



ThermoFisher
SCIENTIFIC

Ion Chromatography for Pharmaceutical Analysis

Jeffrey Rohrer, Ph. D.

The world leader in serving science

- Introduction to Ion Chromatography (IC)
- What IC Offers for Pharmaceutical Analysis
- Review of the Two IC Applications for Pharmaceutical Analysis
- How to Develop an IC for Pharmaceuticals Assay and an Example

At the most basic level....

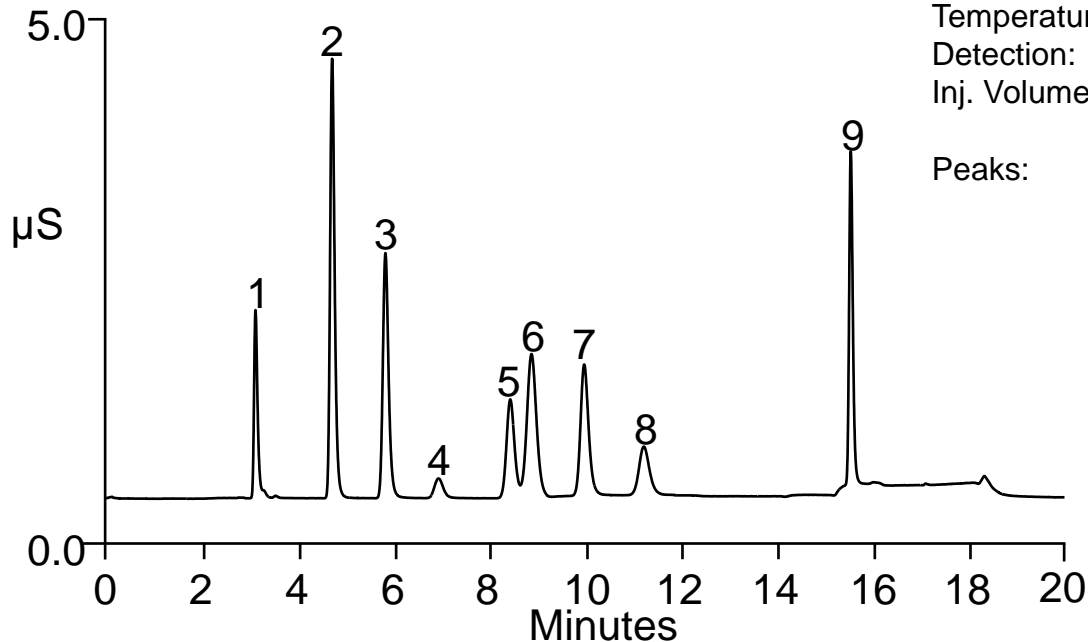
Ion Chromatography = Ion-
Exchange + Chemically
Suppressed Conductivity

Separation of Common Anions and TFA

Column:
Thermo Scientific™ Dionex™ IonPac™ AG18 and AS18

Eluent:
22 mM KOH for 0–6 min,
28 mM KOH from 6–12 min,
50 mM KOH for 12–15 min,
22 mM KOH from 15–20 min

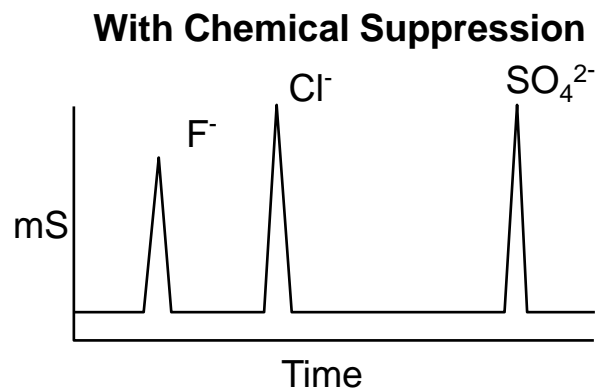
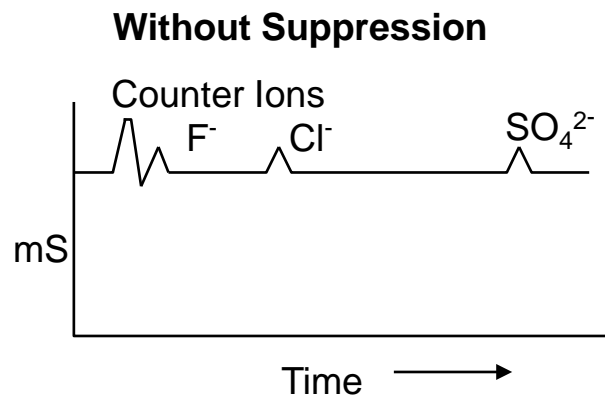
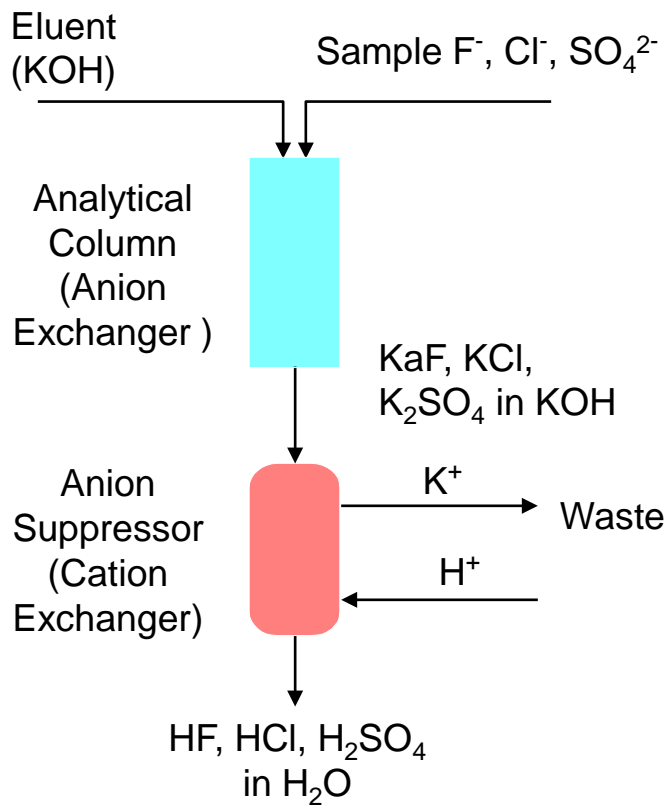
Eluent Source: EG50
Eluent Flow Rate: 1.0 mL/min
Temperature: 30 °C
Detection: Suppressed conductivity
Inj. Volume: 5 µL



