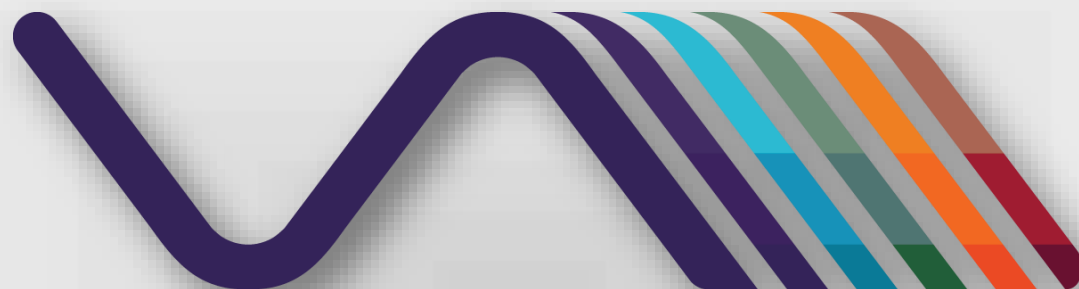


We don't have anything against mass spectrometry.
We just think it's time for a **worthy alternative**.



V U V A N A L Y T I C S

GC-VUV, využití absorpční UV spektrometrie ve vakuové oblasti spektra ve spojení s plynovou chromatografií

František Laštovička, AMEDIS, spol.s r.o.

VITATOX 2021, 4.10.2021

All you need is...
AMEDIS

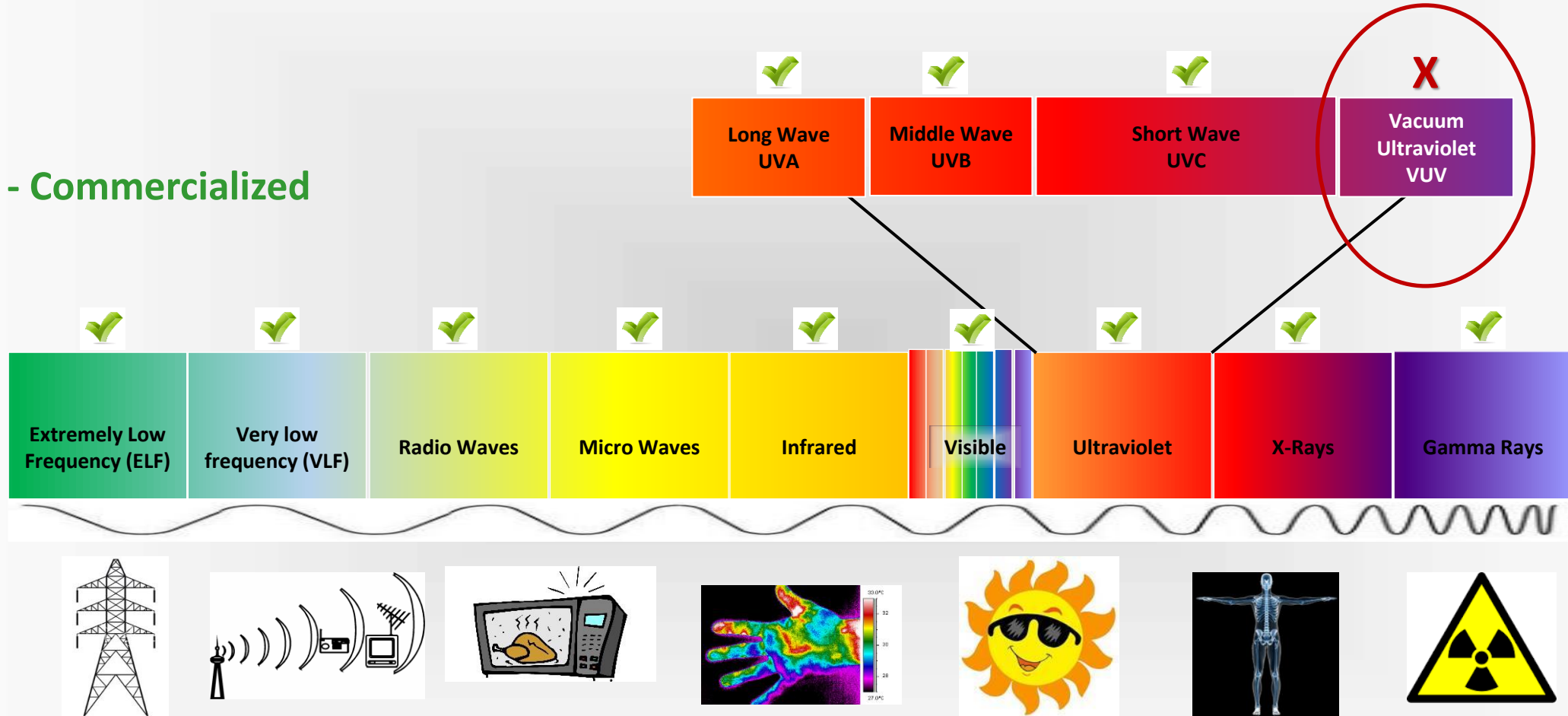


V U V A N A L Y T I C S

VUV Detection Overview

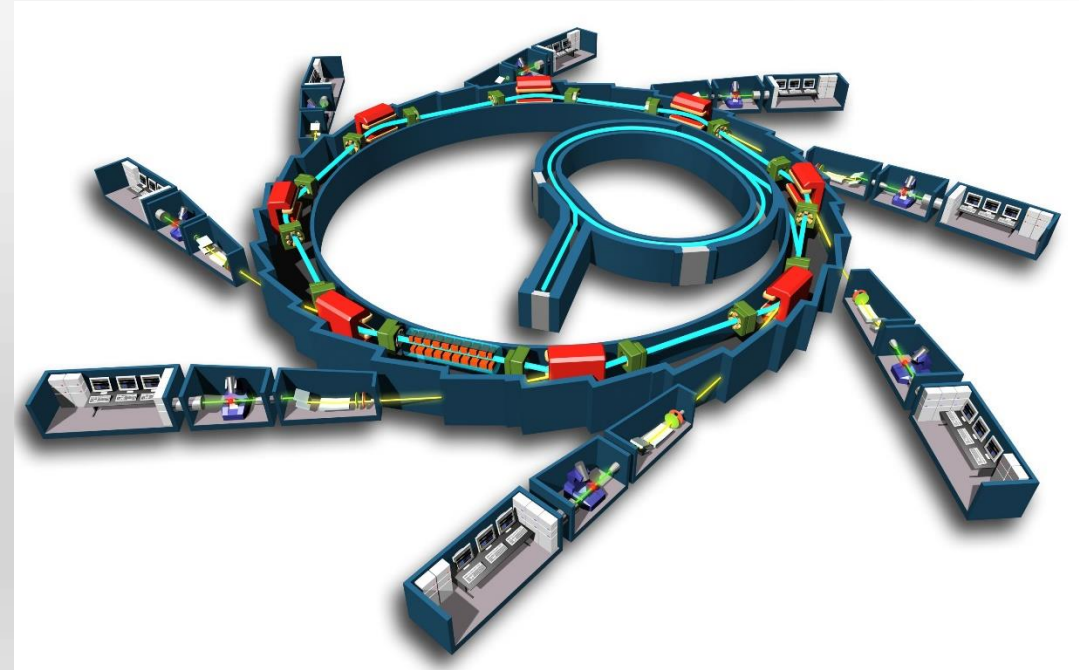
VUV = Vacuum Ultraviolet 10 – 200 nm (120 – 240 – 430 nm)

✓ - Commercialized



The Electromagnetic Spectrum

First VUV Measurements: Synchrotron



BROOKHAVEN
NATIONAL LABORATORY

SOLEIL
SYNCHROTRON

- **National Synchrotron Light Source (NSLS)** at Brookhaven National Laboratory in Upton, New York
- Used to conduct fundamental experiments in Chemistry and Physics
- UHV Chambers, incredibly bright sources
- The National Synchrotron Light Source II (NSLS II) cost \$912,000,000 just to build alone


VUV ANALYTICS

Recent Market Trend

- **Increased data content**

- More selective analytical power / detection
- Simplify background matrix
- More robust data
- Provide more accurate compound identification
- Improve quantitation and repeatability
- Lower detection levels

Introducing the VGA-100 / VGA-101
Another powerful tool in the toolkit!



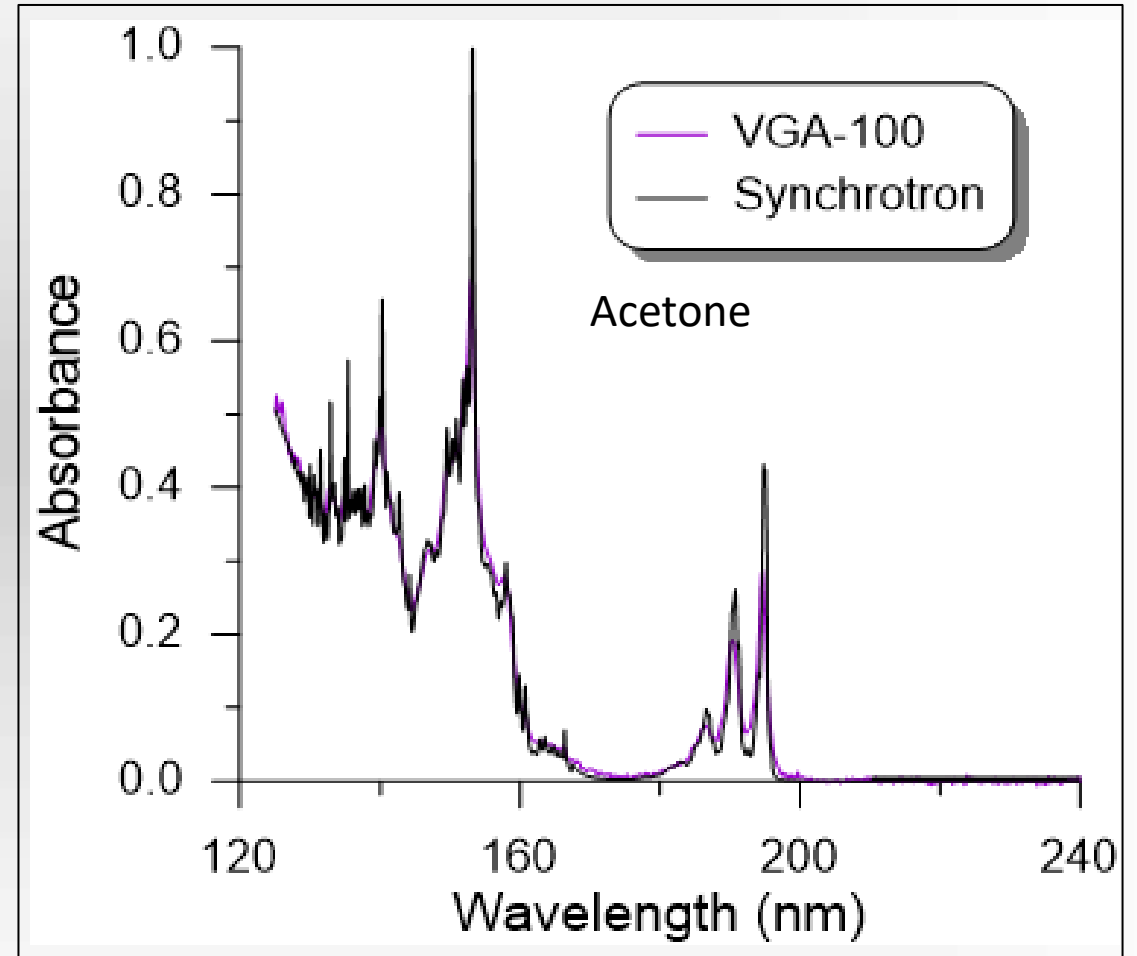
The world's first Vacuum Ultraviolet Absorption
gas chromatography detector

The World's First VUV Absorption Detectors: VGA-100 / VGA-101

- VUV Analytics has made vacuum ultraviolet (VUV) absorption available to the general scientific community
 - VUV absorption (VGA-100: 120-240nm ; VGA-101: 120 – 430 nm) enables powerful detection capabilities and a new unique orthogonal separation
- Benefits of the VUV detector include:
 - Universal detection with Powerful selectivity
 - Strong absorption of all gas phase molecules with exception of hydrogen, helium, nitrogen and argon
 - Unambiguous compound identification (exception: large aliphatic hydrocarbons)
 - Ability to deconvolve co-eluting analytes
 - Reduce matrix contributions with spectral filters
 - Clear and easy isomer differentiation
 - Excellent sensitivity
 - Low picograms,
 - Spectral filters increase analyte sensitivity in targeted wavelength regions
 - Non-destructive analysis
 - No ionization, analyze intact molecules
 - Excellent temporal resolution
 - Up to 100Hz Sampling
 - Dynamic Make-up gas control
 - Predictable linear response / No frequent calibration required
 - Beer's law: 1st principle detection reduces calibration burdens: robust and uniform response factors
 - Reliable & Easy to use
 - No routine maintenance necessary, no vacuum pumps, Ozone cleans
 - Long Life Lamp



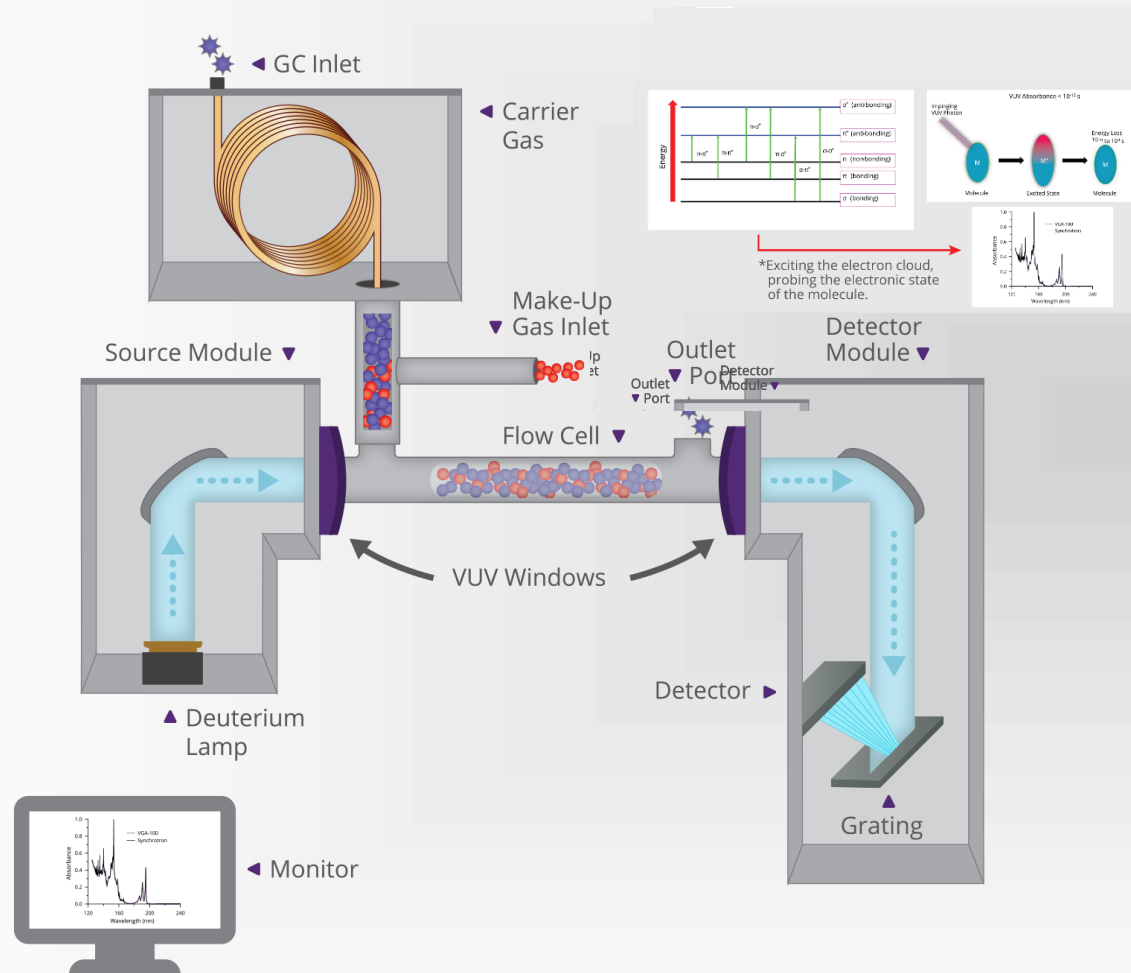
GC-Vacuum Ultraviolet Absorbance Spectroscopy (GC-VUV)



