

Use of automated sample preparation techniques for challenging sample by GC-MS



Sean O'Connor

- 7 Years Flavour Industry
 - GC-FID, GCMS
- 11 ½ Years Unilever SEAC
 LC, GC, GCMS, LCMS, GCQQQ, LCQQQ MPS
- 1 Year Anatune
 - GC, GCMS, GCQQQ, GC/QTOF, MPS





Anatune

- Specialise in GC, MS and automation in many industries
- Agilent VAR
- UK Supplier of Gerstel MPS
 Autosampler
- Based in Girton, Cambridge





Applications Laboratory

- GC/MS
- GC/MS/MS
- GC/QTOF
- All have Dual head MPS2 Autosamplers



• Growing Team - 2 to 7 people in 18 months



Todays Talk

- Why Automate ?
- Metabolomics Derivatisation
 and Extraction
- Multivolatile method (MVM)





WHY AUTOMATE ?





Why Automate ?

- We're too busy
- Automation means losing jobs
- I have done it this way for years and it works



Manual method



- Prepare IS solution (5 minutes)
- Prepare calibration stock solution (5 minutes)
- Prepare 5 standards + 2 AQC (30 minutes)
- Add 100 mL of sample to each extraction flask (1 minute per sample)
- Add 200 µL of IS solution to each sample (10 seconds per sample)
- Add 20 mL of extraction solvent (1 minute per sample)
- Shake for 1 hour and allow to separate 30 minutes
- Remove extract from extraction vessel and transfer to vial for analysis (30 seconds per sample)
- Injection and GC run (30 minutes)
- Dispose of waste and clean glassware for next analysis (30 minutes)





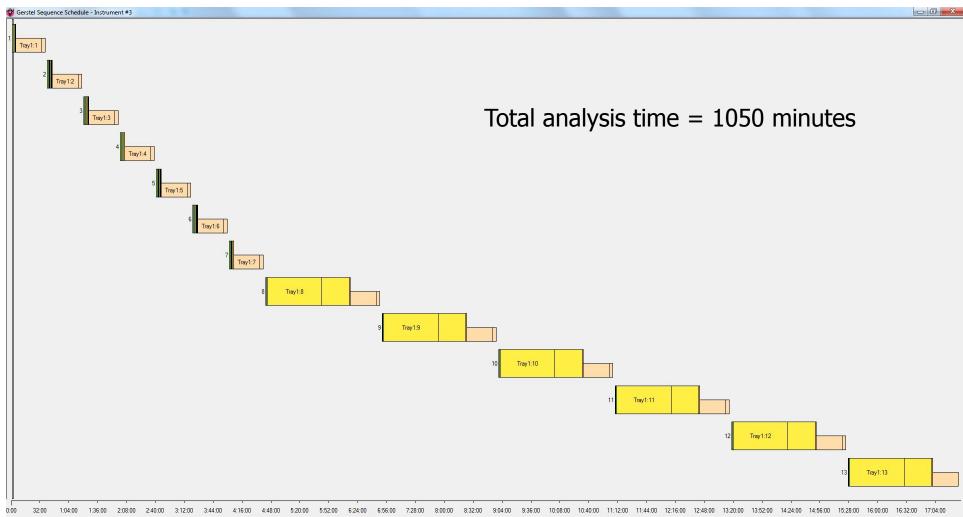
Why Automate ?

- Prepare IS solution (5 minutes)
- Prepare calibration stock solution (5 minutes)
- Add 5 mL of sample to each 10 mL vial (1 minute per sample)
- Prepare 5 standards + 2 AQC (30 minutes)
- Add 10 µL of IS solution to each sample (10 seconds)
- Add 1 mL of extraction solvent (20 seconds)
- Shake for 1 hour and allow to separate 30 minutes
- Directly inject from extract layer (30 minute run time)
- Dispose of vials (30 seconds)