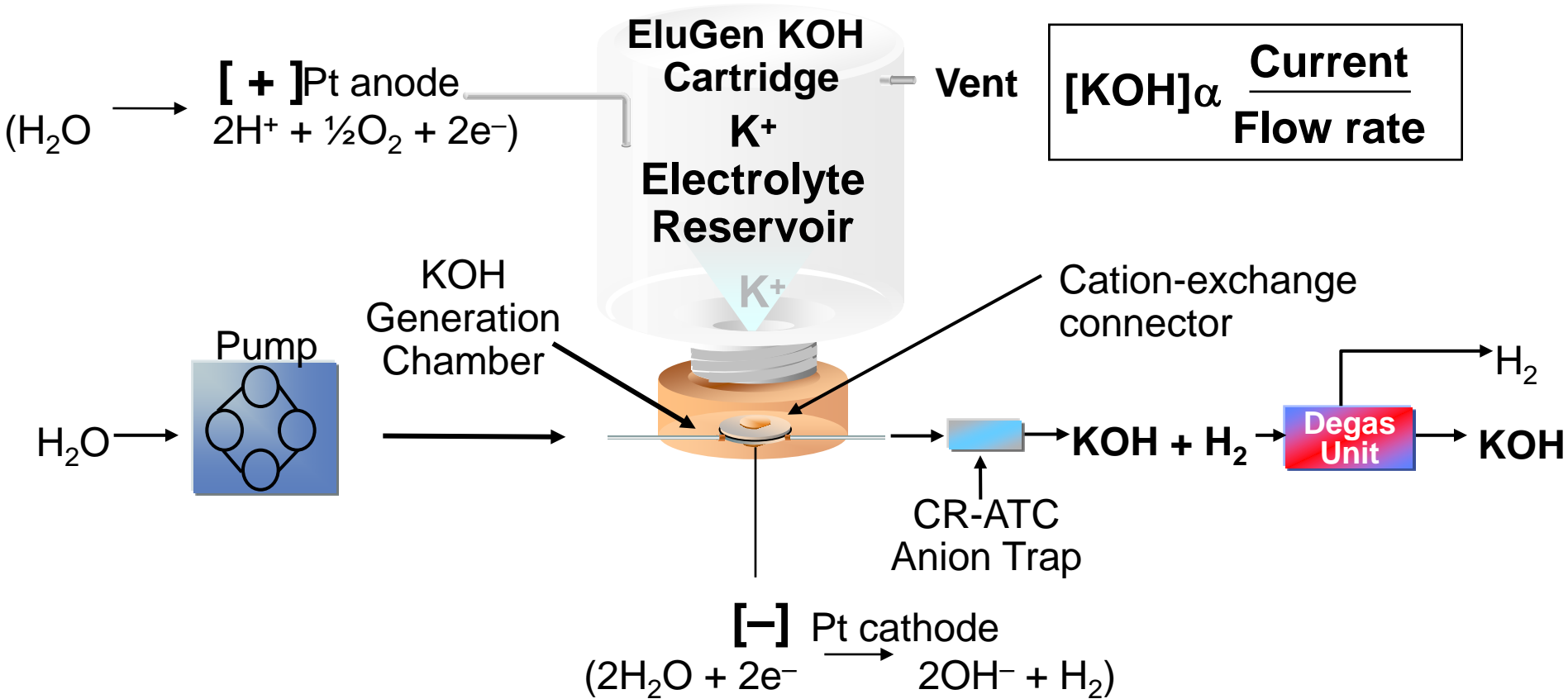
Peaks:

1.	Fluoride	2 mg/L
2.	Chloride	4
3.	Nitrite	10
4.	Carbonate	–
5.	Bromide	10
6.	Sulfate	10
7.	Nitrate	10
8.	Trifluoroacetate	10
9.	Phosphate	20

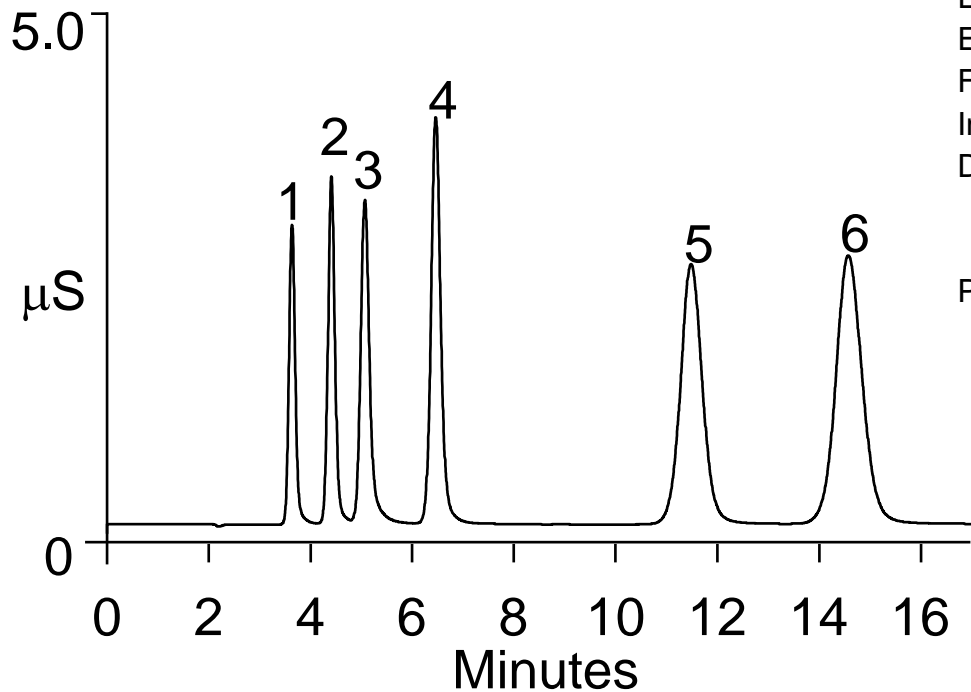
The Role of Chemical Suppression (KOH)