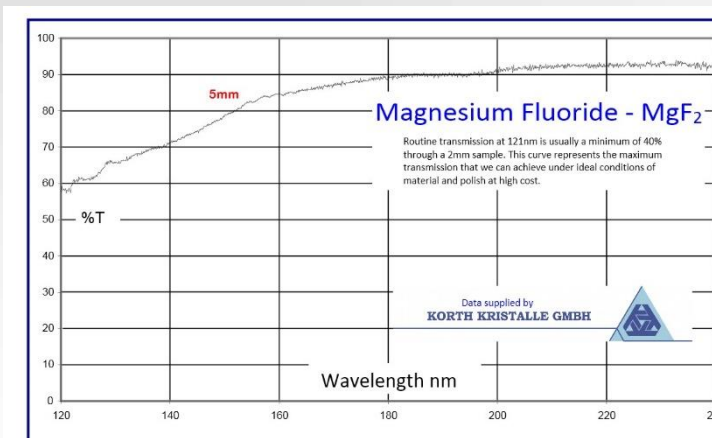
Brookhaven on your benchtop: VUV Spectra *without* a Beamline



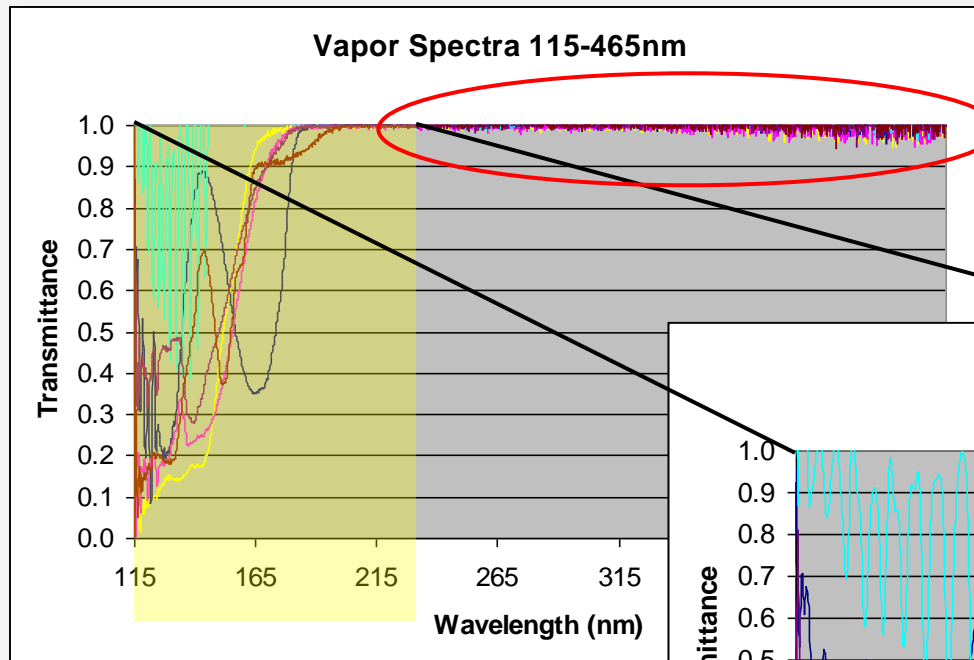
Vacuum Ultraviolet Absorbance Spectrophotometer



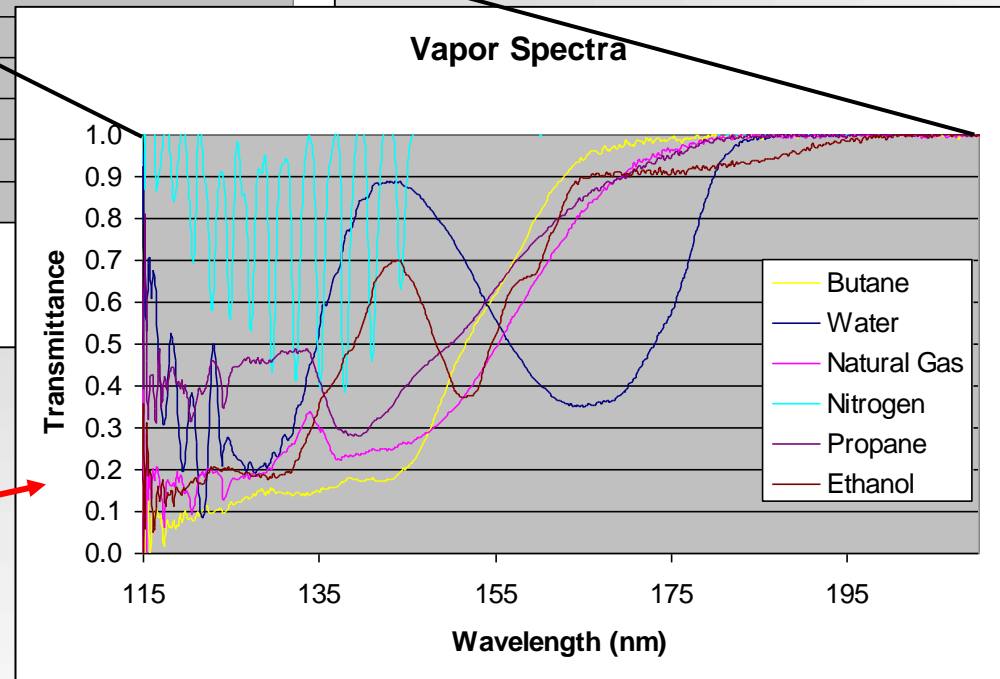
- Material Selection: MgF₂ windows to transmit VUV radiation, Ultraclean Environment
- D₂ Source
- Non-Vacuum environment, Inert makeup gas (N₂, He, Ar) used to ensure constant flow in the flow cell
- Patented Optical Design and Environment
- Optical Mirrors, grating, and backthinned CCD
- Transfer line and cell heated up to 300°C or 430°C (VGA-101)
- Flow cell dimensions 10cm x 0.1cm – 78 ul (VGA-100) or 10cm x 0.05cm – 39 ul (VGA-101)
- Easily interfaced with any GC via a heated transfer line
- Self cleaning Ozon source
- Self diagnostics
- Factory installed and aligned



VUV Gas Phase Spectra



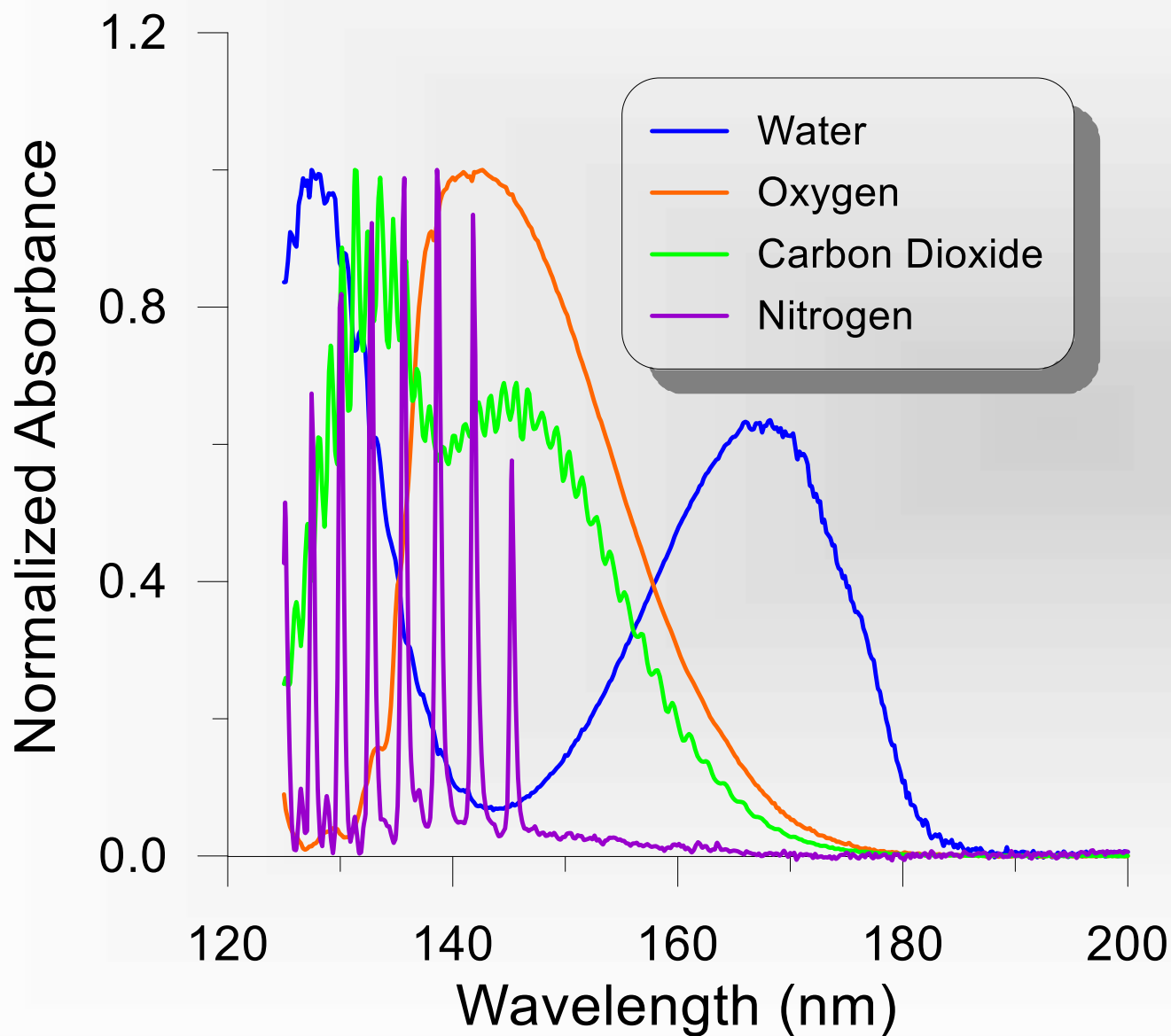
Most gas phase compounds have little or no absorption in the visible or UV region



Everything absorbs in the Vacuum Ultraviolet region

Often hundreds of times stronger than in IR absorption

VUV – All compounds absorb strongly

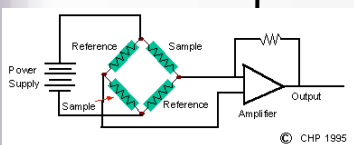
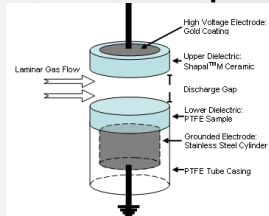
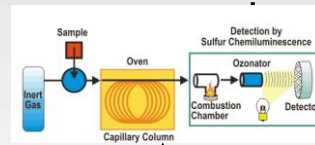
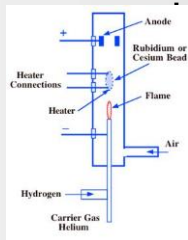
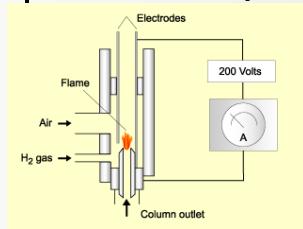
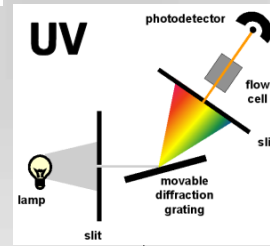
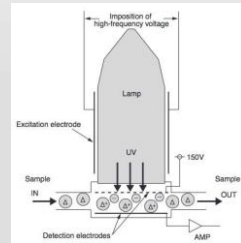
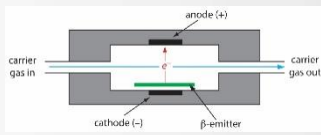
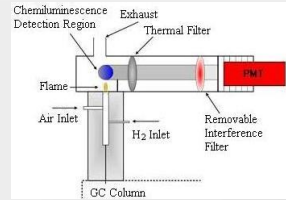
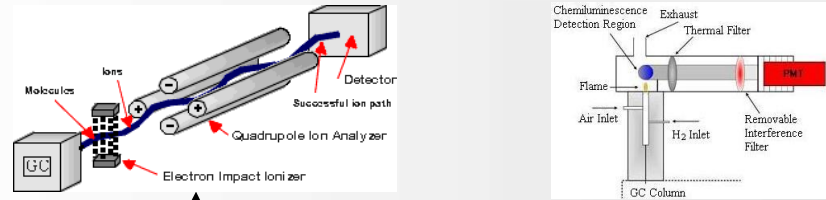


“The excitation **energies associated with** electrons forming **most single bonds are** sufficiently high that absorption occurs in the so-called **vacuum ultraviolet** region ($\lambda < 185\text{nm}$), where components in the atmosphere also absorb strongly. **Because of experimental difficulties associated with the vacuum ultraviolet region, most spectrophotometric investigations of organic compounds have involved longer wavelengths than 185nm.**”

- Principals of Instrumental Analysis,
by Douglas Skoog, Sixth Edition, 2006



The First New GC Detector Technology in ~30 Years!



TCD 1950 MS 1952 FID 1958 ECD 1959 BID 1960 NPD 1961 FPD 1962

PID 1976 SCD 1982 UV 1983

VUV 2014



Product Overview

We don't have anything against mass spectrometry.
We just think it's time for a **worthy alternative**.



VGA-100 Detector

2014 Gulf Coast Conference
VGA-100
"Best New Analytical Instrument"



If your current method
is leaving you in the dark...
Shine a new light on it.



VGA-101 Detector with enhanced
features

The Analytical Scientist
VGA-100
"Top 5 Innovation Award"



R&D-100 Award
VGA-100

"Analytical / Test Category Winner"



Introducing the
SVGA-100 for streaming gases:
Detection Perfection.