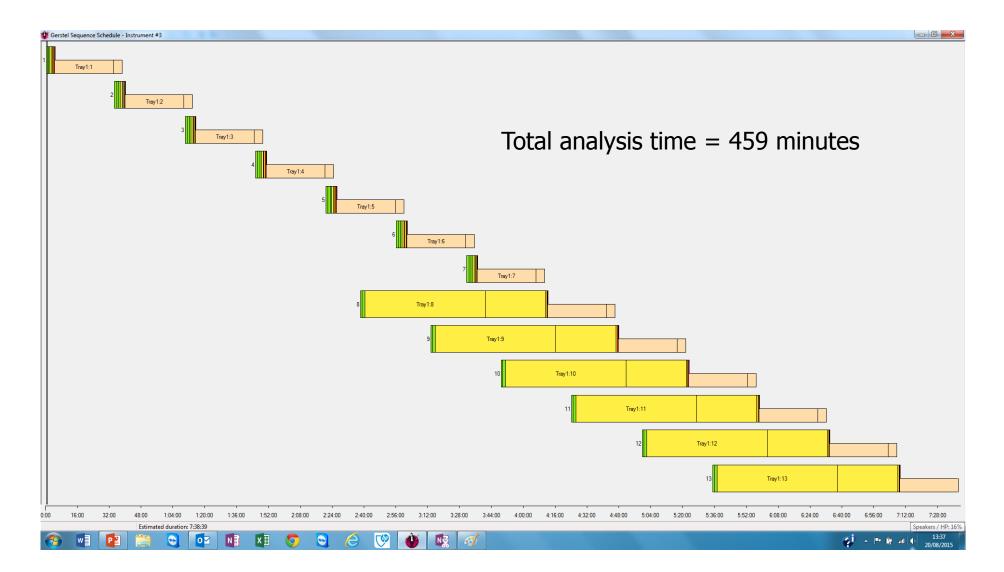
No Prep Ahead



Estimated duration: 17:30:03

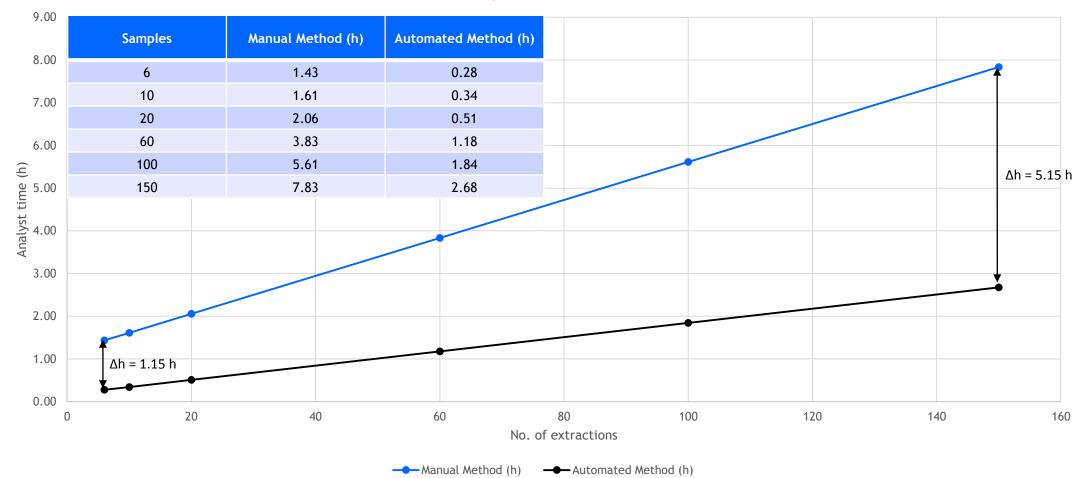


With Prep Ahead





Comparison of Analyst's Time

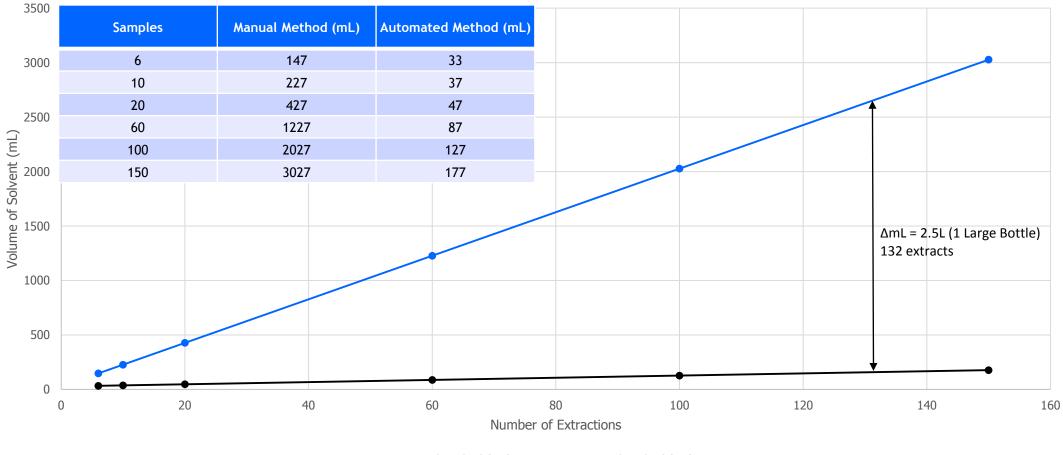


Analyst Time (hours)



Solvent Saving

Volume of Solvent (mL)



---- Manual Method (mL) ---- Automated Method (mL)



Why automate ?