Hydroxide Eluent Generation



Separation of the Common Cations



Column: Thermo Scientific™ Dionex™
IonPac™ CG12A, CS12A, 4 mm

Eluent: 18 mM Methanesulfonic acid

Eluent Source: Eluent Generator

Flow Rate: 1.0 mL/min

Injection: 25 μL

Detection: Suppressed conductivity,
Thermo Scientific™ Dionex™ SC-CSRSTM
ULTRA suppressor, recycle mode

Peaks:

1. Lithium	0.5	mg/L (ppm)
2. Sodium	2.0	
3. Ammonium	2.5	
4. Potassium	5.0	
5. Magnesium	2.5	
6. Calcium	5.0	

- **Conductivity:**
 - **Suppressed**
 - (Non suppressed)

- **UV detection:**
 - **Direct**
 - (Indirect)
 - **Post column derivatization**

- **Amperometry:**
 - Direct current amperometry (DC)
 - **Integrated amperometry (PAD and IPAD)**

- **Mass spectrometry**

HPAE-PAD

High-Performance Anion-Exchange
Chromatography with Pulsed Amperometric
Detection

- Carbohydrates are separated as oxyanions at high pH (>12)
- These separations require hydroxide eluents
- If the carbohydrate is **charged**, acetate or another strong eluent must be added to the hydroxide eluent to elute the carbohydrate

- Carbohydrates are detected on a Au working electrode (WE) at high pH by PAD
- PAD applies a series of potentials (a waveform) to a WE and the carbohydrate is detected by its oxidation at 1 potential
- The waveform is applied at a frequency of 2 Hz, i.e. two times a second
- **Therefore carbohydrates are separated and detected without derivatization, i.e. a direct analysis**

- **Anions:** Chloride, sulfate, fluoride, **nitrite**, **nitrate**, bromide, iodide, bromate, chlorite, chlorate, perchlorate, sulfite, thiosulfate, cyanide, thiocyanate, cyanate, sulfide, benzoate, acetate, formate, silicate, glycolate, oxalate, iodate, lactate, trifluoroacetate, numerous other organic acids and inorganic anions, **carbohydrates**, amino acids
- **Cations:** **Lithium**, sodium, potassium, ammonium, **calcium**, magnesium, barium, strontium, methylamine, dimethylamine, trimethylamine, ethanolamine, diethanolamine, triethanolamine, choline, many transition metals, and numerous amines

What IC Offers for Pharmaceutical Analysis

- Easy (direct) determination of analytes lacking chromophores
- Opportunity to have more automated assays compared to HPLC
- Usually requires no organic solvents
- Separation modes better suited for some analytes
- Counter ion analysis of salt form drug substances to confirm ID and API content

How IC has been Used in Pharmaceutical Analysis

- Assay
- Determination of impurities and degradation products – limit tests and related substances tests for drug substances and drug products
- Counter ion analysis of salt form drug substances to confirm ID and API content
- Excipient analysis

Example IC Methods in the USP-NF

- Assays of Kanamycin B and Amikacin in DS and DP monographs
- USP-NF <345> Assay for Citric Acid/Citrate and Phosphate
- Risedronate Sodium Assay
- Cefepime Hydrochloride—Limit of *N*-methylpyrrolidine
- Methacholine Chloride – Assay and limit of Acetylcholine Test
- Heparin Sodium – Organic Impurities Test
- Sodium Bicarbonate – Limit of Ammonia Test

- Published in 2015
- Eliminated flame tests
- Eliminated wet chemical tests that yielded poor results
- Added better wet chemical tests – EP harmonization too
- Added instrumentation options for identification tests – including ion chromatography and other forms of chromatography.

Example #1

Anion IC Used for an Assay and a Limit Test
of a Drug Substance

- The method was published Pharmacopeia Forum 40(5) as part of a modernization proposal for the USP Sodium Nitrite monograph
- Sodium nitrite part of the treatment for acute cyanide poisoning
- The IC method assays nitrite and would replace a titration with potassium permanganate
- The same method determines nitrate impurity
- We replicated the proposed method in our laboratory, though we used eluent generation

IC Separation – Sodium Nitrite USP Monograph

Column: Thermo Scientific™ Dionex™ IonPac™ AS12A Analytical, 4 x 250 mm
Thermo Scientific™ Dionex™ IonPac™ AG12A Guard, 4 x 50 mm

Eluent: 2.7 mM K₂CO₃ / 0.3 mM KHCO₃

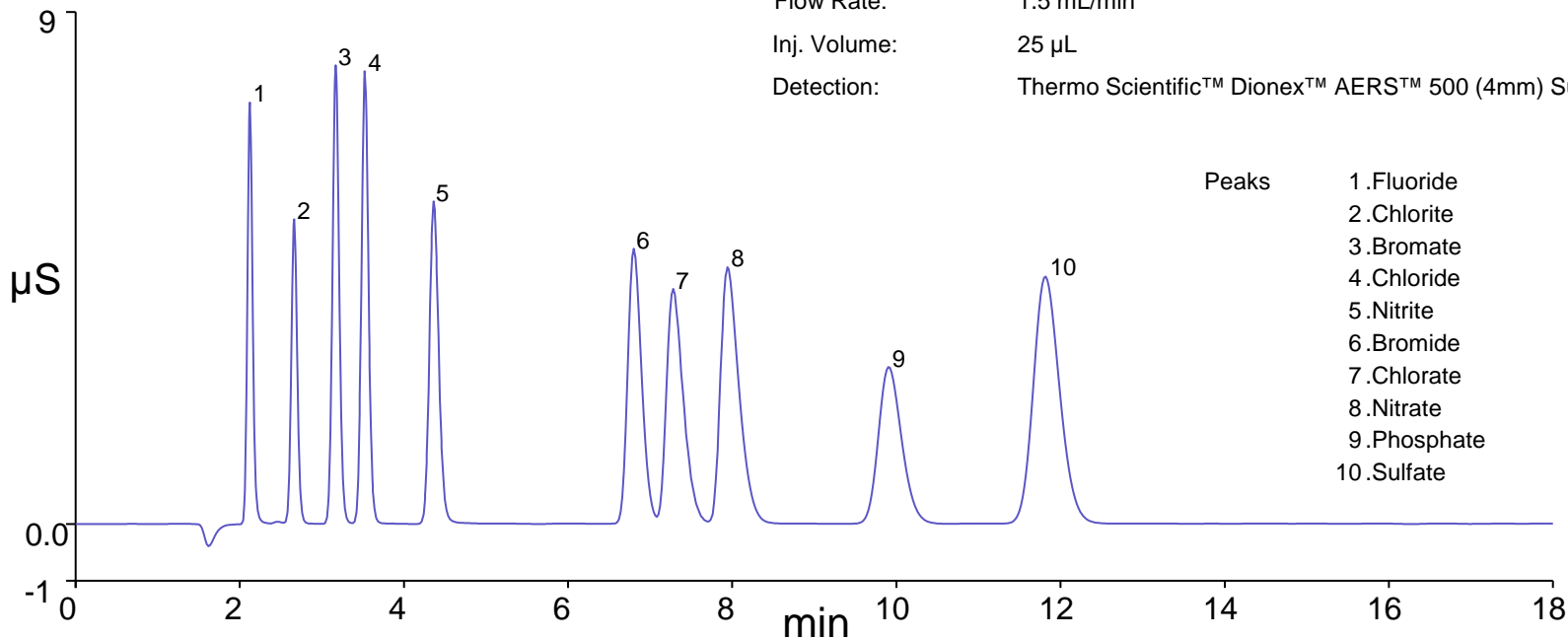
Eluent Source: Thermo Scientific™ Dionex™ EGC 500 K₂CO₃ cartridge with EPM 500

