469 95.8813 -1.2864

3. Estimating Chemical Compounds of Unknown Peaks Detected by Py-GC/MS Analysis of Polymers

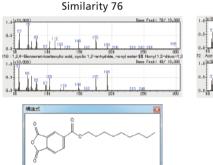


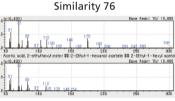
Results of Analyzing a Cable Sheath Using Py-GC/MS (El and CI Methods)

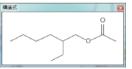
• Similarity Search of NIST 14

レポート(9) 表示(Y) 化合物循級(C) 処理(2) ヘルプ(日)								
·는 1881	后 計	化合物名	分子量	分子式	ライブラリ名	-		
	76 21	1,2,4-Benzenetricarboxylic acid, cyclic 1,2-	316	016H2205	NIST146.Ib			
	76	Carbonic acid, propergyl 2-ethylhexyl ester	212	C12H2003	NEST14.JB			
	76	1-Hexanol, 2-ethyl-, acetate \$\$ Acetic aci	172	C10H20O2	WR1 OR IB			
	75	Carbonic acid, 2~chloroethyl 2~ethylhexyl e	239	011 H210108	NIST14JIb			
	75	Carbonic acid, ally! 2-ethylhexyl ester	214	C12H22O3	NIST14JIb			
	75	Formic acid, 2-ethylhexyl ester	158	C0H18O2	NEST14JIL			
	74	Carbonic acid, 2nethylnexyl actyl ester	200	01710403	NIST14JIb			

Similarity of all results is below 80.







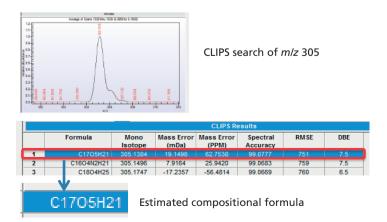
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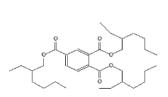
Chemical Formula: C<sub>17</sub>H<sub>20</sub>O<sub>5</sub> Exact Mass: 304.13 Estimated chemical

compound (Probably a degradation

product of TOTM)

Composition Analysis with MassWorks





Tri (2-ethylhexyl) trimellitate (TOTM)



## Conclusion

- 1. We have developed a new EI/CI ion source capable of both EI and CI that does not require physical exchange between EI and CI ion sources. The new EI/CI ion source demonstrated almost the same performance with both EI-specific and CI-specific ion source.
- 2. The composition of seven phthalate esters was estimated using the new EI/CI ion source and MassWorks, and the correct component compositions appeared.
- 3. A real-world sample was subject to Py-GC/MS using EI and CI methods. MassWorks was used to estimate the composition of peak where the corresponding compound could not be identified, and the unknown chemical compound was estimated to probably be a TOTM degradation product.

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