Manual Preparation

- Preparation restricted to working
 hours
- Different people have different ideas as to how things are done
- Samples and standards are prepared all at the same time
- Glassware clean up required before next use
- Exposure to solvents a potential hazard / safety risk

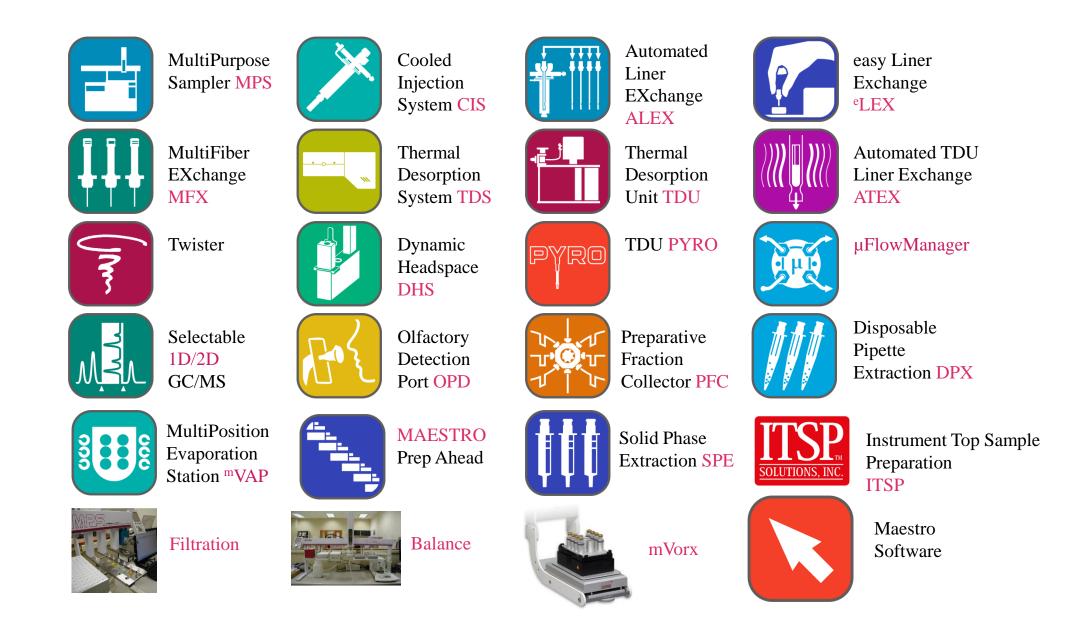
Automated Preparation

- Works 24/7
- Consistency
- Samples are prepared just in time for analysis
- Solvent (Cost) saving Every 132 extracts (on method shown) saving a 2.5 L bottle of solvent (£50-£100)
- Analysis done all in vial fewer losses
- Exposure to solvents reduced



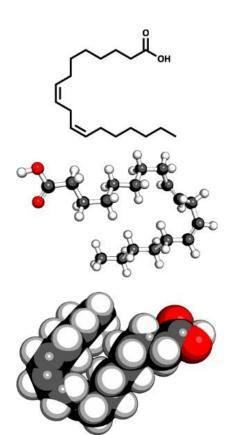


What tools do we have ?