Temperature: Ambient (~24 °C)

Flow Rate: 1.5 mL/min

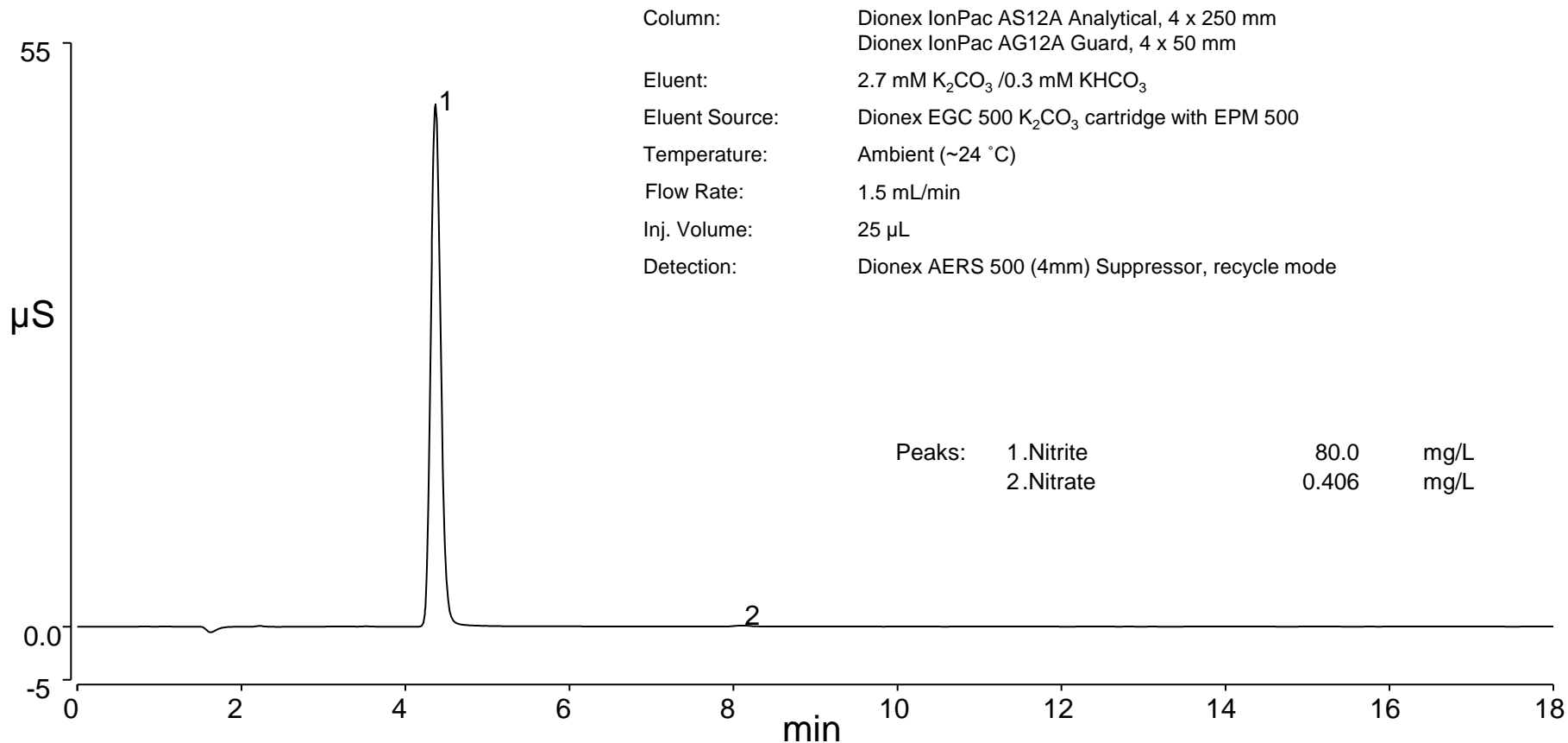
Inj. Volume: 25 µL

Detection: Thermo Scientific™ Dionex™ AERS™ 500 (4mm) Suppressor, recycle mode



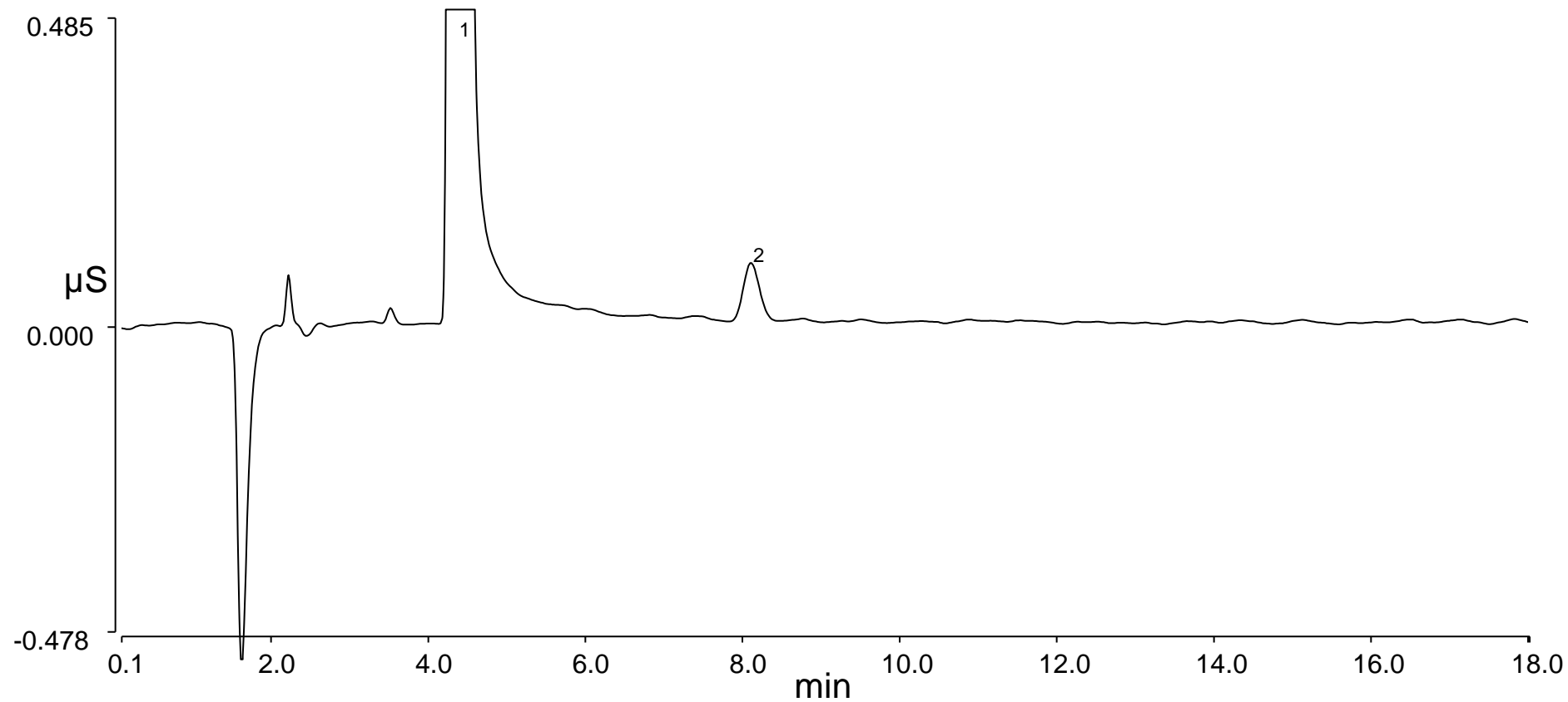
Peaks		
1.	Fluoride	3 mg/L
2.	Chlorite	10
3.	Bromate	20
4.	Chloride	6
5.	Nitrite	10
6.	Bromide	20
7.	Chlorate	20
8.	Nitrate	20
9.	Phosphate	30
10.	Sulfate	20

Sodium Nitrite Assay by the Proposed USP Monograph



Column: Dionex IonPac AS12A Analytical, 4 x 250 mm
Dionex IonPac AG12A Guard, 4 x 50 mm
Eluent: 2.7 mM K_2CO_3 / 0.3 mM KHCO_3
Eluent Source: Dionex EGC 500 K_2CO_3 cartridge with EPM 500
Temperature: Ambient ($\sim 24^\circ\text{C}$)
Flow Rate: 1.5 mL/min
Inj. Volume: 25 μL
Detection: Dionex AERS 500 (4mm) Suppressor, recycle mode

