

Inductively Coupled Plasma Mass Spectrometer ICPMS-2040

Application News

Analysis of Drinking Water by U.S. EPA Method 200.8 Using ICPMS-2040 with Collision Cell

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User Benefits

- ICPMS-2040 can achieve accurate analysis and high stability, as well as lower Ar gas consumption and running costs with mini-torch.
- Trace elements in drinking water can be analyzed without H_2 gas.
- Avoids a complex investigation of conditions by using the analytical conditions from preset methods.

Introduction

Method 200.8¹⁾, developed by the U.S. Environmental Protection Agency (EPA), is an ICP-MS method for the analysis of trace elements in groundwater, surface water, and drinking water. EPA 200.8 is a method based on analysis with no gas mode. On the other hand, ICP-MS is commonly equipped with collision/reaction cell technology to eliminate interferences such as polyatomic ions. In this application news, drinking water and a certified reference material were analyzed by ICPMS-2040. Most elements were measured in collision mode to eliminate interferences. Spike recovery and long-term stability were evaluated with reference to EPA 200.8 quality control (QC) requirements.

The analytical conditions used in this application can be easily registered from preset methods, enabling measurements to be performed without the need to develop them.



Fig. 1 ICPMS-2040 equipment

Sample Preparation

 Certified reference material (CRM) JSAC 0302-4a (The Japan Society for Analytical Chemistry)

JSAC 0302-4a (CRM for river water) was used to evaluate the accuracy of ICPMS-2040 measurement.

Drinking water

Drinking water was prepared to contain $1\nu/\nu\%~\text{HNO}_3$ and 100 $\mu\text{g/L}~\text{Au}.$

Standard Samples

• Calibration Standards

Calibration standards were prepared by mixing standard solutions (Be, Na, Mg, Al, K, Ca, V, Cr, Mn, Co, Ni, Cu, Zn, As, Se, Mo, Ag, Cd, Sb, Ba, Hg, Tl, Pb, Th, U) and adding HNO_3 and Au standard solution. Au standard solution is added to retain Hg in solution according to EPA 200.8. The concentrations in each calibration curve sample are shown in Table 1.

Internal Standard Solution

The internal standard was prepared by mixing standard solutions (Li, Sc, Ga, Y, Rh, In, Tb, Ho, Lu, Bi) and adding HNO_3 and Au standard solution. The concentrations in internal standard are shown in Table 2.

• Continuing Calibration Verification (CCV) Samples

CCV was prepared at the same concentration as STD4.

Elements	Calibration Standards (µg/L)					
	STD1	STD2	STD3	STD4	STD5	
Be, Al, V, Cr, Mn, Co, Ni, Cu, Zn, As, Se, Mo, Cd, Sb, Ba, Tl, Pb, Th, U	0	1	10	50	100	
Ag	0	1	10	50		
Hg	0	0.05	0.5	2.5	5	
Na, Mg, K, Ca	0	250	2500	12500	25000	
Au	100					
HNO ₃	1v/v%					

Table 1 Calibration Standards

Table 2 Internal Standard

Elements	Internal Standards (µg/L)
Li, Sc, Ga	1000
Y, Rh, In, Tb, Ho, Lu, Bi	100
Au	100
HNO ₃	1v/v%

Equipment Configuration and Analytical Conditions

The configuration of the ICP-MS system is shown in Table 3. To reduce running costs, analysis was performed with a mini-torch that consumes less argon gas than a typical plasma torch. To reduce the labor involved in sample preparation, the internal standard sample was added online using an online internal standard kit.

The analytical conditions used are shown in Table 4. The analytical conditions used in this study can be easily registered from LabSolutions[™] ICPMS preset methods.

Table 3 ICP-	MS System Configuration			
System:	ICPMS-2040			
Nebulizer:	Nebulizer, DC04			
Chamber:	Cyclone Chamber			
Torch:	Mini-Torch			
Skimmer Cone:	Nickel			
Autosampler:	AS-20			
Internal Standard Elements	Online Internal Standard Kit			
	(sample: internal standard = about 9 : 1)			
Table 4	Table 4 Analytical Conditions			
RF Power: 1.20 kW				
Plasma Gas Flowrate:	9.0 L/min			
Auxiliary Gas Flowrate: 1.10 L/min				

Carrier Gas Flowrate:

Dilution Gas Flowrate:

Collision Gas:

0.45 L/min

0.40 L/min

He

Collision Cell

EPA 200.8 is a method based on analysis in the no gas mode, which corrects for interference using theoretical equations. However, there are also interferences that can not be corrected by theoretical formulas in ICP-MS analysis. For example, ⁴⁴Ca¹⁶O⁺ interferes with ⁶⁰Ni due to Ca in drinking water. This interference of polyatomic ions cannot be corrected by theoretical equations, which can lead to measurements that are larger than the true value. He collision is effective means of eliminating the interferences of polyatomic ions. In this Application News, most elements were measured in He collision for accurate analysis.