OMIC DERIVATISATION AND EXTRACTION



'Omics Requirements

- Reliable
- Reproducible
- Lots of data points



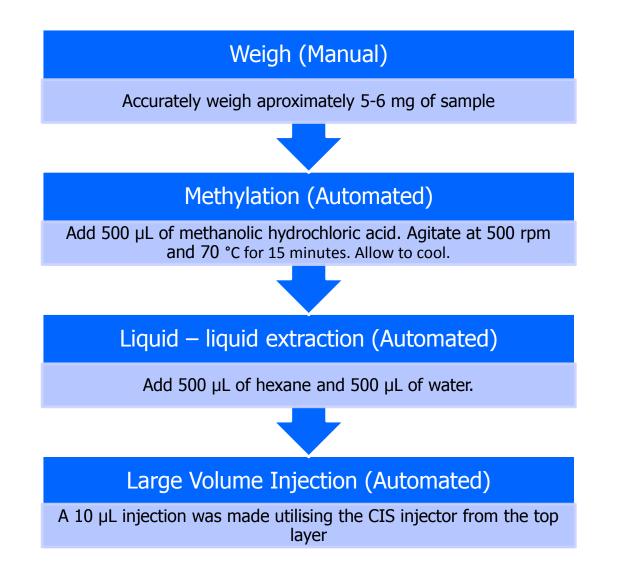






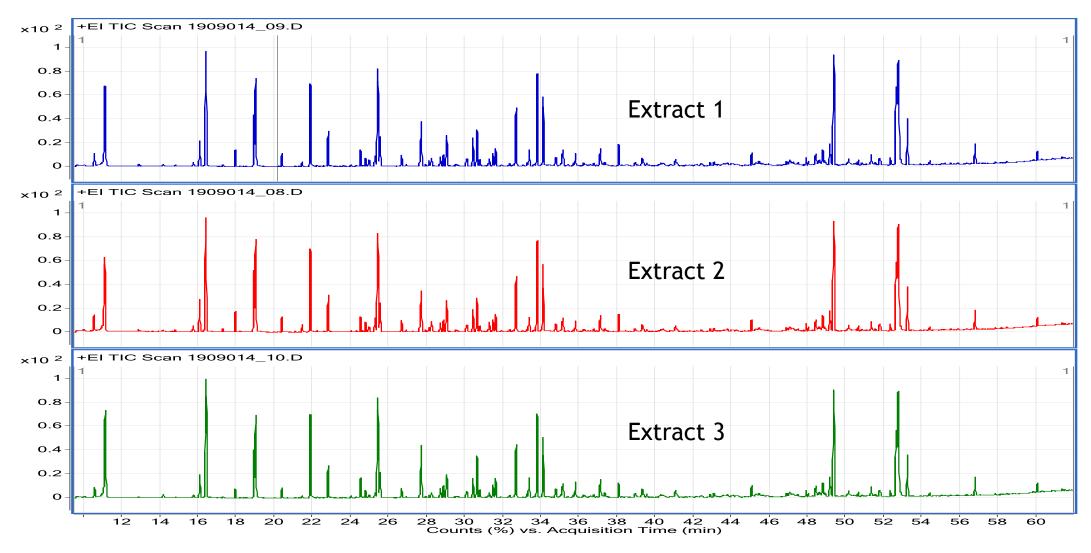






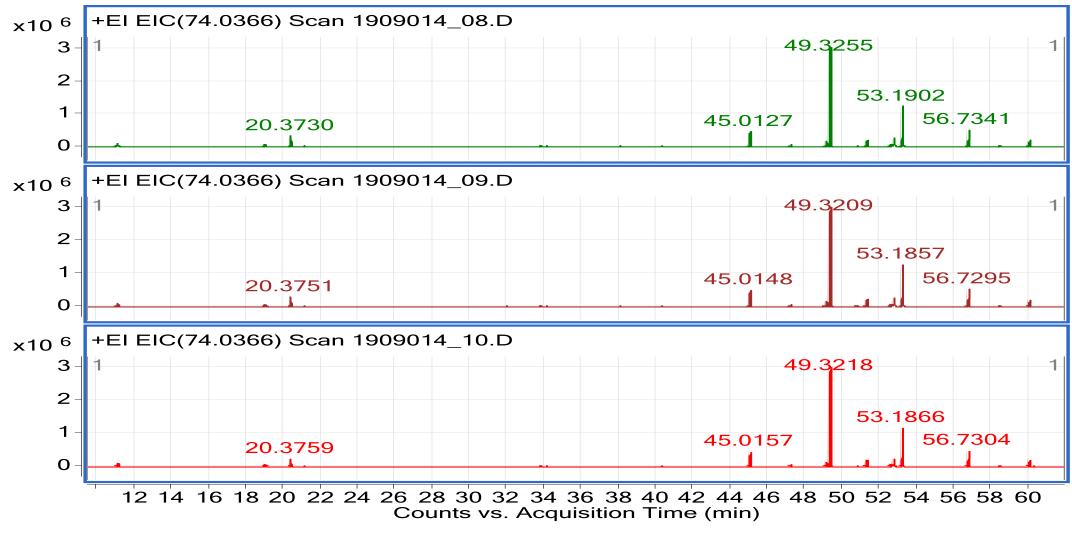
Hexane layer containing FAMES 0.5 m 5 11 Blank Sample





Total ion chromatogram of the hexane layer from the three extracts.