Enlarged to View the Nitrate Peak

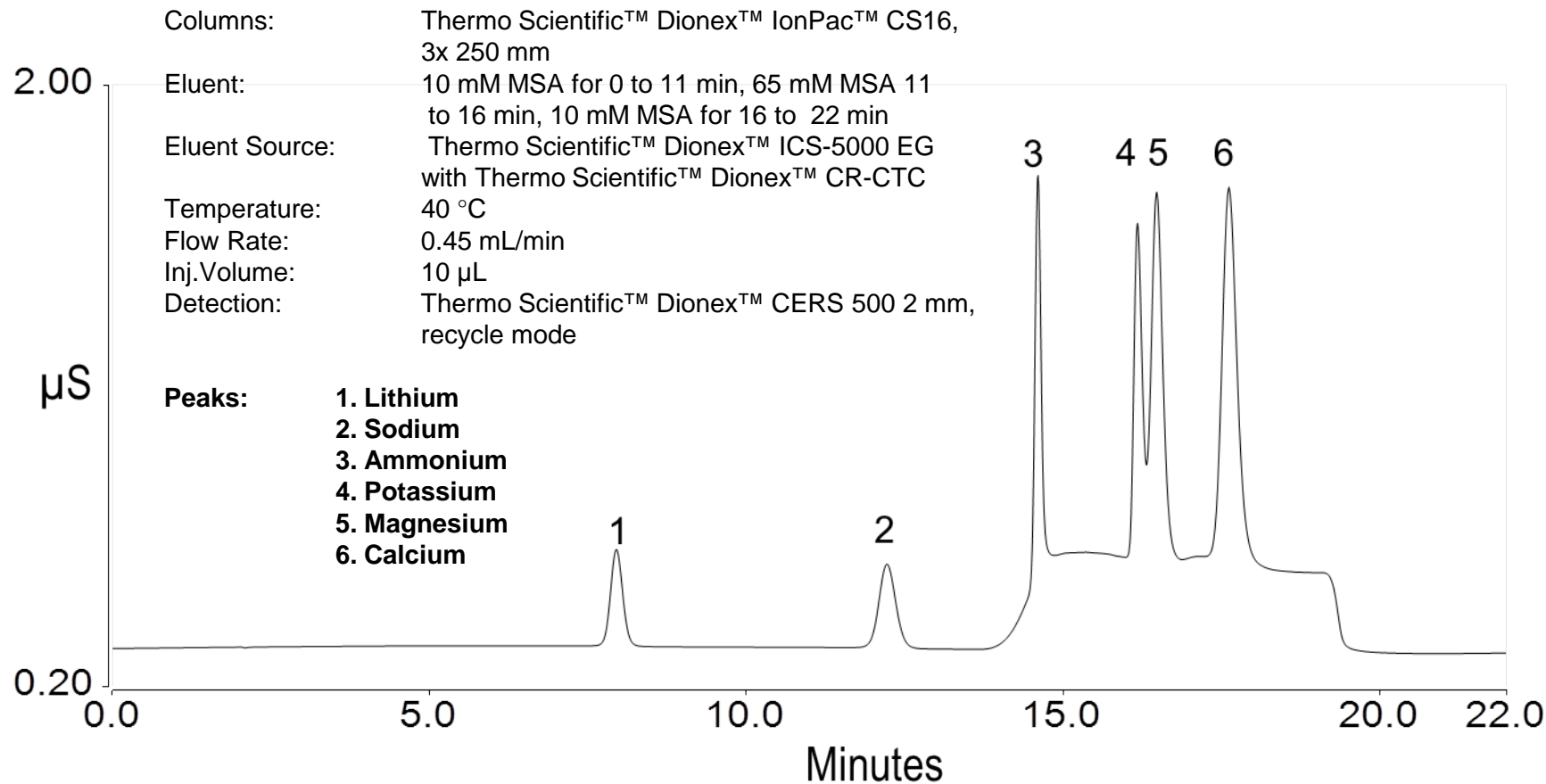


Example #2

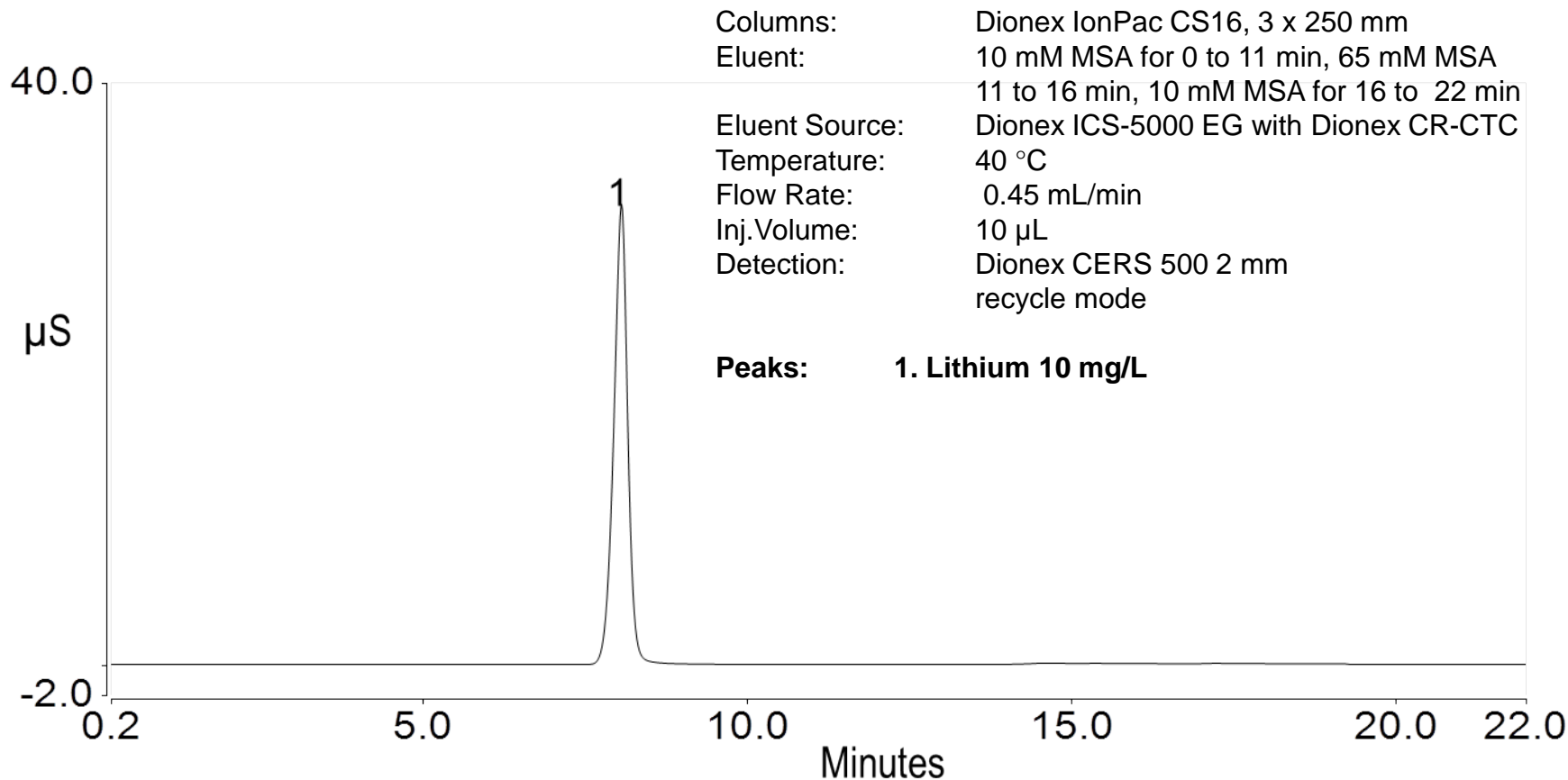
Cation IC Used for an Assay and Limit Test
of a Drug Substance

- The method to assay lithium was developed in our lab for a proposal to modernize the USP lithium hydroxide monograph
- The same method was developed to allow the measurement of calcium that is also required in the LiOH monograph
- Our work has been reported in Application Note 1144