SVGA-100 Analyzer

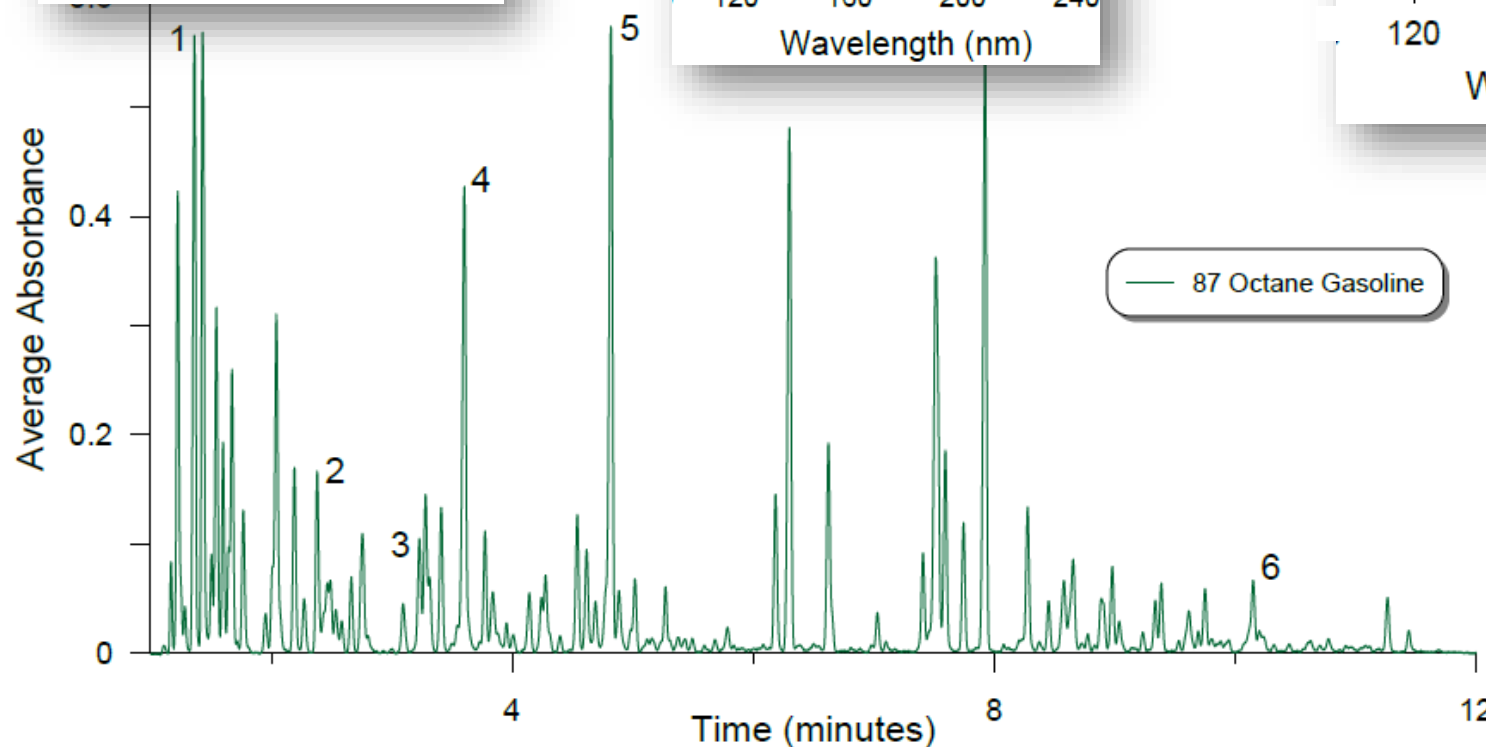
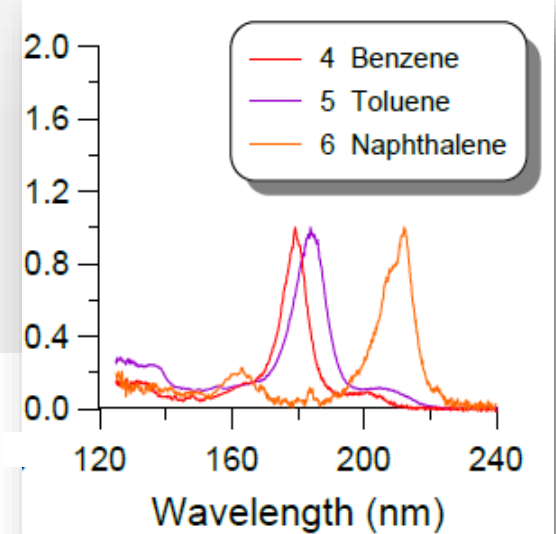
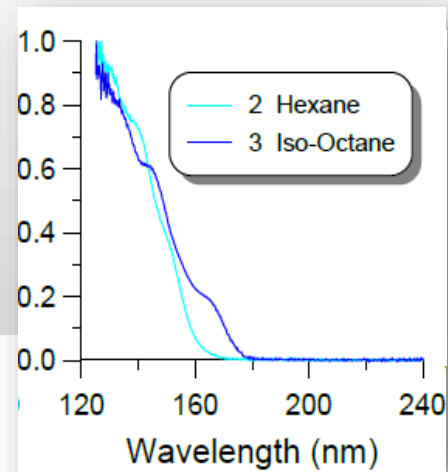
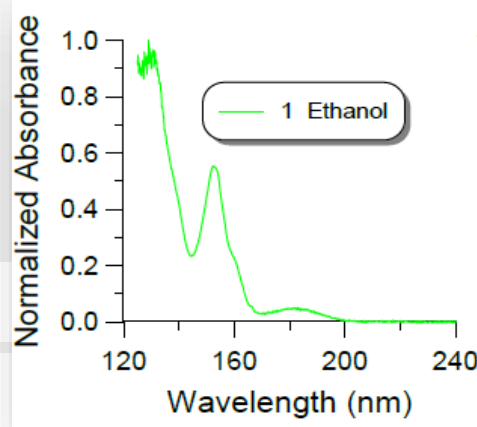
2016 ISA Analysis Symposium
SVGA-100
"Innovative Product of the Year"
Innovative Product of the Year Award



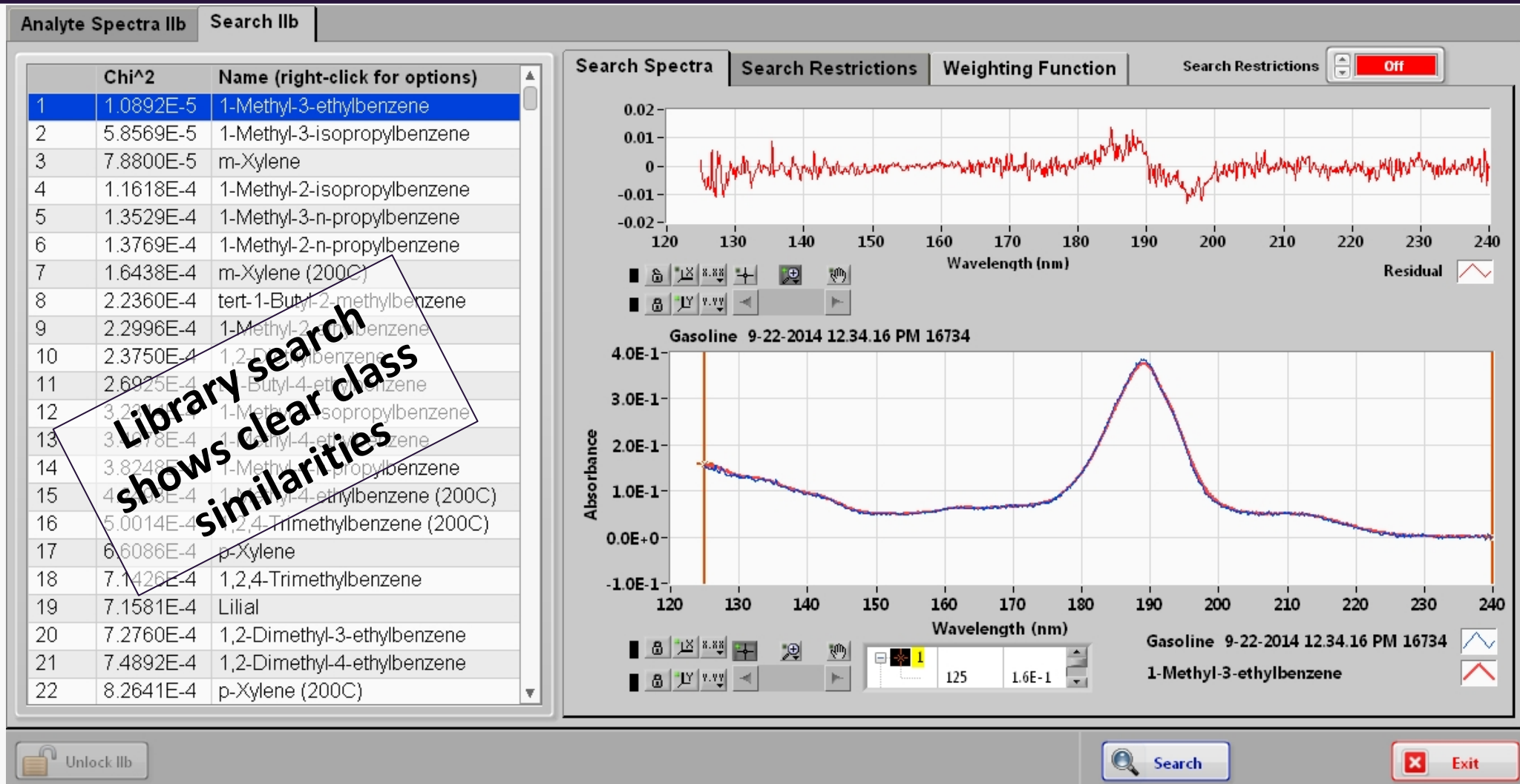
Compound Identification and Compound Deconvolution

Compound Specific Absorption Spectra

- All chromatogram peaks represent unique compound absorption
- Compound classes show like absorption responses



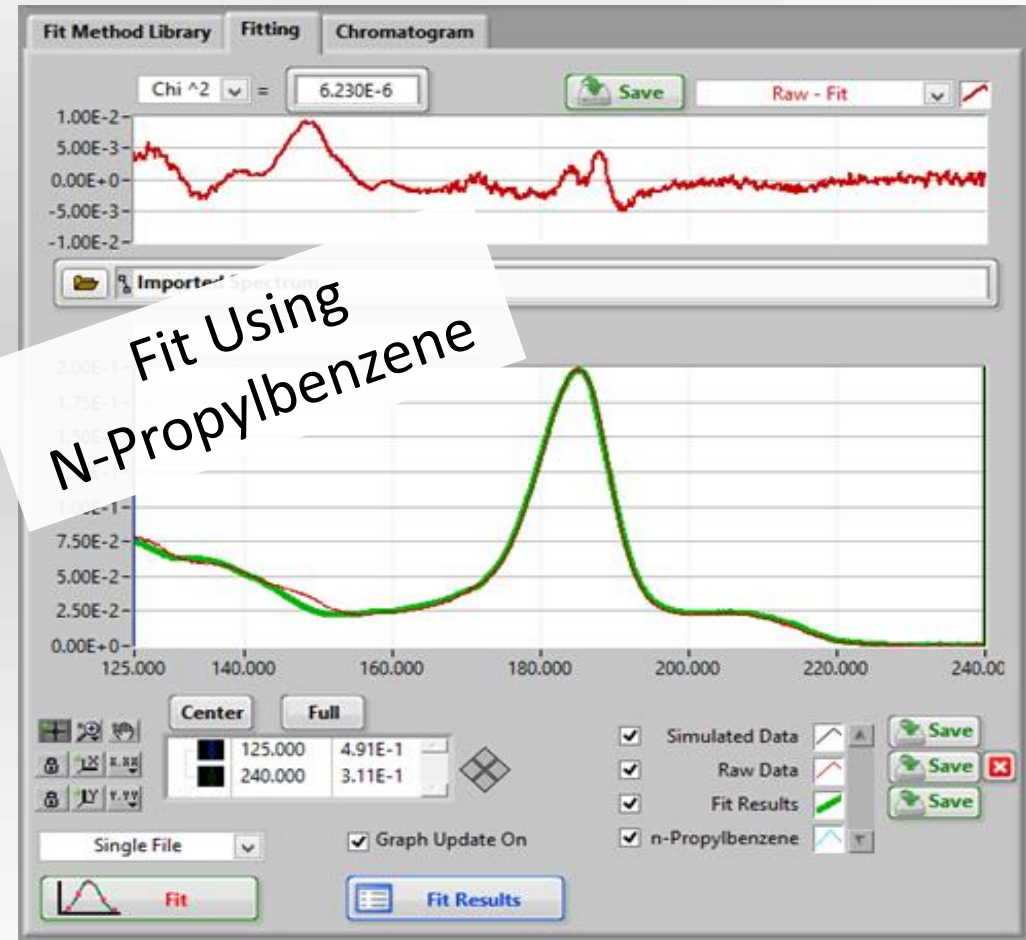
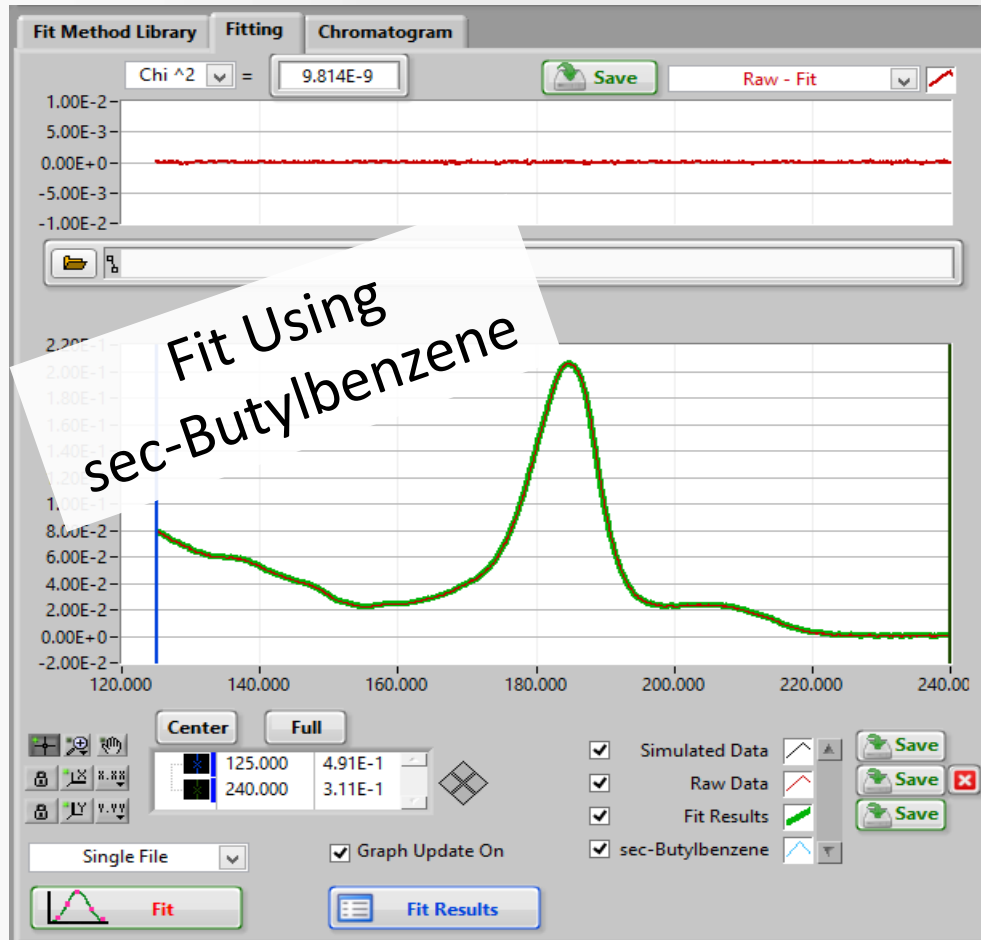
Library search and Unambiguous Compound Identification



Spectra are very robust;
No ghost components in the library match list

Similar But Very Distinct

Visual similarities are easily distinguished in the fitting routine;
minor differences are significant



The **chi-squared distribution** is the distribution of a sum of the squares of k [independent standard normal](#) random variables
The chi-squared distribution is used in the common [tests](#) for [goodness of fit](#) of two criteria of [qualitative data](#),