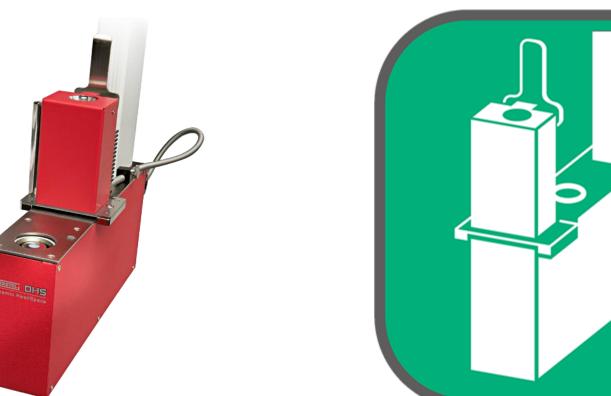


Extracted Ion Chromatogram of m/z = 74.0377 (McClafferty rearrangement of the ester grouping)



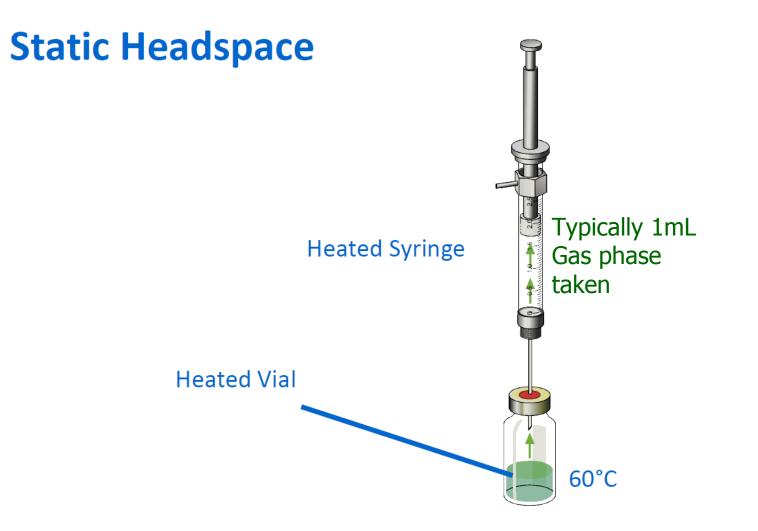
| FAME | % CV |
|-------------------------------|------|
| Methyl tetradecanoate (C14:0) | 6.9 |
| Methyl hexadecanoate (C16:0) | 4.7 |
| Methyl octadecanoate (C18:0) | 6.2 |
| Methyl eicosanoate (C20:0) | 5.8 |
| Methyl docosanoate (C22:0) | 3.9 |

DYNAMIC HEADSPACE AND MULTI-VOLATILE METHOD





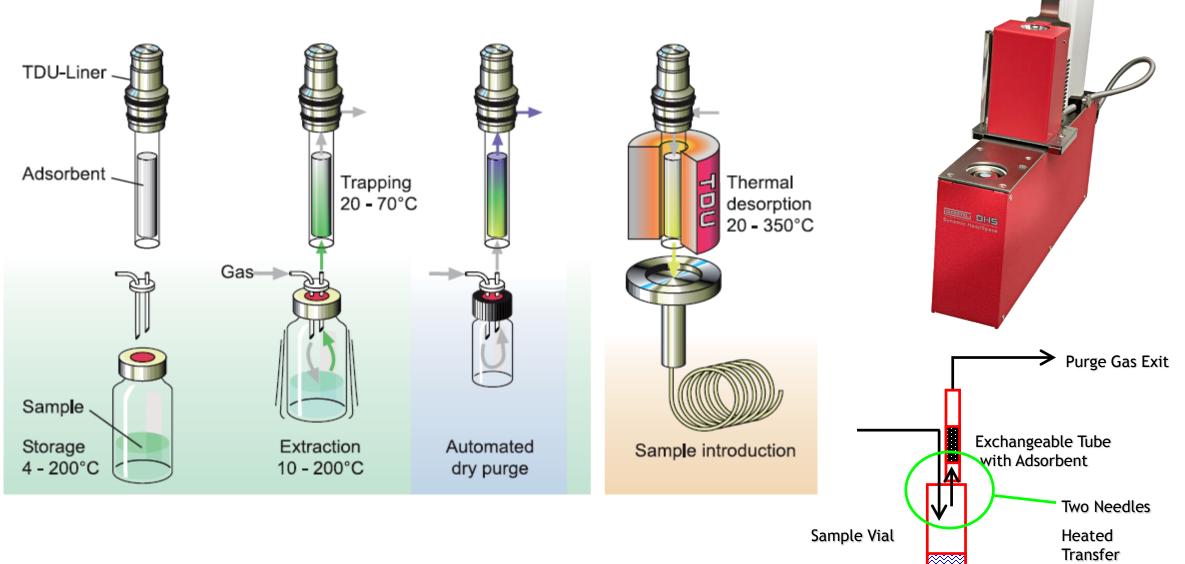




Dynamic Headspace (DHS)