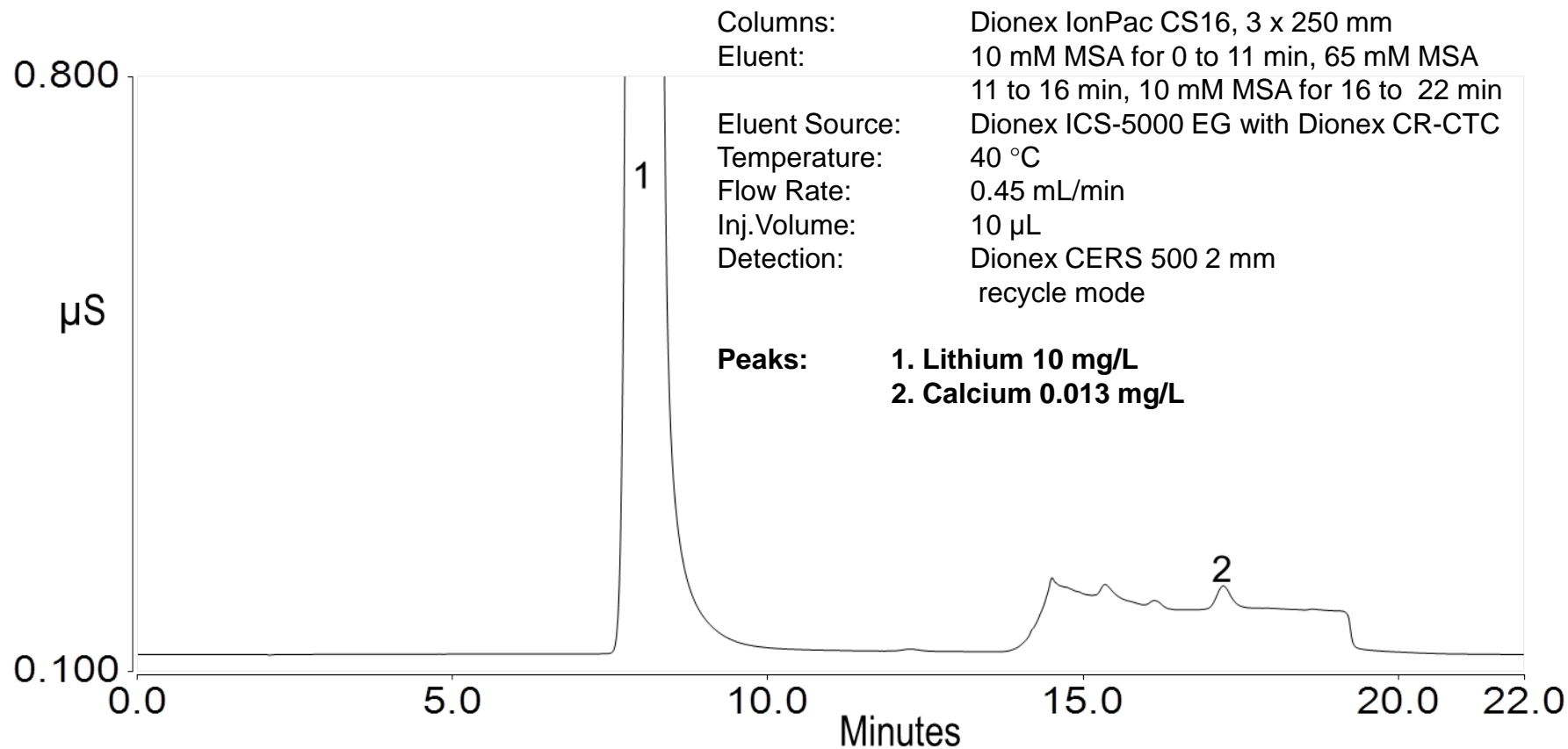
Separation of Six Common Cations



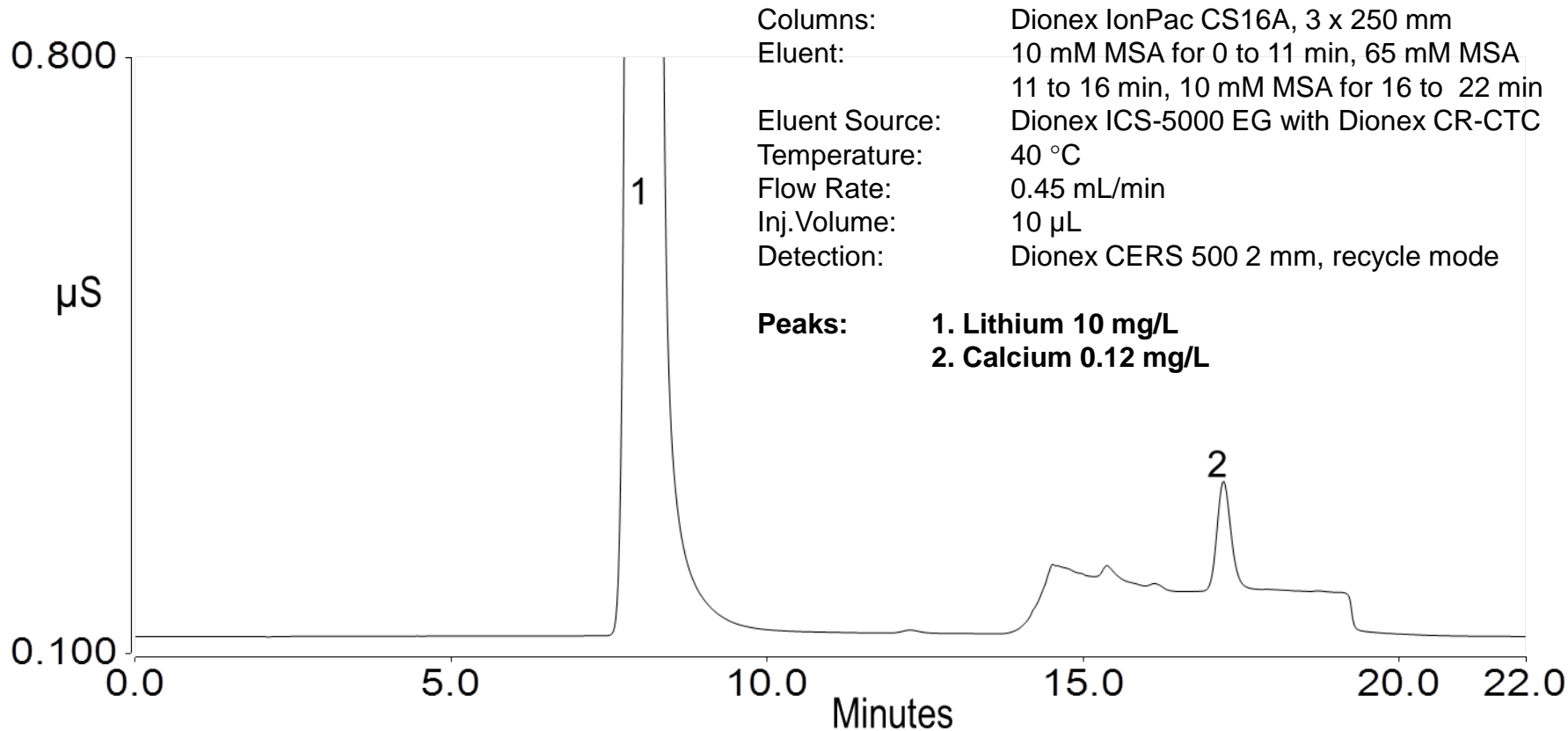
Determination of 10 mg/L Lithium in 10 mM Acetic Acid