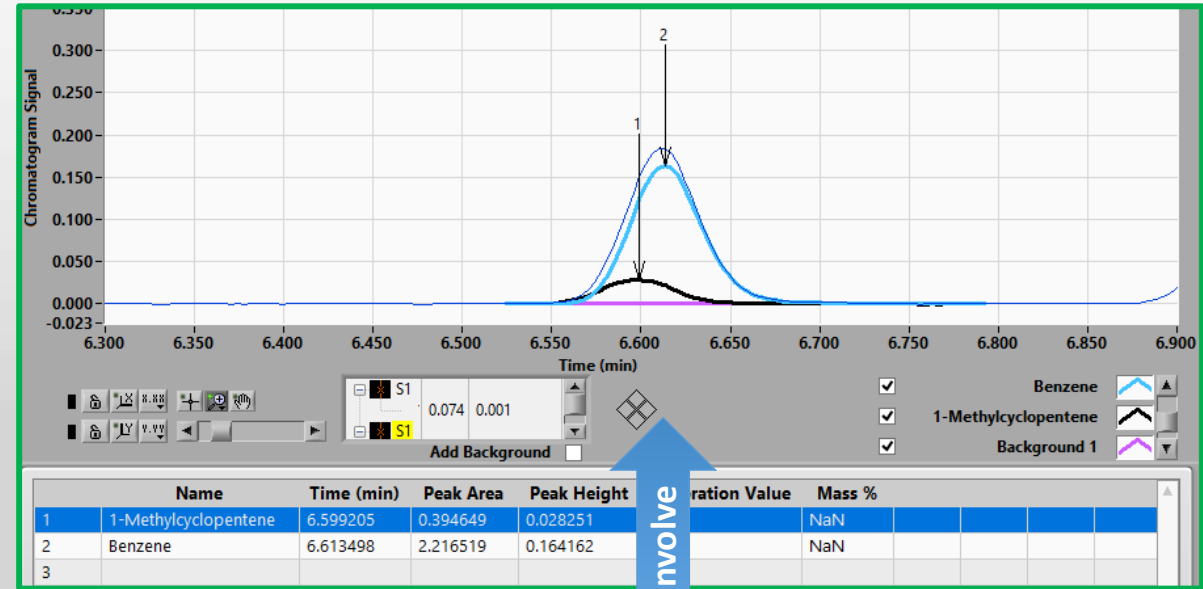
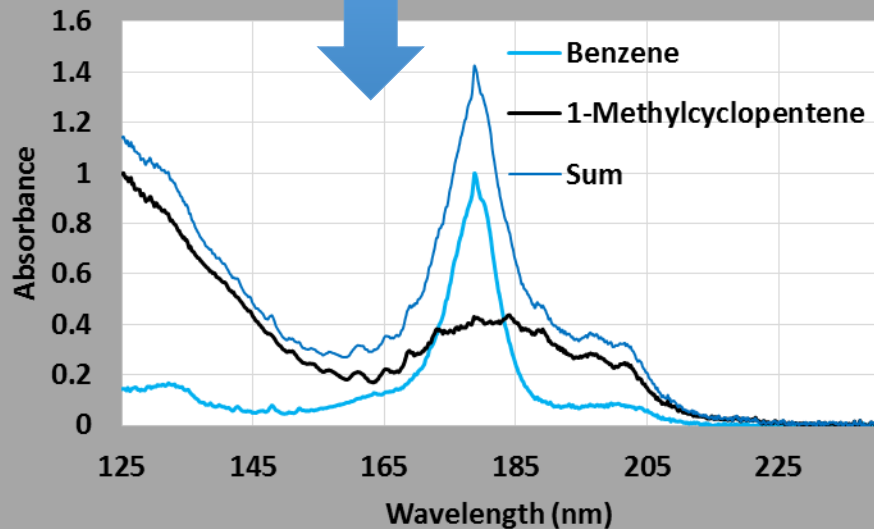
Deconvolution

- Total absorption is proportional to the product of the concentration and the absorption cross section
- Co-elution is a sum of these products
- Linear regression allows for easy deconvolution of the compound concentrations; even for co-eluting compounds

Beer's Law: $A = \epsilon bC$

$$A_{\text{total}} = A_1 + A_2 + \dots + A_n$$

Additive



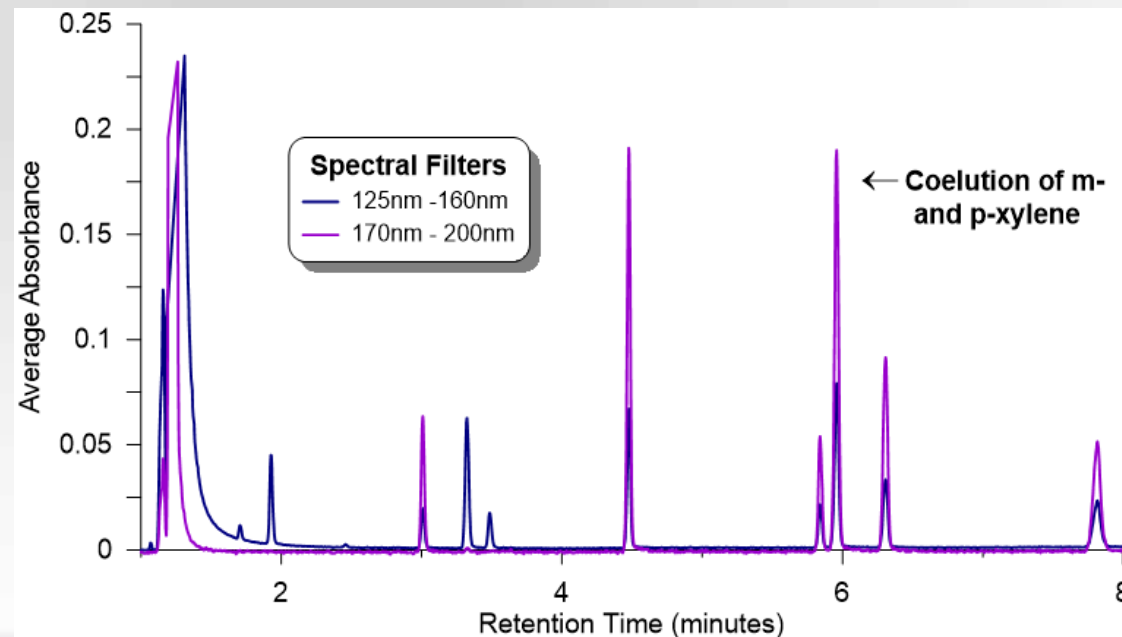
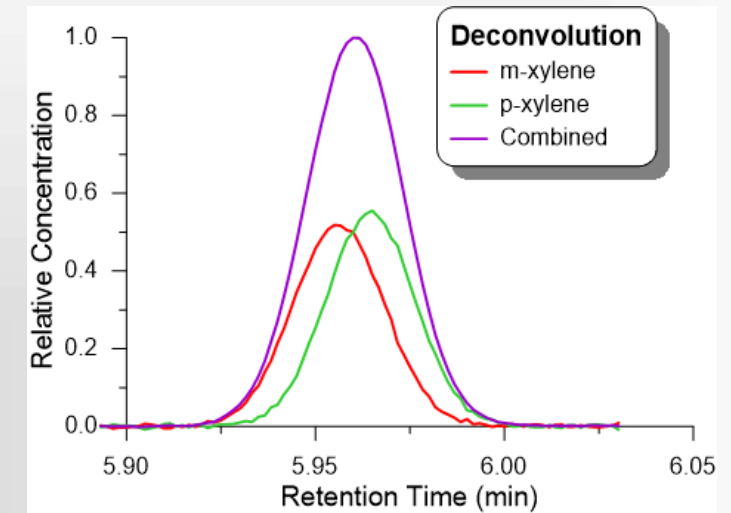
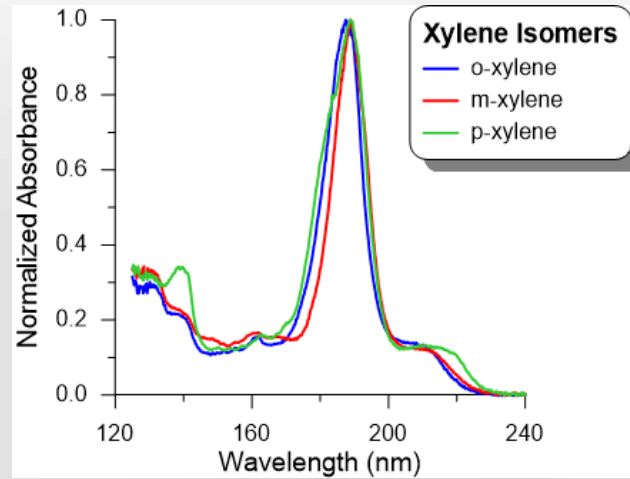
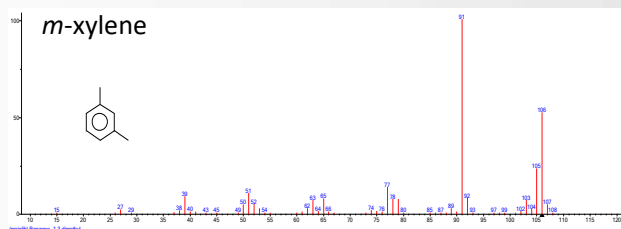
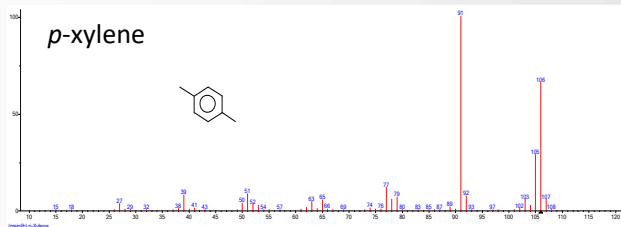
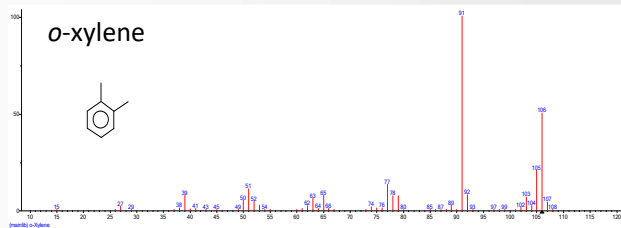
Deconvolve

$$A_{\text{co-elution}} = f_1 A_1 + f_2 A_2$$

A_i = the absorbance of component i at a given λ

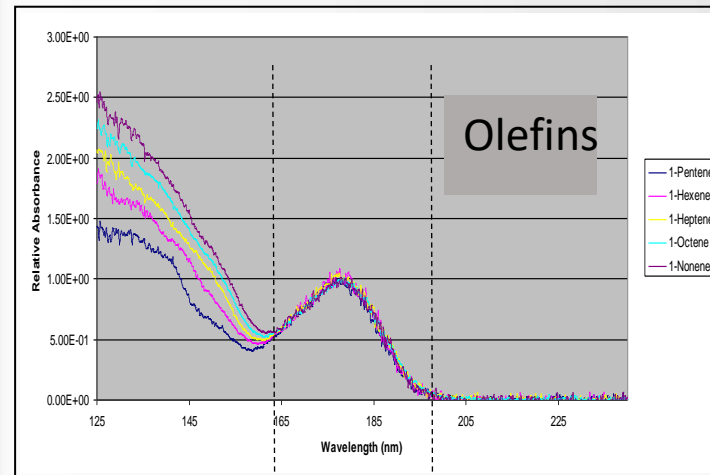
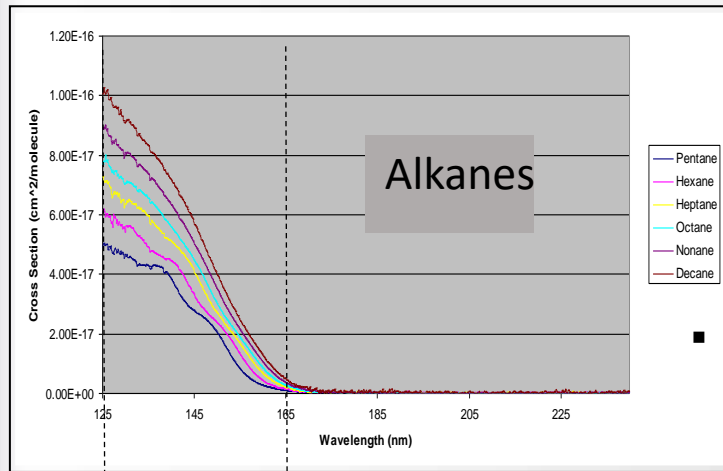
f_i = the fit coefficient of component i

Spectral Deconvolution of m&p Xylene



Class Analysis (Use of Spectral Filters)

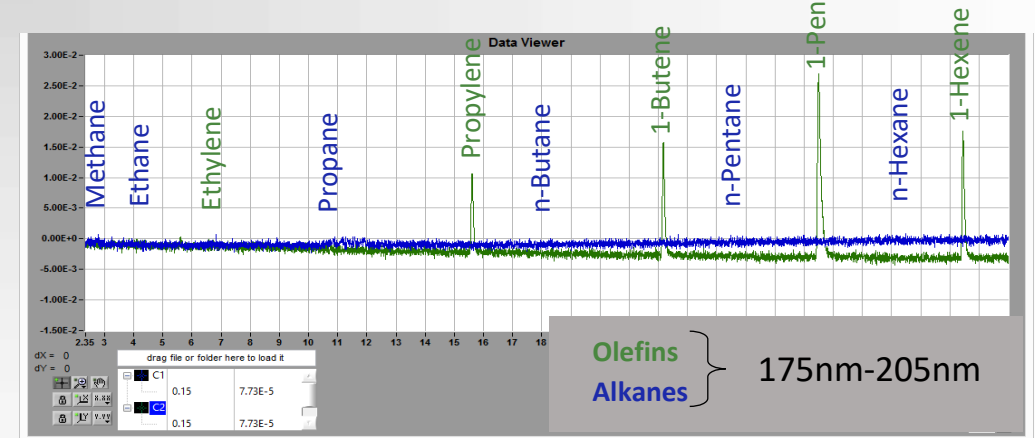
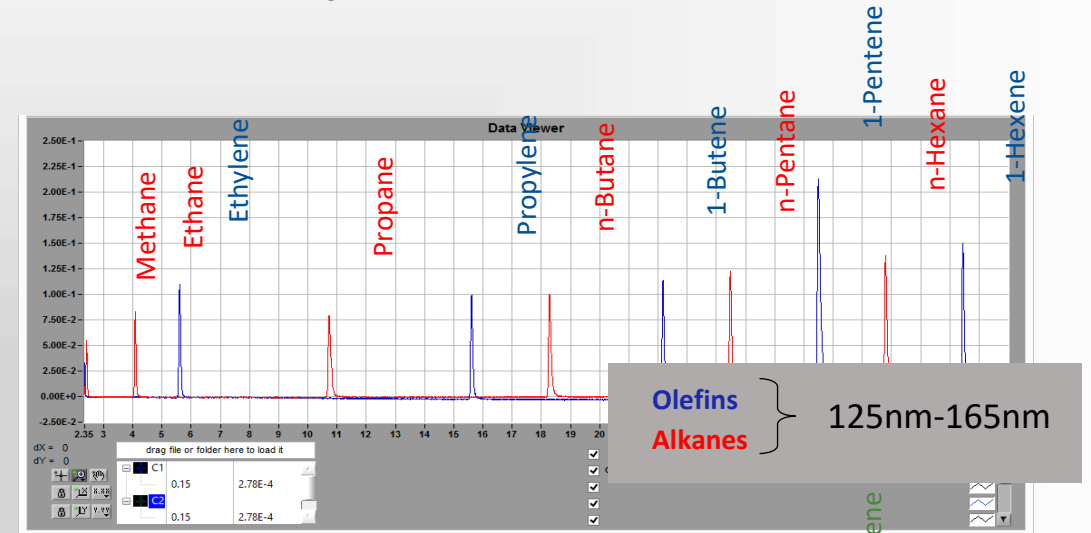
Spectral Filters – “Class Discrimination”