Zone

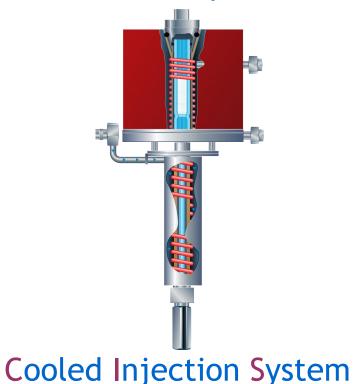


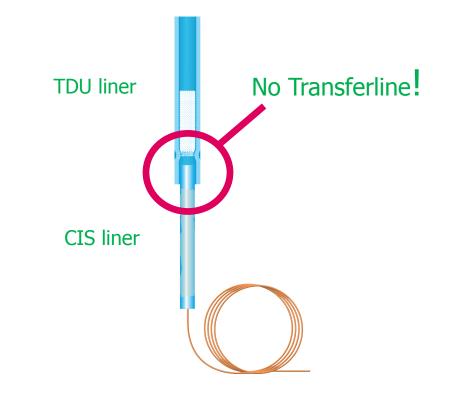




Thermal Desorption Unit (TDU) and Cooled Inlet System (CIS)

Thermal Desorption Unit



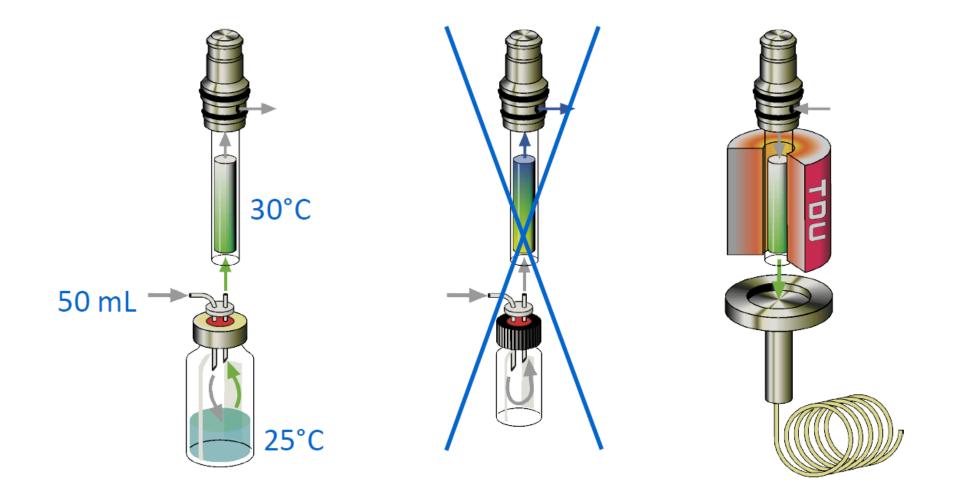




DHS – Very Volatile Analytes

Dynamic Headspace

Method 1: Very Volatile Analytes

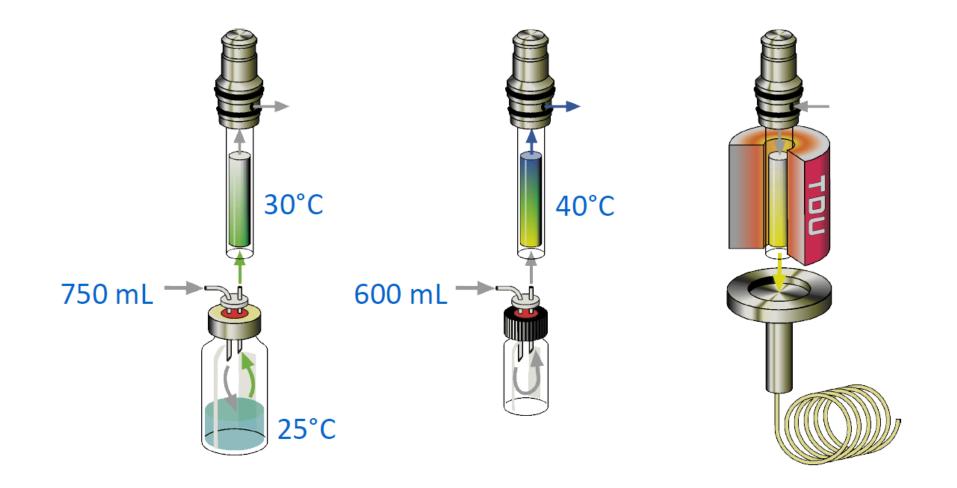




Dynamic Headspace – Volatile or Semi Volatile Analytes

Dynamic Headspace

Method 2: Volatile or Semi Volatile Analytes



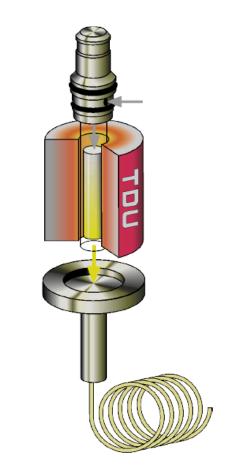


Fully Evaporative Technique - FET

Dynamic Headspace

Method 3: Volatile, non volatile and hydrophillic analytes







Multivolatile method

Journal of Chromatography A, 1371 (2014) 65–73



Multi-volatile method for aroma analysis using sequential dynamic headspace sampling with an application to brewed coffee $\!\!\!\!\!^{\bigstar}$