Enlarged to View the Calcium Peak



Lithium Sample Spiked with Calcium at the USP Limit



- Question 1: Has someone already done this or similar analysis?
 - Web/Literature search
 - Search AppsLab <https://appslab.thermofisher.com/>
- If yes, does it apply to my sample and can it be improved?
 - Check the column used and see if Thermo Fisher Scientific has a better column or a different format
- If no, Question 2: Is my analyte an anion, cation, carbohydrate, or other compound amenable to IC?

- Question 3: Are there compounds of the same charge in the sample?
- If yes, will need a column to resolve those compounds
- If yes, and the concentration of the other compounds are high relative to the analyte of interest, then a high-capacity column will be needed.
- Question 4: How can my analyte be detected & what sensitivity do I need?
 - Detection choice impacts the column and mobile phase chosen

Example

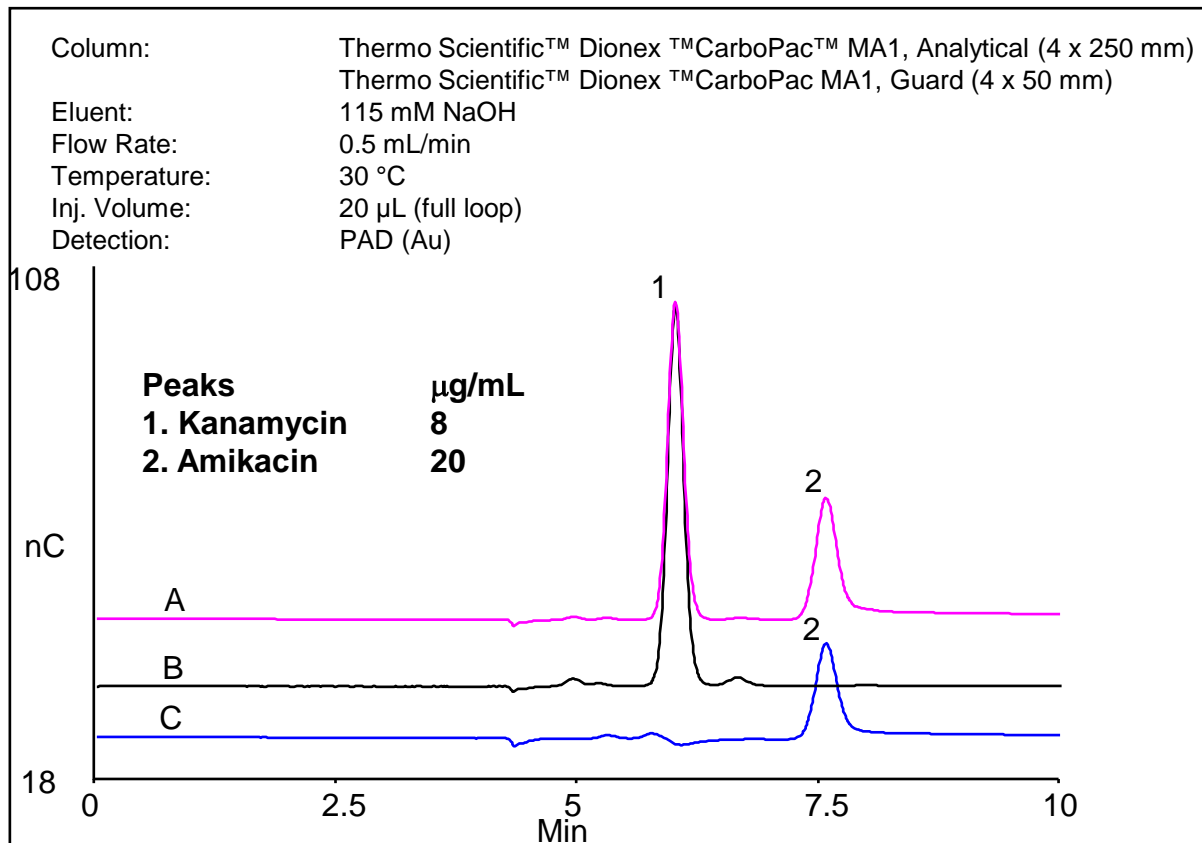
Designing an Assay for Kanamycin Sulfate and Amikacin

- Question 1: Has someone already done this or similar analysis?
 - Yes – a USP-NF method – But let's pretend not
- Question 2: Is my analyte an anion, cation, carbohydrate, or other compound amenable to IC?
 - Carbohydrate and some aminoglycosides have been analyzed by HPAE-PAD
- Question 3: Are there compounds of the same charge in the sample?
 - Yes – there are 2 analytes, but one low and the other high

- Question 4: How can my analyte be detected & what sensitivity do I need?
 - Detection by pulsed amperometry (PAD)
 - No requirement for the amount of the compound at lower concentration

- The solution:
 - As a carbohydrate – HPAE-PAD
 - Because they are even weaker anions than typical carbohydrates – a high pH and therefore high capacity column is required – CarboPac MA1

Separation of Amikacin and Kanamycin



- This method used for assay for both compounds in drug substance and drug products – 7 monographs

- IC is finding greater application in pharmaceutical laboratories to develop methods for drug products and drug substances
- IC methods have a greater degree of automation compared to other chromatographic techniques
- IC is one of the techniques being used to modernize pharmacopeia methods

- Jingli Hu – Sodium Nitrite
- Sachin Patil – Lithium Hydroxide
- Lipika Basumallick – Kanamycin and Amikacin

**Applications of
ION CHROMATOGRAPHY
for PHARMACEUTICAL
and BIOLOGICAL
PRODUCTS**

Edited by
Lokesh Bhattacharyya
Jeffrey S. Rohrer

 **WILEY**


ftp://
SITE AVAILABLE

Thank you for your attention!