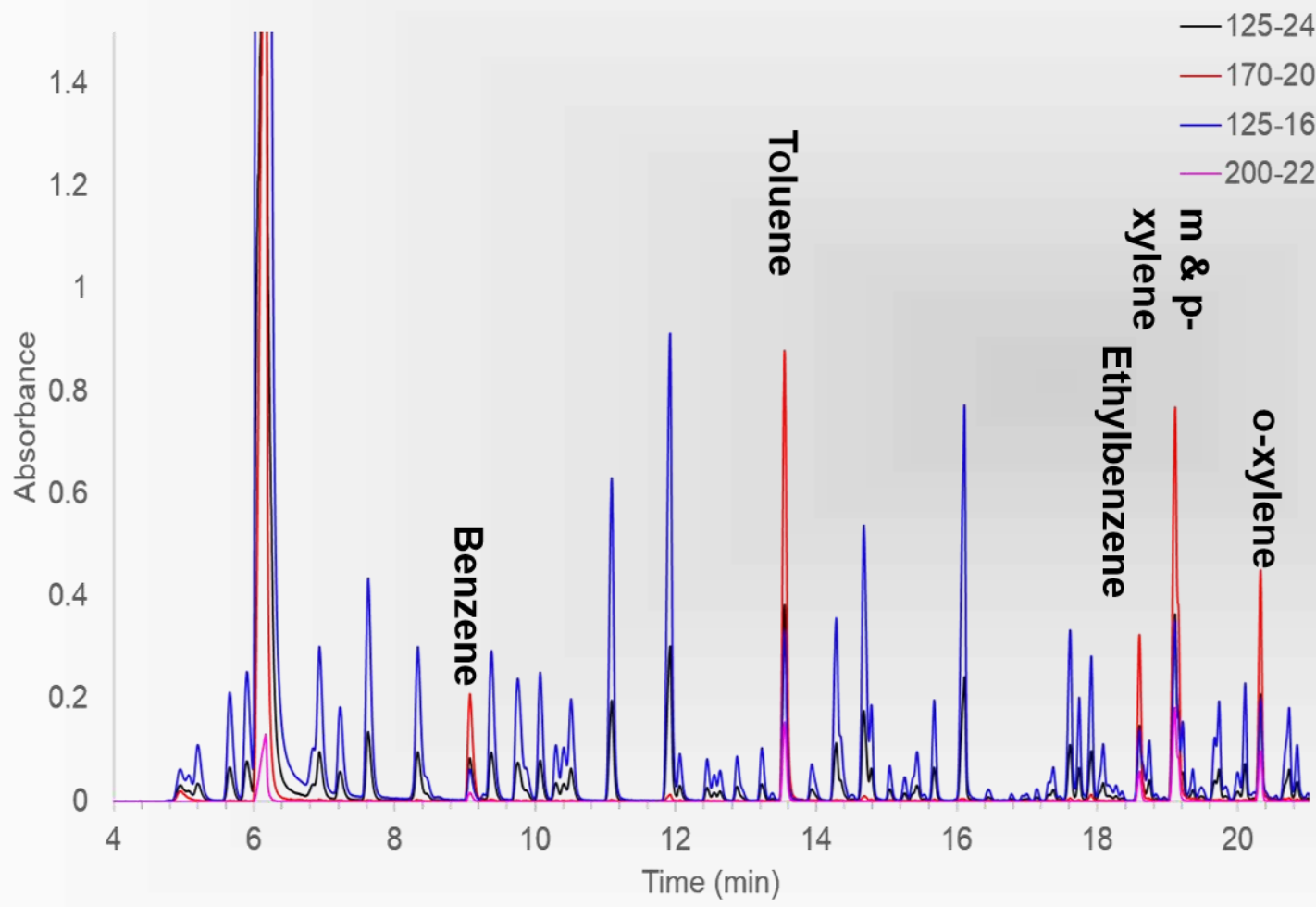
Spectral filters are useful because some compounds absorb more strongly than others in certain regions of the VUV spectrum

- Saturated aliphatic compounds absorb strongest in the 125-160 nm region
- Aromatics and other unsaturated compounds have characteristic λ_{Max} values in the 170-200 nm region

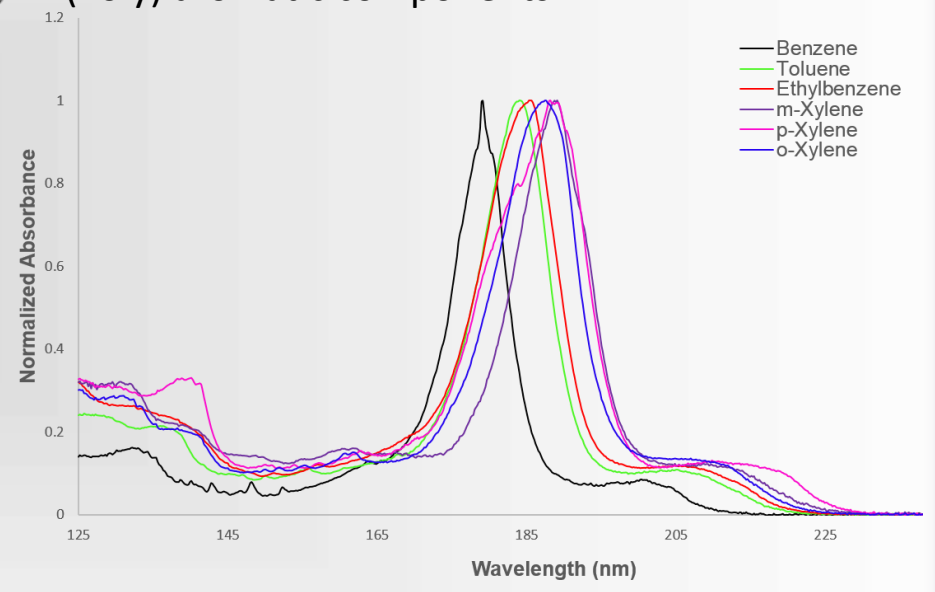
Overlay of Alkane and Olefin Standards



Peak ID and VUV Absorbance Spectra



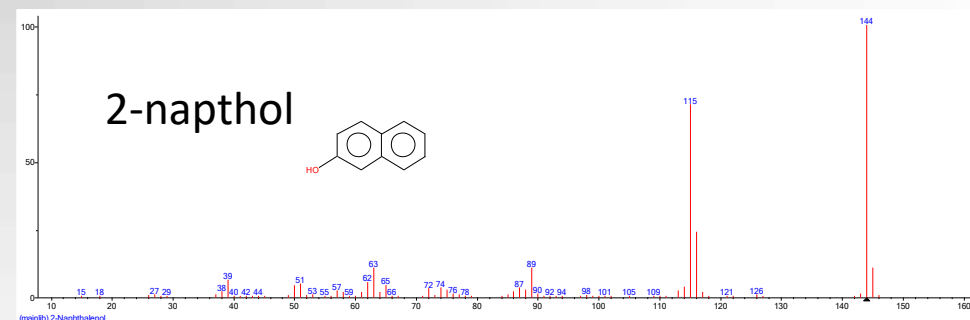
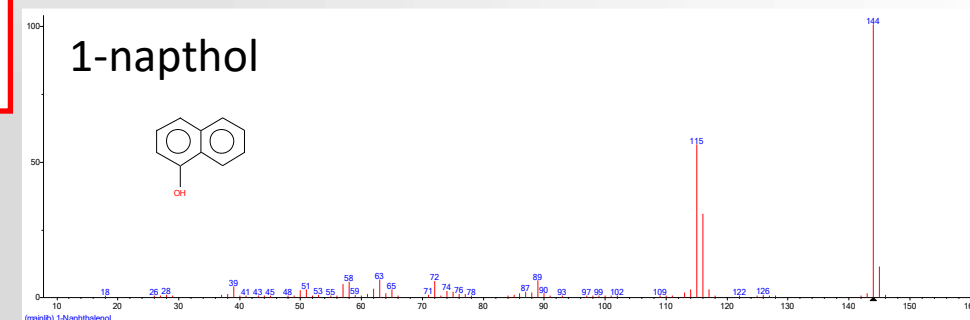
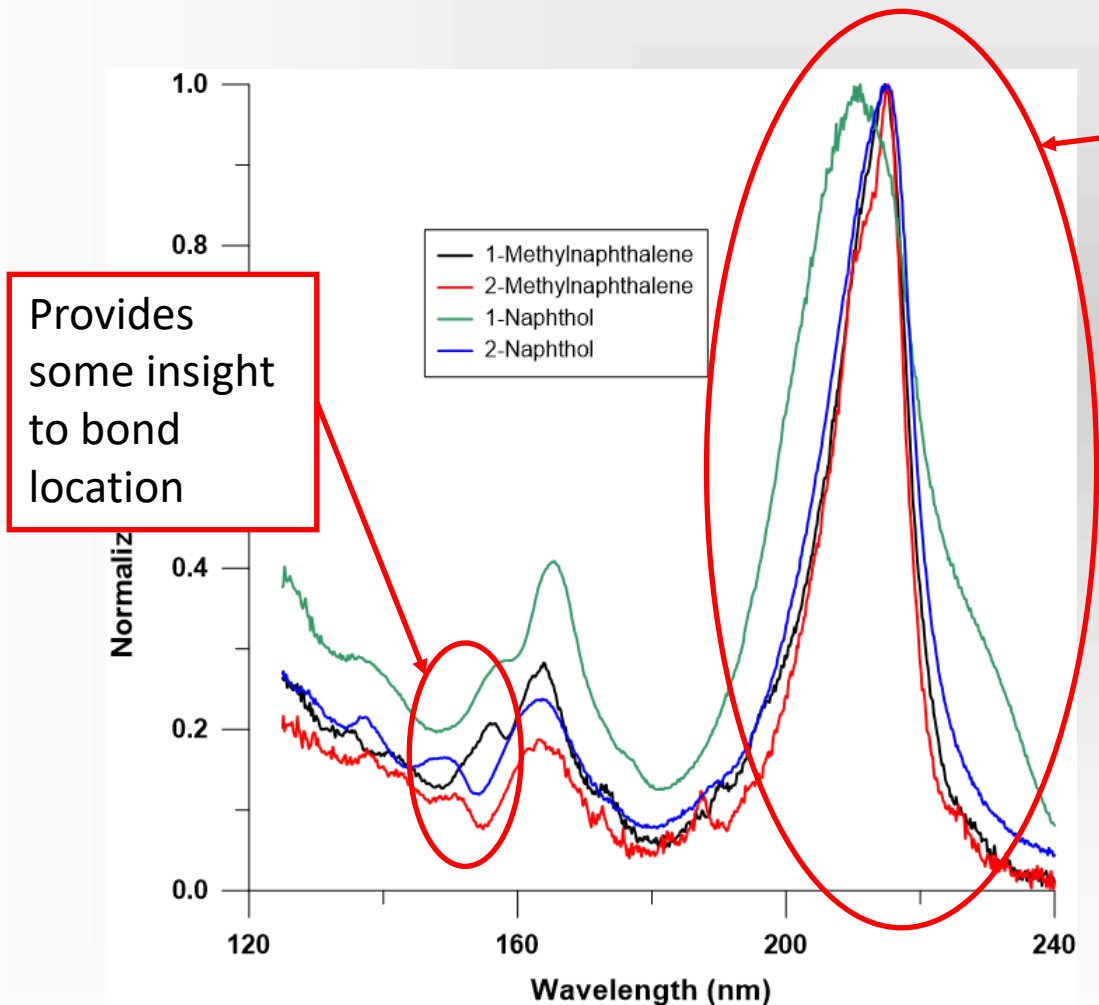
- 125-240 Full spectral data
- 170-200 Aromatic components
- 125-160 Aliphatic components
- 200-220 (Poly) aromatic components