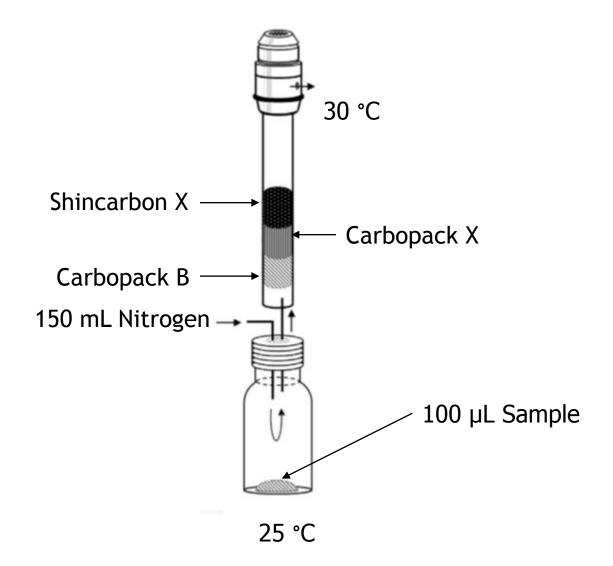


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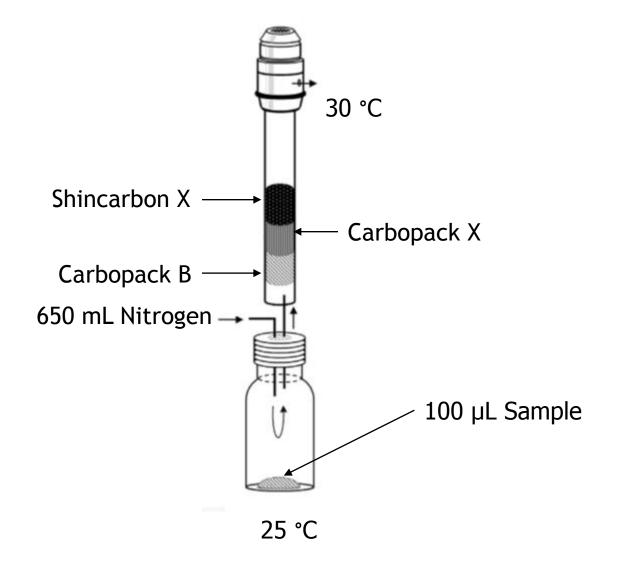