Detection Limits

Calibration curves were prepared using the calibration standards shown in Table 1. Instrument Detection Limits (IDLs) and Method Detection Limits (MDLs) are shown in Table 5. Following EPA 200.8, IDLs were calculated from 3 times the standard deviation (σ) of 10 measurements of the calibration blank (STD1). MDLs were determined from σ of 7 measurements of blank spiked with concentrations two to five times the estimated detection limit.

IDL = 3 $\times \sigma$ (STD1) \times slope of the calibration curve

MDL = t \times σ (spiked blank) \times slope of the calibration curve where :

t = Student's t value for a 99% confidence level and a standard deviation estimate with n-1 degrees of freedom [t = 3.14 for seven replicates]

 σ = standard deviation of the replicate analyses

Elements	Gas Mode	Internal Standard	IDL (μg/L)	MDL (µg/L)
⁹ Be	No Gas	⁴⁵ Sc	0.008	0.03
²³ Na	He	⁴⁵ Sc	20	10
²⁴ Mg	He	⁴⁵ Sc	0.8	2
²⁷ AI	No Gas	⁴⁵ Sc	0.01	0.02
³⁹ K	He	⁴⁵ Sc	7	10
⁴⁴ Ca	He	⁴⁵ Sc	9	30
⁵¹ V	He	⁷¹ Ga	0.02	0.02
⁵² Cr	He	⁷¹ Ga	0.009	0.02
⁵⁵ Mn	He	⁷¹ Ga	0.009	0.03
⁵⁹ Co	He	⁷¹ Ga	0.004	0.007
⁶⁰ Ni	He	⁷¹ Ga	0.03	0.03
⁶³ Cu	He	⁷¹ Ga	0.005	0.006
⁶⁶ Zn	He	⁷¹ Ga	0.02	0.06
⁷⁵ As	He	⁷¹ Ga	0.03	0.04
⁷⁸ Se	He	⁷¹ Ga	0.2	0.2
⁹⁸ Mo	He	¹⁰³ Rh	0.004	0.004
¹⁰⁷ Ag	He	¹⁰³ Rh	0.001	0.007
¹¹¹ Cd	He	¹¹⁵ ln	0.01	0.01
¹²¹ Sb	He	¹¹⁵ ln	0.006	0.009
¹³⁷ Ba	He	¹¹⁵ ln	0.02	0.04
²⁰² Hg	He	²⁰⁹ Bi	0.006	0.005
²⁰⁵ TI	He	²⁰⁹ Bi	0.001	0.003
²⁰⁸ Pb	He	²⁰⁹ Bi	0.002	0.003
²³² Th	He	²⁰⁹ Bi	0.0007	0.001
²³⁸ U	He	²⁰⁹ Bi	0.0003	0.0009

Table 5 IDLs and MDLs for each element

Analysis of CRM JSAC 0302-4a

CRM JSAC 0302-4a was quantitatively analyzed using the calibration curves. The results of CRM analysis are shown in Table 6. Recoveries of 98 to 107 % of the certified value were achieved. These results certify the EPA 200.8 QC requirement of 90 to 110 % CRM recoveries. The accuracy of ICPMS-2050 measurement is confirmed.

Table 6 Result of CRM JSAC 0302-4a Analysis (N = 3)

Elements	Certified Value (µg/L)	Mean Measured Value (µg/L)	Recovery (%)	
9Be	0.99	0.98	99	
²³ Na	3900	4030	103	
²⁴ Mg	3100	3100	100	
²⁷ AI	79	80.1	101	
³⁹ K	470	466	99	
⁴⁴ Ca	12700	12500	98	
⁵² Cr	9.96	9.90	99	
⁵⁵ Mn	5.7	5.65	99	
⁶⁰ Ni	16.8	18.0	107	
⁶³ Cu	10.2	10.9	107	
⁶⁶ Zn	10.3	10.8	105	
⁷⁵ As	5.2	5.32	102	
⁷⁸ Se	5.0	5.1	102	
⁹⁸ Mo	0.38	0.382	101	
¹¹¹ Cd	0.98	1.00	102	
¹³⁷ Ba	0.89	0.90	101	
²⁰⁸ Pb	9.7	10.2	105	

Recovery (%) = Mean Measured Value / Certified Value imes 100

Analysis of Drinking Water and Spike Recovery

Unspiked and spiked drinking water were quantitatively measured using calibration curves, and spike recoveries were calculated. The results are shown in Table 7. Spike recoveries of 94 to 107% were obtained for all the measured elements. The results meet EPA 200.8 QC requirement of 70 to 130% recoveries. The matrix of drinking water cause no effect on the ICPMS-2040 analysis.

Long-Term Stability

Drinking water was analyzed for about seven hours to evaluate the long-term stability of ICPMS-2040. CCV was measured every 10 samples to confirm the validity of the calibration curves.