Isomer Identification

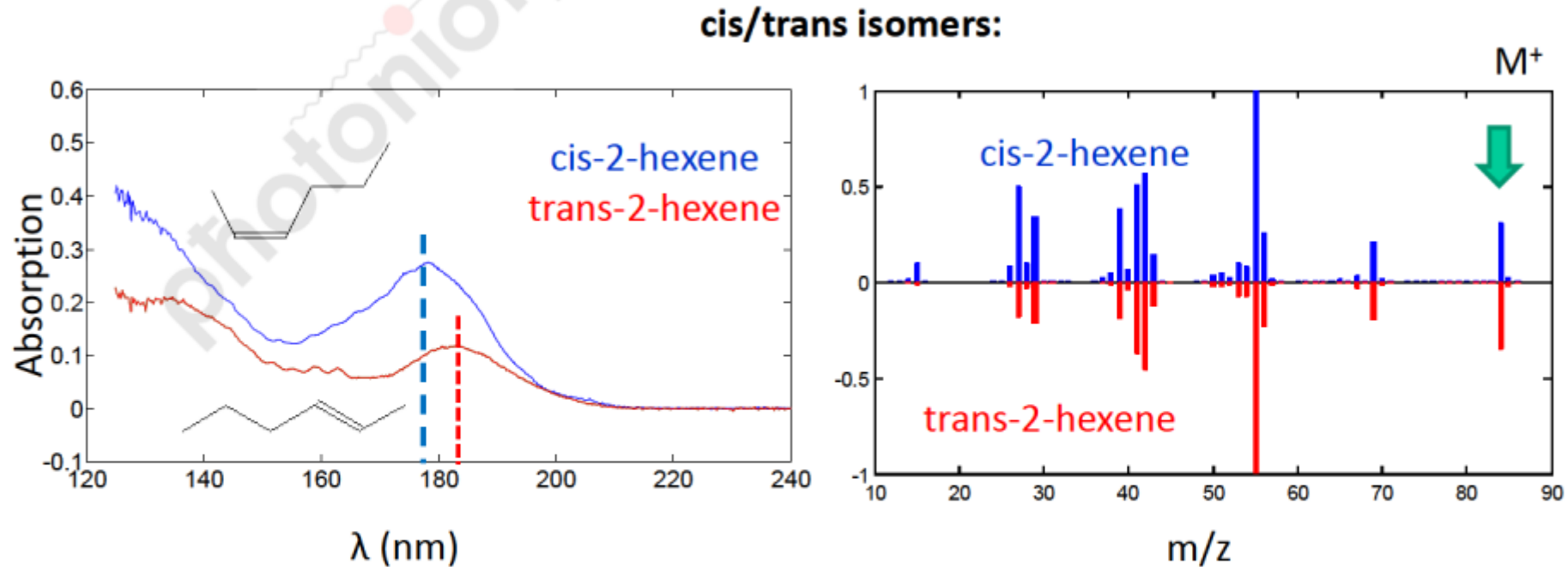
Isomer Identification

Many isomers have unique absorption spectra



Chiral compounds (i.e. optical isomers) are the exception

Spectral differences Cis-/Trans-



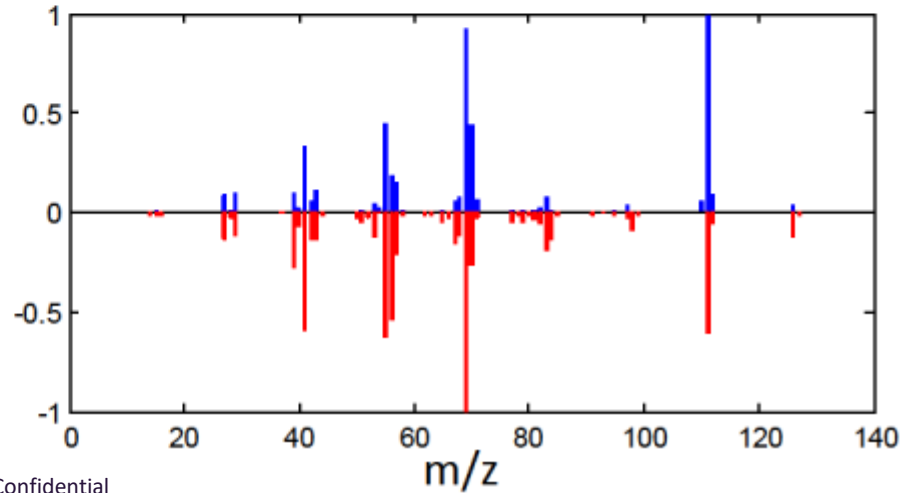
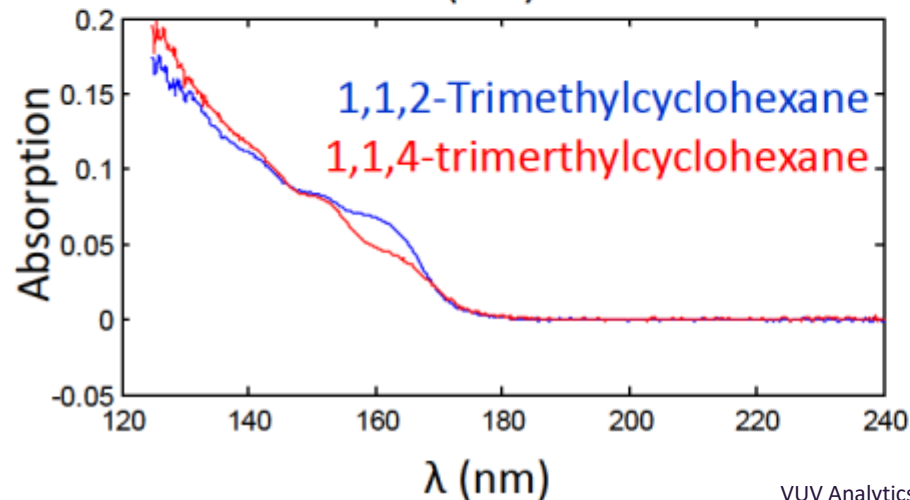
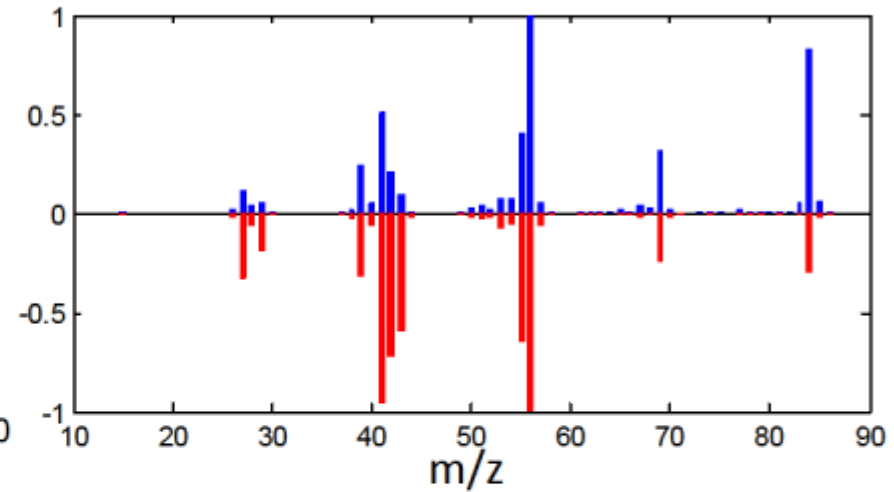
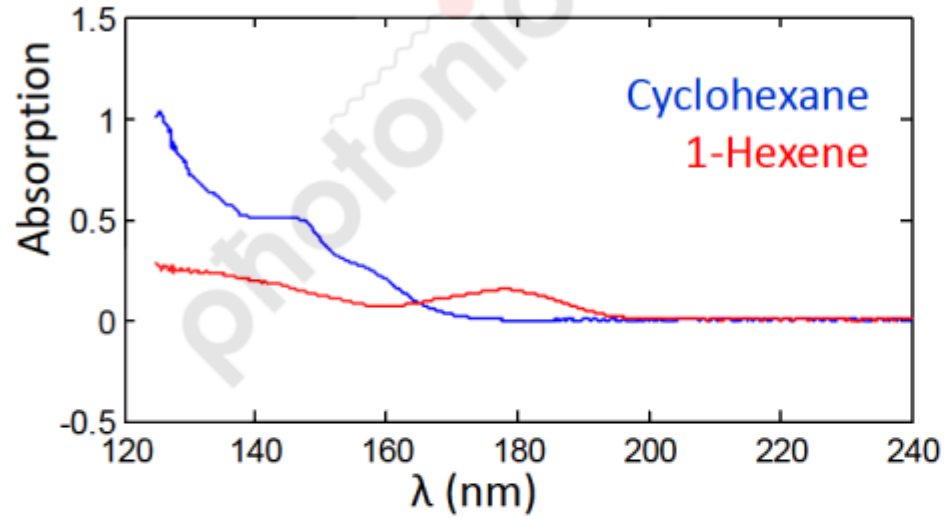
VUV Spectra show clear shift of maxima and differences in shape

- 70eV EI Spectra does not provide selective Information
- soft Ionization would also not provide additional information

Spectral differences

Constitutional isomers

Common separation and identification problem for (GCx)GC-MS (e.g. petro chemistry)



Application for this Universal Detector

Environmental



Specialty Gas



Oil & Gas



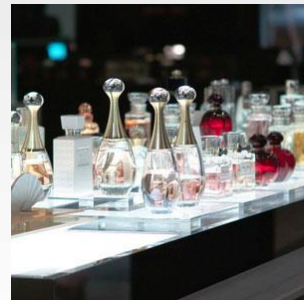
Agrochemical



Forensics



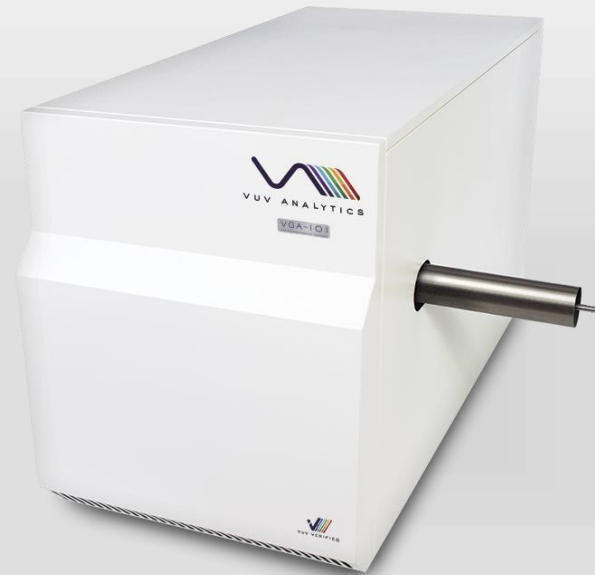
Fragrances & Flavors



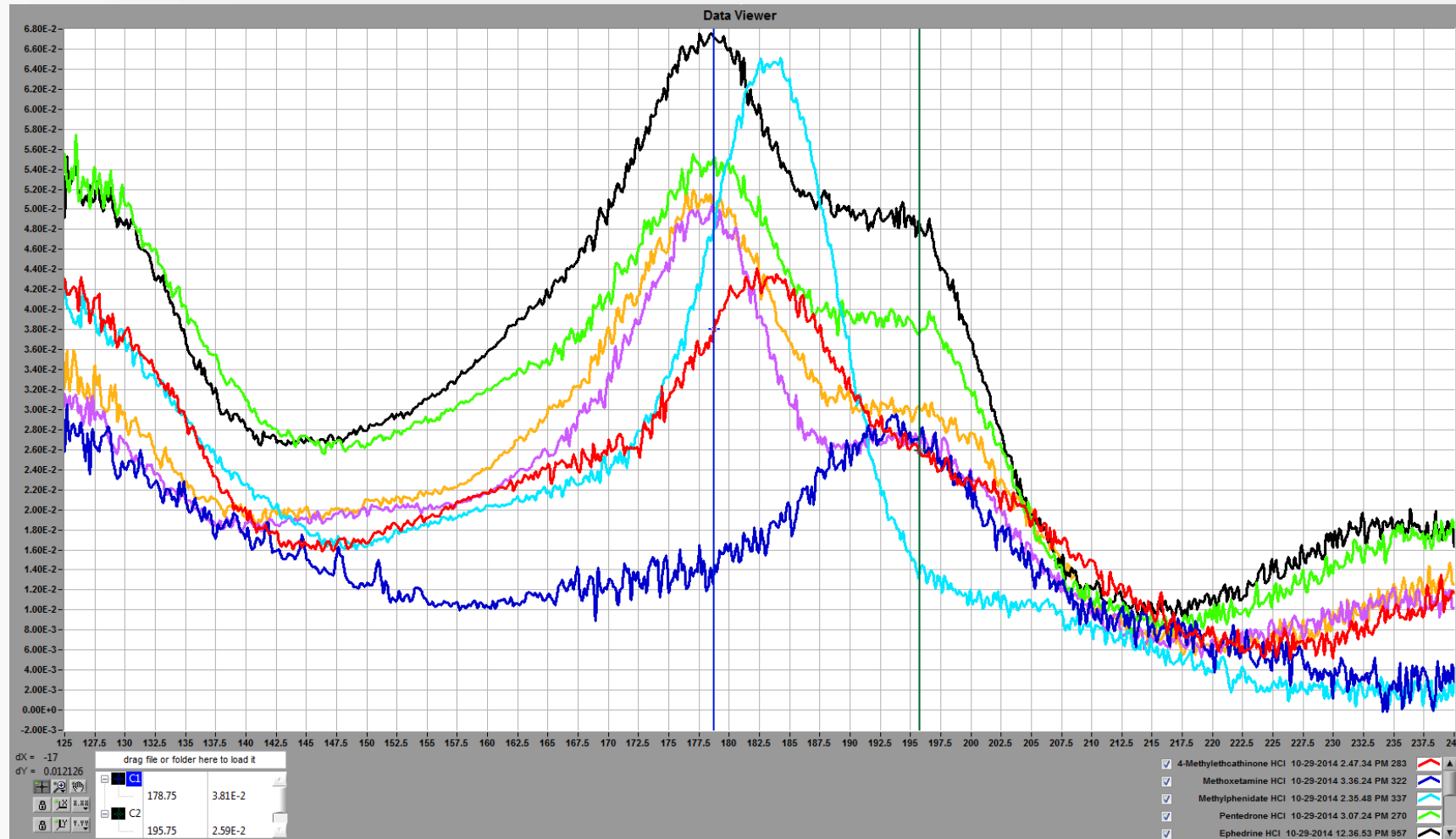
Petrochemical



Food & Beverage Safety

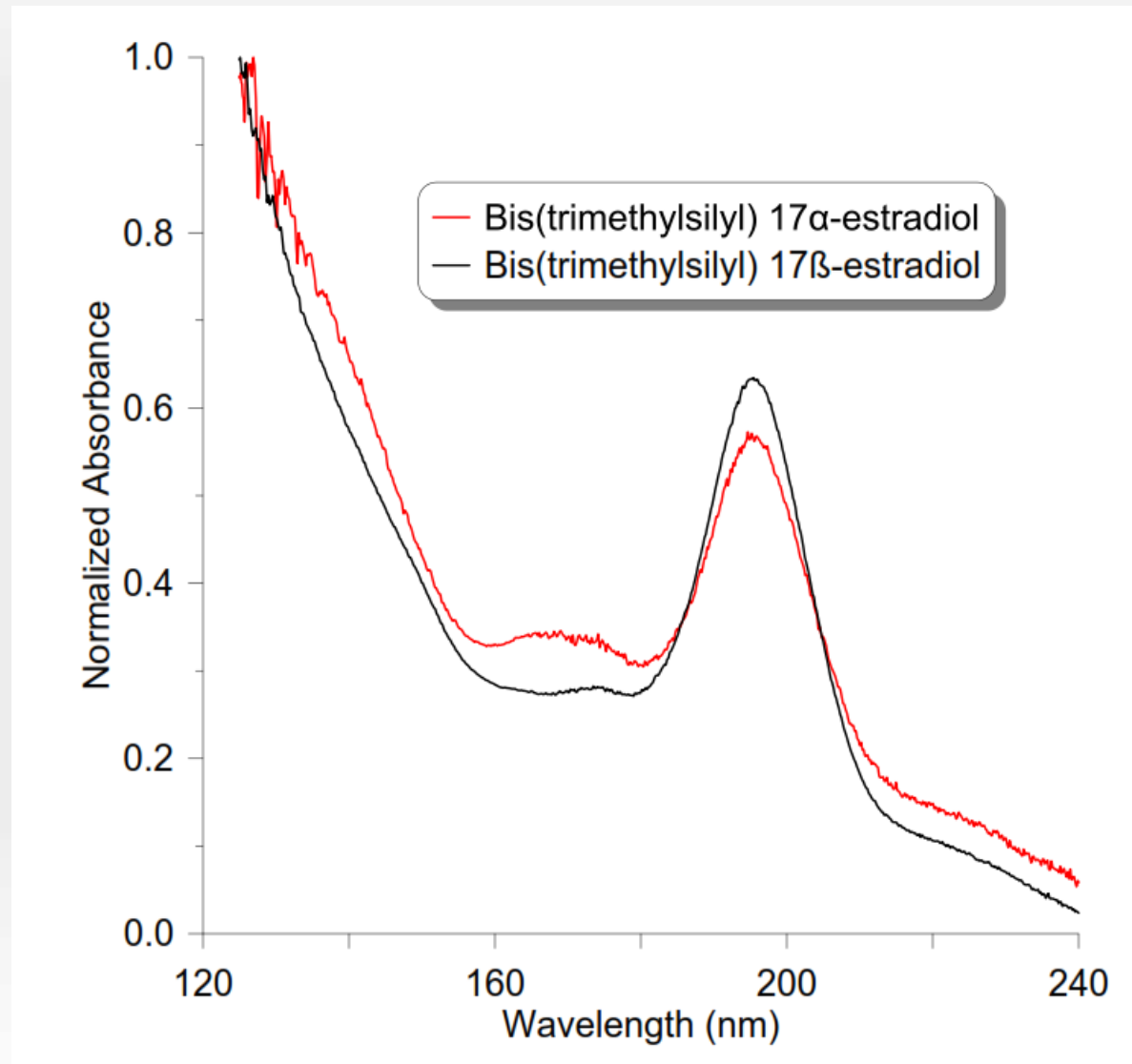


Drug Compounds

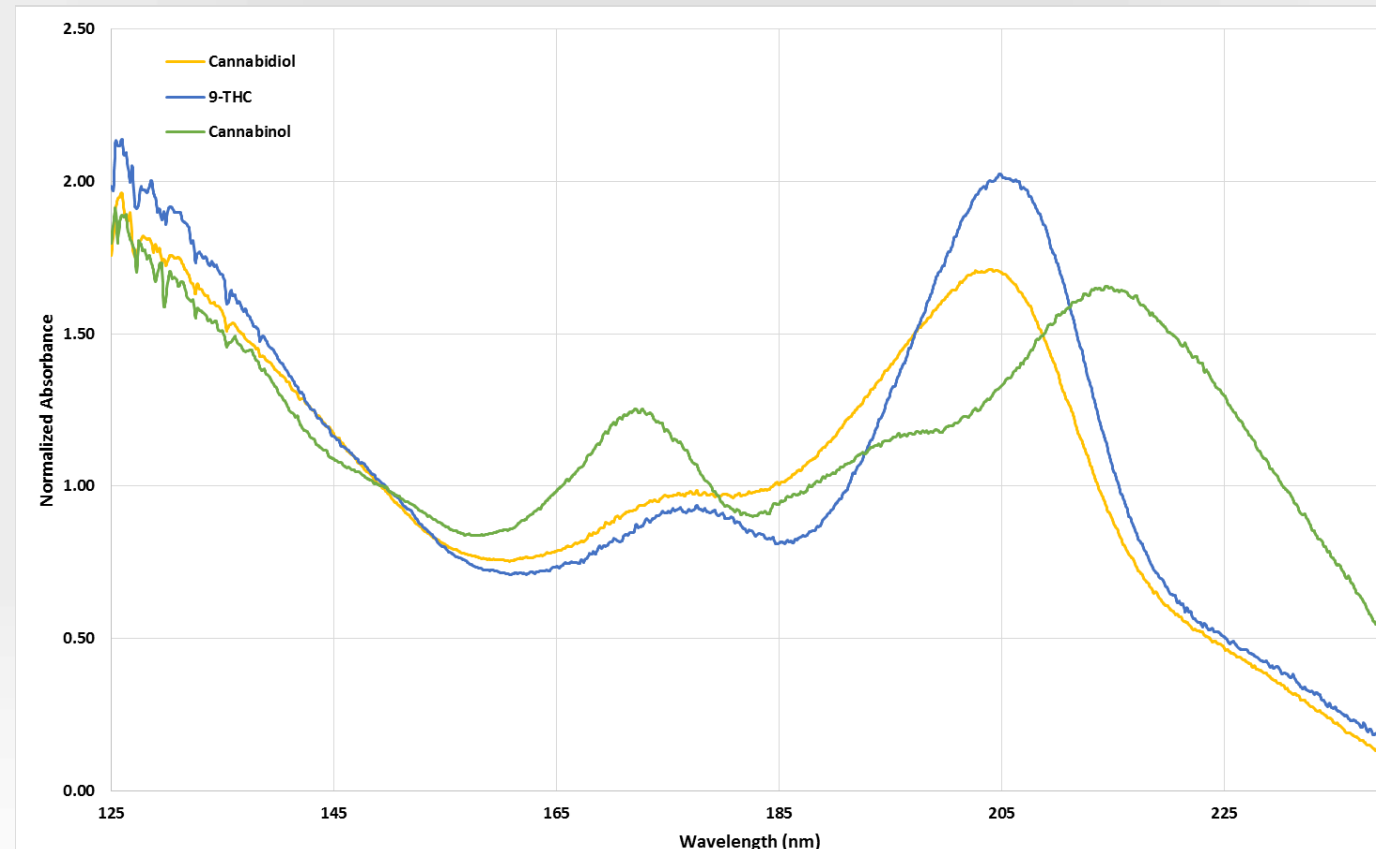
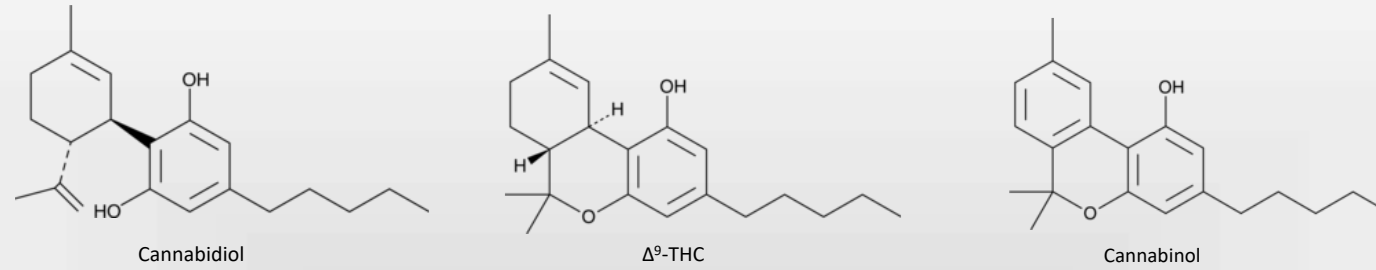