Trap 1 – Very Volatile Components



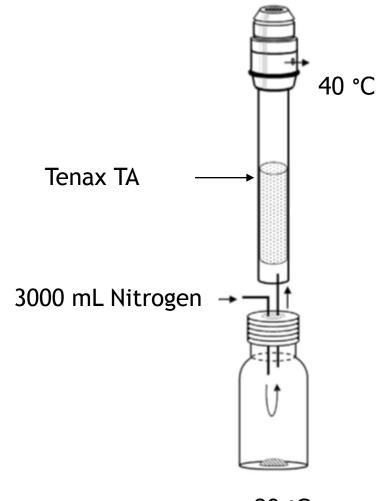


Trap 2 – Volatile/Semi Volatile Compounds





Trap 3 - FET





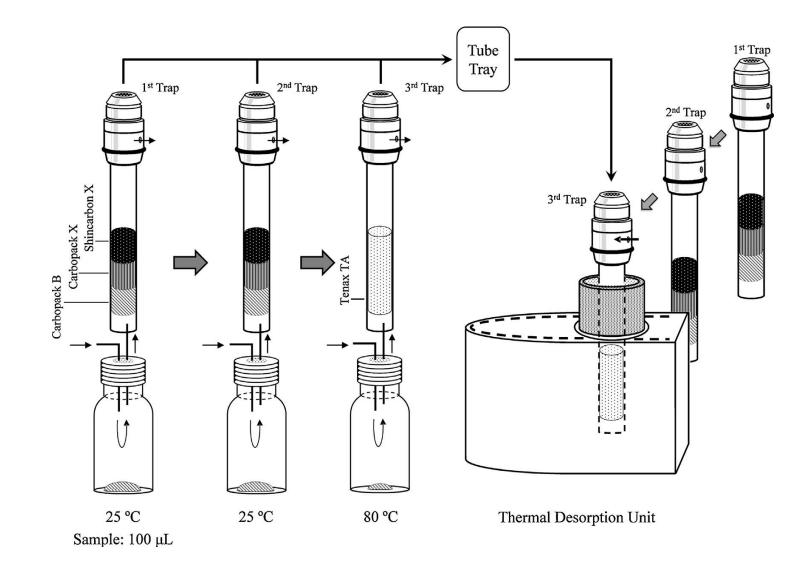
Multivolatile Method

| | 120 | г | | | | | | | | | | | | | | | Ē | ∃Trap | 1 🛛 | Trap 2 | 2 🖾 | Trap 3 |
|----------------|----------------------------------|-----------------|----------------------|---------------------|--------------|------------------|----------------|------------------|-----------------|------------------|----------|---------|----------------------|---------------------|---------------|---------------|-----------|-----------------|-----------|-------------------|---------------------|----------|
| [Recovery (%)] | 100 80 60 40 20 0 | 98 0.1 71 | 98 29 0. 57 | 95 111 0.1 49 | 101 1 0.0 | 100 0.3 67 | 95 82 35 | 103 .41 59 | 99 115 81 | | 95 86 | 53 6 | 27 1 8.5 | 5 90 70 41 | 95 64 | 96 77 4 | 25 5.2 | 92 67 58 | 21 5.3 | 10 | 8 98 76 47 | 2.8 |
| | Ū | Acetaldehyde | Furan | DMS | Propanal | 2-Methylfuran | Butanal | 2,3-Butanedione | Pentanal | 2,3-Pentanedione | DMDS | Pyrrole | 2,5-Dimethylpyrazine | Furfural | cis-3-Hexenol | 1-Hexanol | Guaiacol | Ethyl decanoate | Indole | gamma-Nonalactone | beta-Damascenone | Coumarin |
| | Vapor Pre [kPa, 25 | 120 | 79 | 64 | 42 | 21 | 14 | 6 | 4.2 | 3.9 | 3.1 | 1.3 | 0.39 | 0.11 | 0.11 | 0.28 | 0.015 | 0.0055 | 0.0016 | 0.0016 | 0.0015 | 0.000088 |

Fig. 3. Comparison of recoveries between three DHS sampling conditions for the test aroma compounds in 100 µL of water spiked at 100 ng mL⁻¹.



Multivolatile Method





Multivolatile Method

| 120 | ſ | | | | | | | | | | | | | | | ⊟Tr | ap 1 | ⊠Tr | ap 2 | ⊠Tr | ap 3 |
|---------------------------------|--------------|----------|-------|------------|---------------|-----------|-----------------|----------|------------------|-----|------------|----------------------|------------|---------------|-----------|----------|-----------------|-----------|-------------------|------------------|------------|
| 100 80 [%] 60 40 20 | | | | | | | | | | | | | | | | | | | | | |
| 0 | Acetaldehyde | Furan II | DMS I | Propanal I | 2-Methylfuran | Butanal I | 2,3-Butanedione | Pentanal | 2,3-Pentanedione | | Pyrrole II | 2,5-Dimethylpyrazine | Furfural 🔳 | cis-3-Hexenol | 1-Hexanol | Guaiacol | Ethyl decanoate | Indole II | gamma-Nonalactone | beta-Damascenone | Coumarin 👿 |
| Vapor Pressure [kPa, 25 °C] | 120 | 79 | 64 | 42 | 21 | 14 | 6 | 4.2 | 3.9 | 3.1 | 1.3 | 0.39 | 0.28 | 0.11 | 0.11 | 0.015 | 0.0055 | 0.0016 | 0.0016 | 0.0015 | 0.000088 |

Fig. 4. Total recoveries (sum of the recoveries obtained from the three DHS conditions and three individual thermal desorption-GC-MS analysis) for the test aroma compounds in 100 µL of water spiked at 100 ng mL⁻¹.



Conclusions – Part 1

 Automation can save technician time, solvent and therefore money

• Automation can increase precision and accuracy

 Automation can reduce contact with hazardous chemicals and therefore is safer



Conclusions Part 2

• Automation can produce reliable results

• Automation can provide reproducible results

 Automation working with Prep Ahead and 24/7 schedule can produce many data points



Conclusions Part 3

- DHS is can be used to detect low trace level concentrations
- MVM is excellent technique for the extraction of compounds in aqueous matrices from the very volatile (acetaldehyde) to the semi volatile (Vanillin, Coumarin)



Acknowledgements





Thank You Questions ?