Recoveries of CCV during the analysis are shown in Fig. 2. The CCV recoveries of all the measured elements during the analysis were within 90 to110% (red dotted line). If the CCV recoveries fall outside the 90 to 110%, recalibration is required.

The internal standard recoveries during analysis are shown in Fig. 3. All measured internal standard recoveries were within the EPA 200.8 QC requirement of 60 to 125% (red dotted line). If the internal standard recoveries also fall outside this range, recalibration is needed.

The results of CCV and internal standard recoveries showed good long-term stability of ICPMS-2040 analysis.

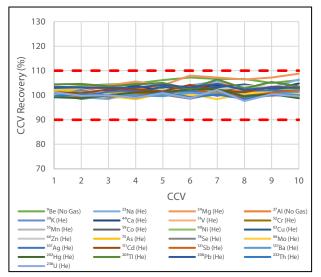


Fig. 2 CCV recoveries over 7 hours analysis

Unspiked Drinking Water		STD3 Spiked Drinking Water			STD4 Spiked Drinking Water		
Elements	Measured Value (µg/L)	Spike Conc. (µg/L)	Measured Value (µg/L)	Recovery (%)	Spike Conc. (µg/L)	Measured Value (µg/L)	Recovery (%)
⁹ Be	N.D.	10	10.3	103	50	50.0	100
²³ Na	6530	2500	8990	98	12500	19200	101
²⁴ Mg	9200	2500	11600	(96)	12500	22100	103
²⁷ AI	8.95	10	18.8	99	50	57.5	97
³⁹ K	887	2500	3300	97	12500	12800	95
⁴⁴ Ca	21200	2500	23700	(100)	12500	32900	94
⁵¹ V	15.8	10	25.9	101	50	64.8	98
⁵² Cr	0.591	10	10.8	102	50	50.6	100
⁵⁵ Mn	0.866	10	10.8	99	50	50.1	98
⁵⁹ Co	N.D.	10	10.6	106	50	51.8	104
⁶⁰ Ni	0.20	10	10.6	104	50	51.5	103
⁶³ Cu	0.256	10	10.9	106	50	53.1	106
⁶⁶ Zn	0.33	10	10.9	106	50	52.4	104
⁷⁵ As	0.18	10	10.4	102	50	51.2	102
⁷⁸ Se	N.D.	10	10.2	102	50	48.4	97
⁹⁸ Mo	0.135	10	9.7	96	50	50.1	100
¹⁰⁷ Ag	0.017	10	10.3	103	50	50.5	101
¹¹¹ Cd	N.D.	10	10.5	105	50	51.8	104
¹²¹ Sb	0.012	10	10.6	106	50	52.0	104
¹³⁷ Ba	0.88	10	11.3	104	50	52.4	103
²⁰² Hg	N.D.	0.5	0.536	107	2.5	2.65	106
²⁰⁵ TI	N.D.	10	10.5	105	50	50.6	101
²⁰⁸ Pb	0.009	10	10.6	106	50	52.1	104
²³² Th	0.0064	10	10.1	101	50	51.0	102
²³⁸ U	0.0458	10	10.4	104	50	52.0	104

Table 7 Spike Recoveries of Drinking Water

N.D. = Not Detected (< MDL)

Recovery (%) = (Spiked Sample – Unspiked Sample) / Spike conc. \times 100

(): According to EPA 200.8, spike recovery calculations are not required if spike concentration is less than 30% of unspiked sample concentration.

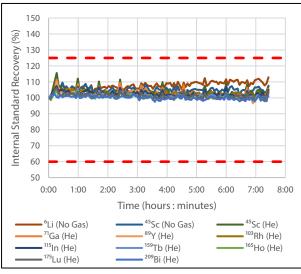


Fig. 3 Internal standard recoveries over 7 hours analysis

Conclusion

In this Application News, drinking water was analyzed using ICPMS-2040. Good recoveries of CRM and spiked samples were obtained, confirming the accuracy of ICPMS-2040 analysis. The CCV recoveries of all the measured elements were within 90 to 110%. Also, all measured internal standard recoveries were within the EPA 200.8 QC requirement of 60 to 125%, demonstrating the high stability of the ICPMS-2040.

ICPMS-2040 can achieve accurate analysis and high stability, as well as lower argon gas consumption and lower running costs, thanks to the use of a mini-torch. In addition, the analytical conditions can be easily registered from the preset methods, and analysis can be started without the need to develop the analytical method.

<References>

EPA Method 200.8 Determination of Trace Elements in Waters 1) and Wastes by Inductively Coupled Plasma - Mass Spectrometry Revision 5.4

<Related Applications>

Analysis of Drinking Water by U.S. EPA Method 200.8 Using ICPMS-2050 1. with Collision/Reaction Cell, Application News 01-00573_EN

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