- 4-Methylethcathinone HCL
- Methoxetamine HCL
- Methylphenidate HCL
- Pentedrone HCL
- Ephedrine HCL

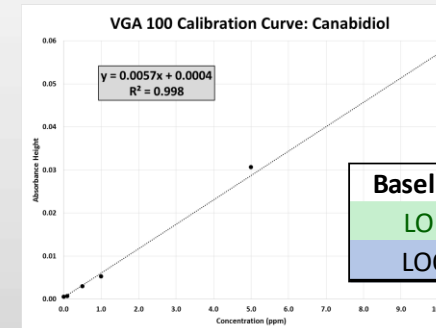
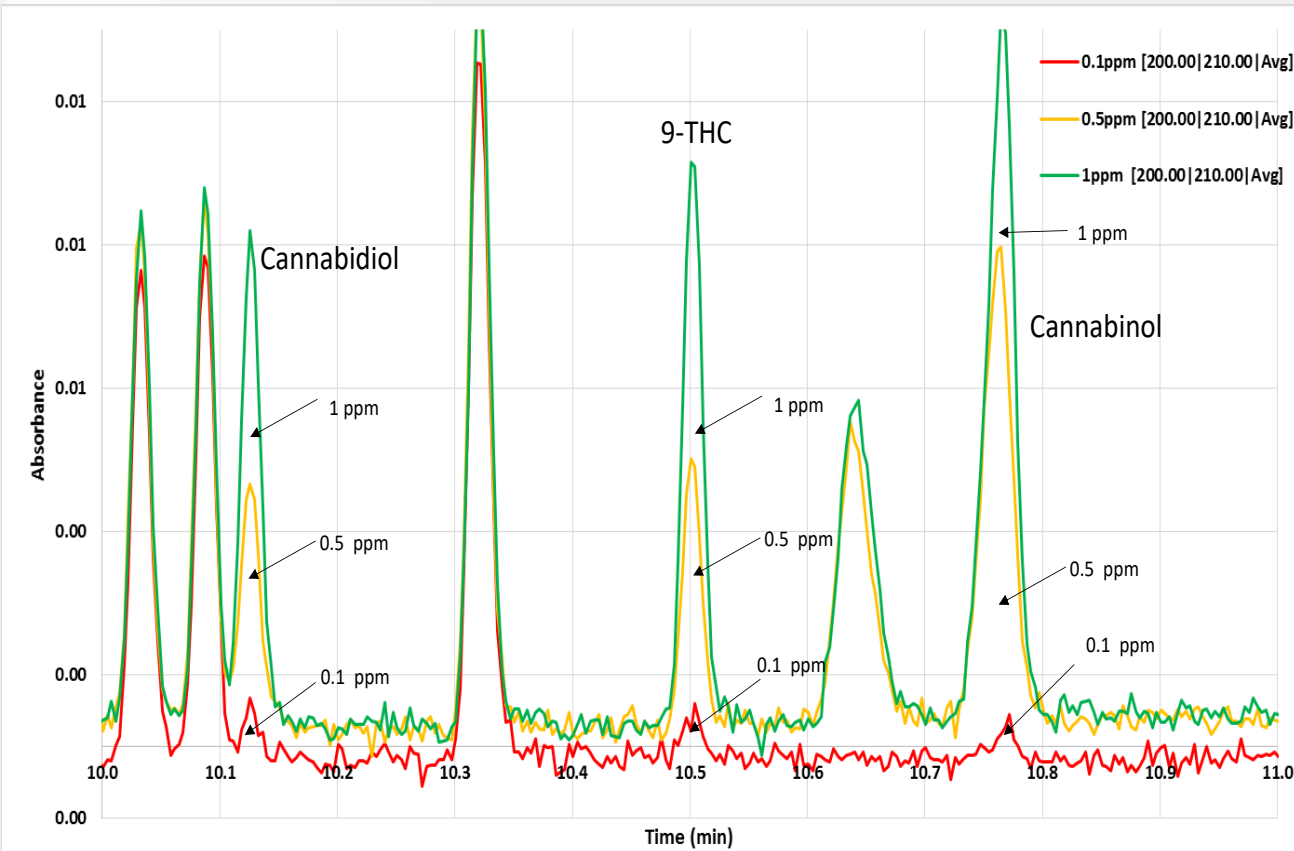
Estradiols



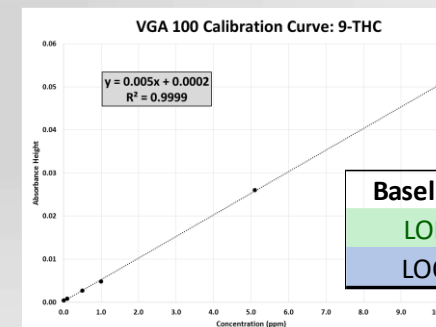
Cannabinoid Standard Analysis: Comparison of Spectra



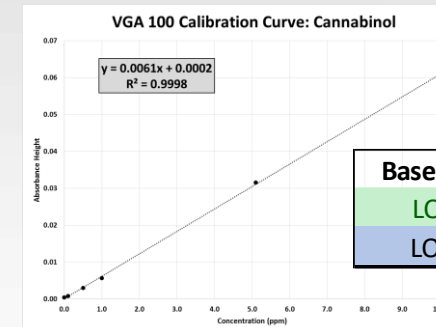
Cannabinoid Standard Analysis: Chromatographic and Quantitative Comparison



Baseline	ppm	ppb	Min ppb	Max ppb
LOD	0.0967	96.7	94.6	98.9
LOQ	0.3224	322.4	315.3	329.8



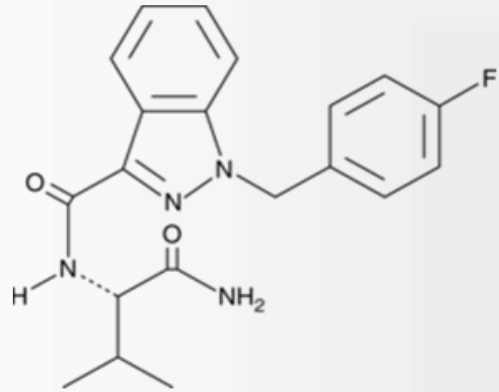
Baseline	ppm	ppb	Min ppb	Max ppb
LOD	0.0806	80.6	80.2	81.1
LOQ	0.2688	268.8	267.3	270.3



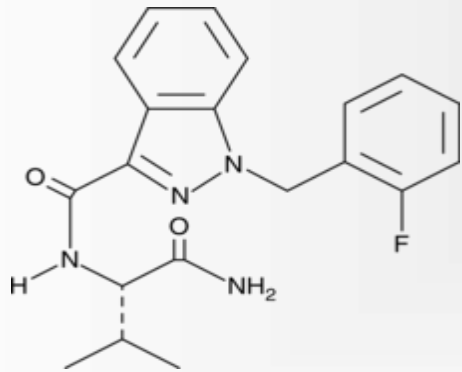
Baseline	ppm	ppb	Min ppb	Max ppb
LOD	0.0522	52.2	51.8	52.6
LOQ	0.1739	173.9	172.7	175.2

*Data acquired by VGA-100 detector

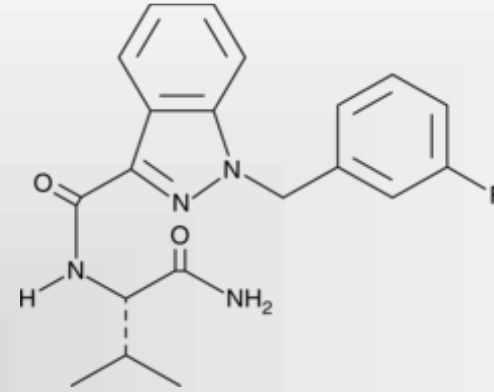
Synthetic Cannabinoid Analysis: Indazole Carboxamide Compound Class



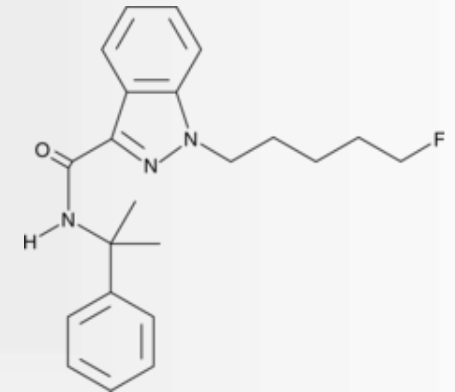
AB-FUBINACA



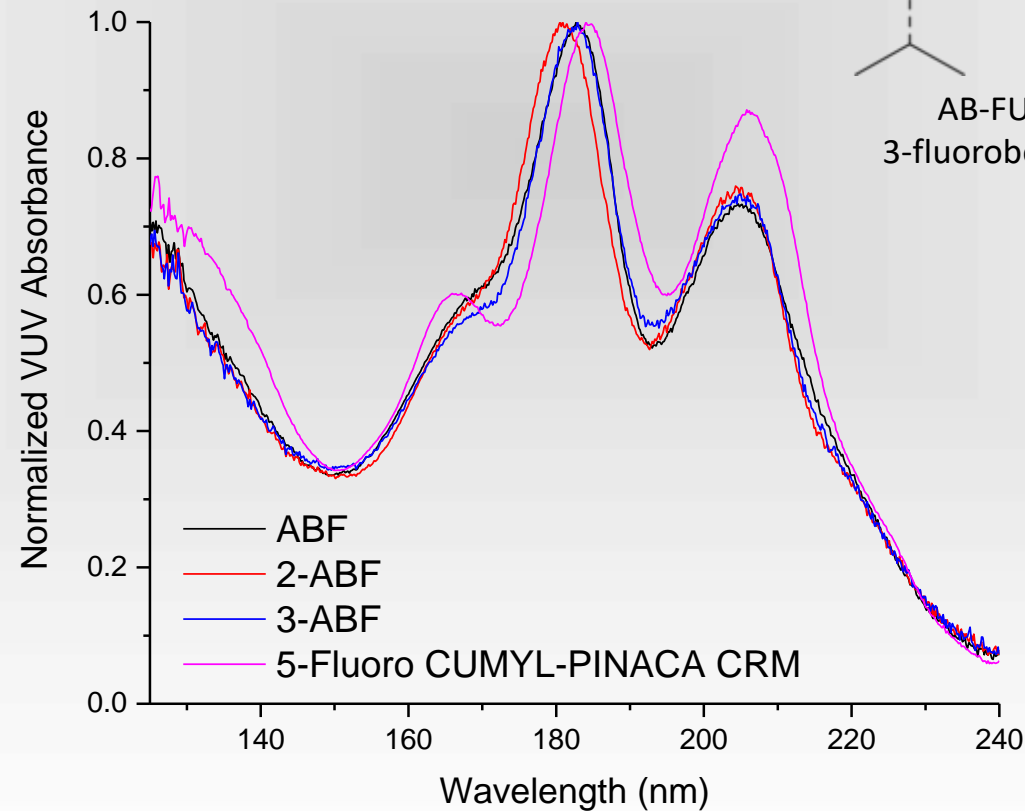
AB-FUBINACA
2-fluorobenzyl isomer



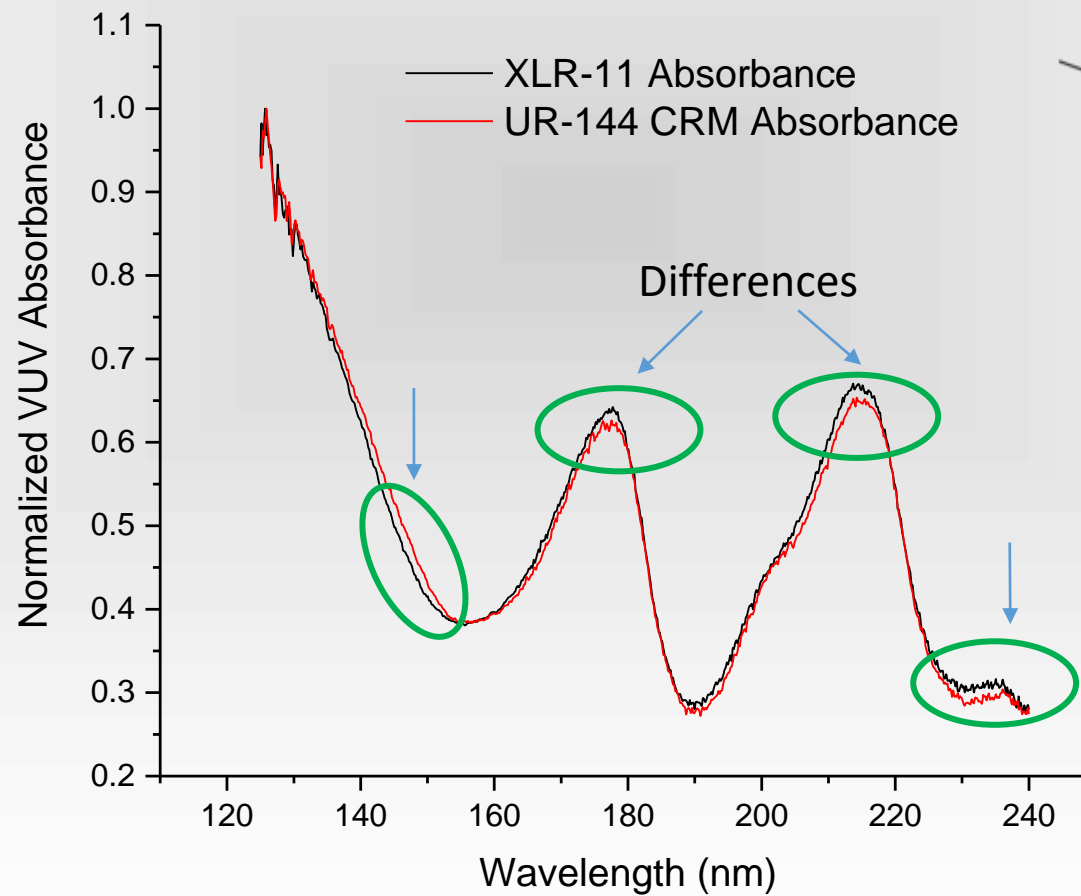
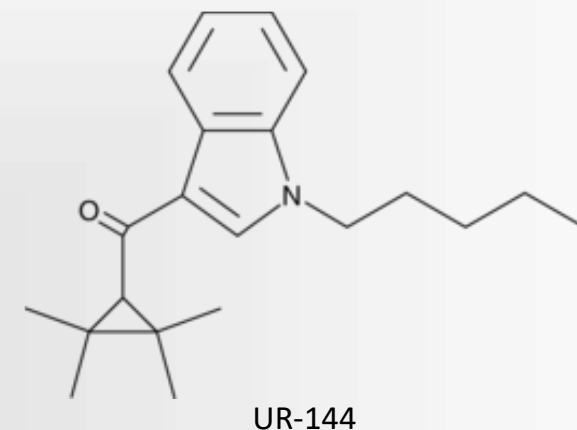
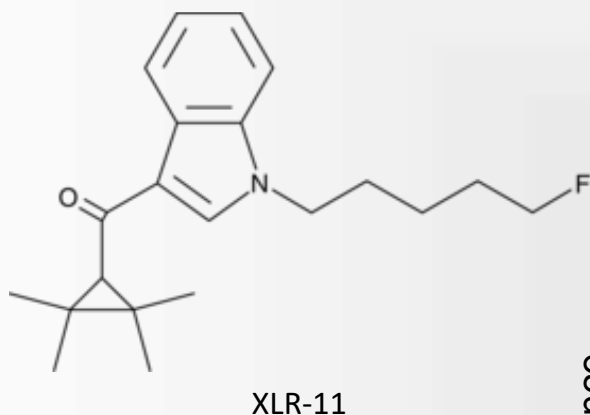
AB-FUBINACA
3-fluorobenzyl isomer



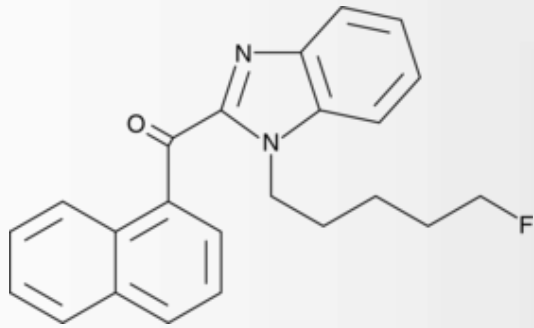
5-fluoro CUMYL-PINACA (CRM)



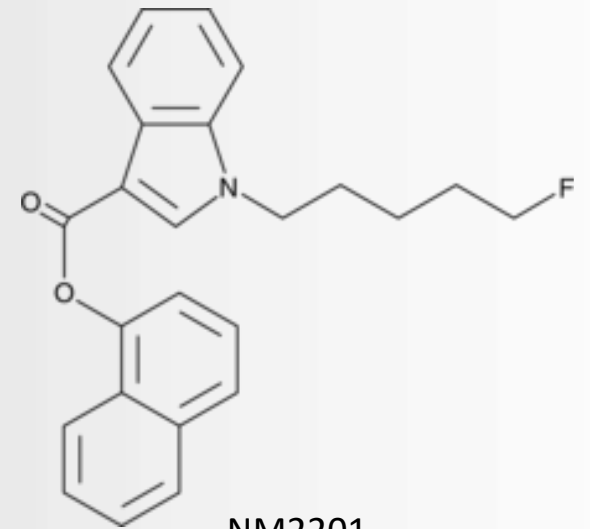
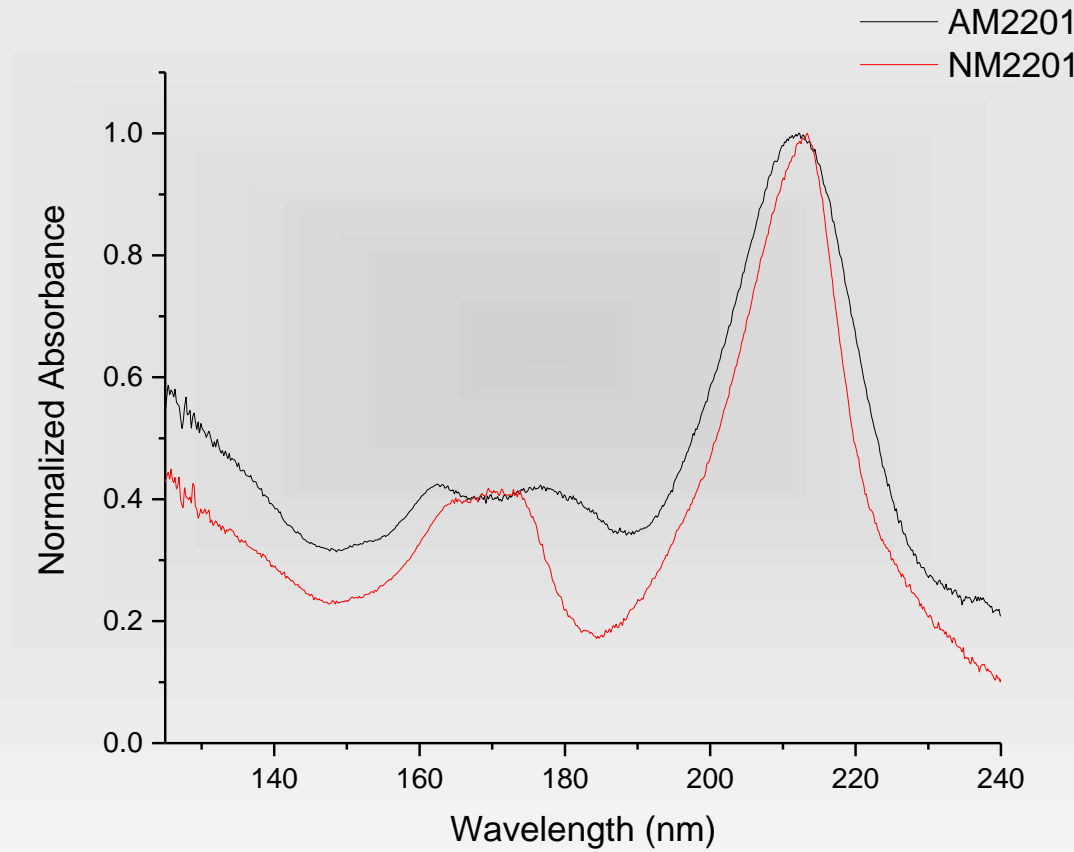
Synthetic Cannabinoid Analysis: Cyclopropyl Compound Class



Synthetic Cannabinoid Analysis: Naphthoylindole Compound Class

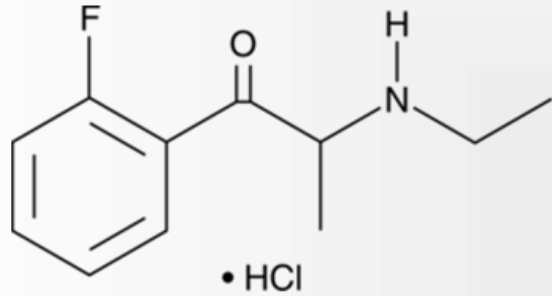


AM2201 benzimidazole analog

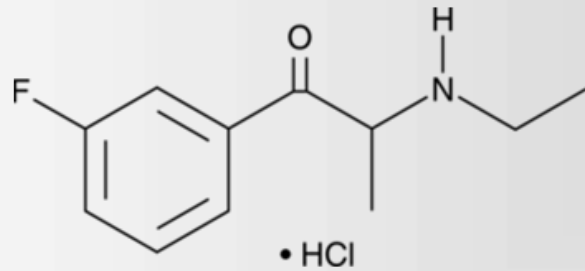


NM2201

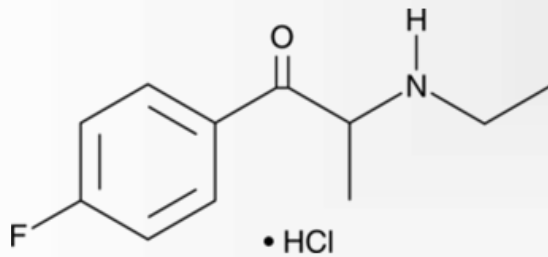
Stimulant Analysis: Cathinone Compound Class



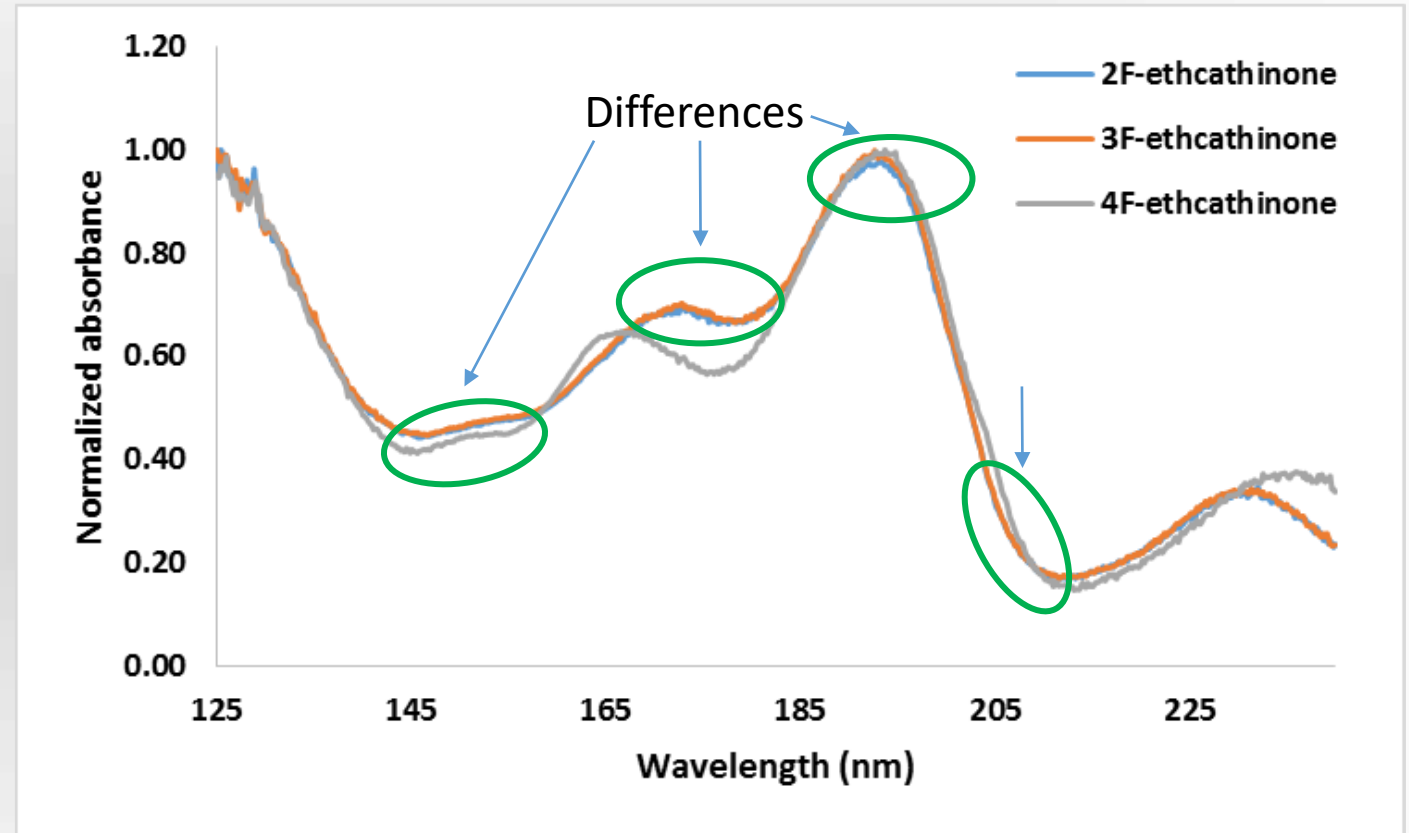
2F-ethcathinone



3F-ethcathinone



4F-ethcathinone



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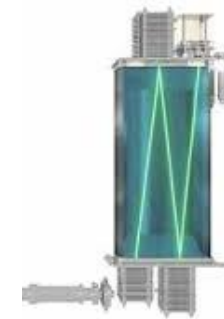
6600+ & SelexION



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SCIEX TripleTOF® and X500 Series



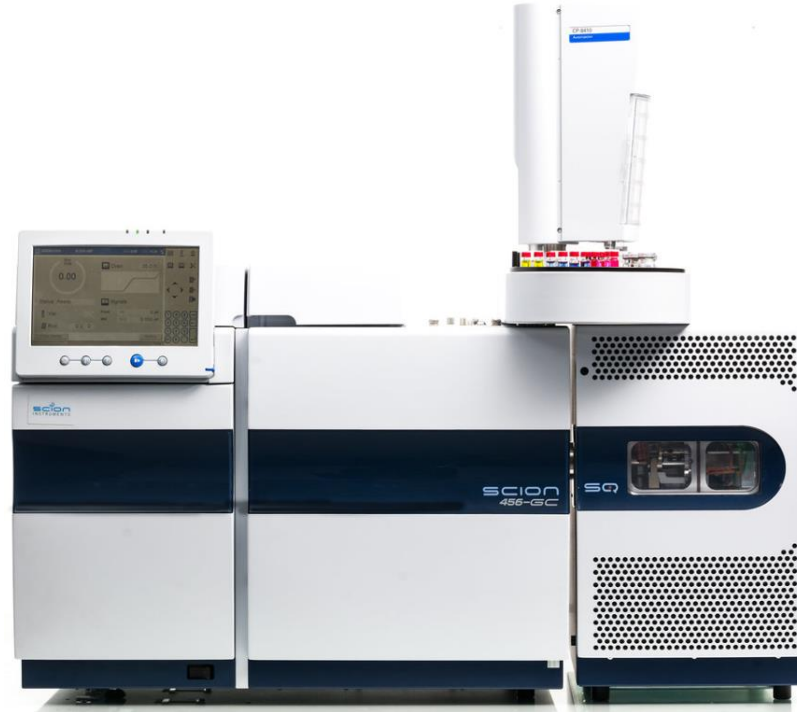
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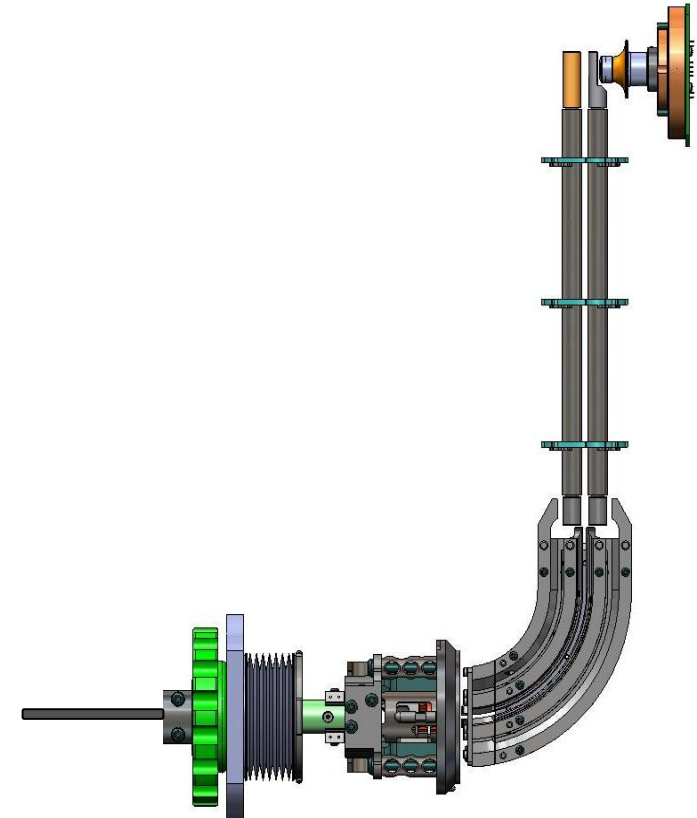
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