Winning The Last Battle Against Edman Degradation: Reliable Leucine/Iso-leucine Differentiation In Peptide Sequencing Using an Orbitrap Fusion Mass Spectrometer

Lebedev A,¹ Damoc E,² Makarov A,² Samguina T¹ ¹Moscow State University, Moscow, Russia; ²ThermoFisher Scientific, Bremen, Germany





Introduction

The problem of differentiation between isomeric leucine and isoleucine has been the most relevant for *de novo* sequencing of peptides by means of mass spectrometry. The most successful approaches involve secondary fragmentation of the odd electron z-ions with formation of *w* ions in MS2 and MS3 experiments (SCHEME 1).



Discrimination of isomeric residues deals with the characteristic losses from the side chains: -43 Da (Leu) and -29 Da (IIe). R. Zubarev et al. used ECD¹ and HECD¹ (ē energy up to 11 eV) for this purpose in MS2 experiments. Unfortunately, further fragmentation of z ions does not have a general character and the corresponding w ions are often not present in the spectra. McLuckey et al.² and Balaram et al.³ carried out MS3 experiments when ETD formation of z ions was followed by CID. The efficiency of the targeted fragmentation increased, however there were too many fragment ions in the spectra while the process of radical site migration⁴ along the backbone often resulted in uncertainties when both Leu and IIe residues were present in the original z-ion. Here we report new results on reliable and straightforward discrimination of Leu/IIe in natural peptides using the power of the Thermo Scientific[™] Orbitrap Fusion[™] Tribrid[™] mass spectrometer (ETD/HCD).

Methods

Sample Preparation

Six natural peptides were isolated from the skin secretion of Russian frog *Rana ridibunda*. Their length was in the range between 15 and 37 amino acids in the backbone, the number of basic residues (lysine and arginine) – between 3 and 6, and the number of targeted isomeric (leucine and isoleucine) residues – between 1 and 7 (24 cases altogether).

Natural	peptides	used in	the ex	periments:
---------	----------	---------	--------	------------

Name	Sequence	Mass, Da	Comment
Ranatuerin 2R	AVNIPFKVKFR <u>CKAAFC</u>	1939.0325	Confirmed structure
Brevinin 1Ra	VIPFVASVAAEMMQHVY <u>CA</u> <u>ASRRC</u>	2636.2485	Confirmed structure
Brevinin 1E	FLPLLAGLAANFLPKIF <u>CKIT</u> <u>RKC</u>	2674.5220	Confirmed structure
Brevinin 2Ec	GILLDKLKNFAKTAGKGVLQ SLLNTAS <u>CKLSC</u>	3516.9161	Confirmed structure
Ranatuerin2a	KXXXNPKFR <u>CKAAFC</u>	1748.9583	Not confirmed structure
Esculentin 2R	G <mark>XX</mark> SXVKGVAKXAGKTFAK EGGKFGXEFXA <u>KVTNQC</u>	3823.0893	Not confirmed structure

Mass Spectrometry

Experiments were carried out with an Orbitrap Fusion mass spectrometer using the standard Thermo ScientificTM TuneTM 1.0 instrument software and a Thermo ScientificTM EASY-Max NGTM ion source in infusion mode. Selection of different peptide ions for MS2(ETD) fragmentation as well as *z*-ions selection for MS3(HCD) fragmentation was performed manually in Tune page.

SCHEME 2. The sequence of mass spectrometry stages



Sequencing in Orbitrap Fusion MS

The principal features of the experiment

1. Precise extraction of the targeted z ions (with N-terminal Leu/IIe) is quite easy due to the rich choice of the required z ions with the charges from +2 to +7. 2. HCD collision energy is varied in the range 10 - 40 NCE.

Outcome: Extremely selective fragmentation of the targeted z ions is achieved. MS3 spectra as a rule contain exclusively precursor z ions and w - product ions.

SCHEME 3. Possible MS3 losses from side chain of Leu/Ile in the method.



Results

Single lle in z ions

Figures 1-3 exemplify identification of Leu/Ile for solitary residues within a peptide.

FIGURE 1. MS3 spectrum of z_{14}^{+2} ion of ranatuerin 2R



The difference of 14.519 Da between the precursor and the main product ion corresponds to the loss of C_2H_5 and identifies ⁴IIe. There is no trace of an alternative for Leu loss of isopropyl radical (43.0546 Da). Formation of w ions in the case of IIe involves the loss of the ethyl group (Scheme 3). However alternative elimination of the methyl group is also possible. The intensity of this ion is usually low according to the maximal alkyl loss rule. Nevertheless, due to very pronounced fragmentation in ETD/HCD mode the loss of methyl becomes visible as well.





 CH_{3} , C_2H_5 and C_4H_8 losses from the triply charged z_{23} ion generated by ETD from $[M+4H]^{4+}$ of brevinin 1Ra confirm the presence of ²IIe. In all other cases with one Leu/IIe residue in the backbone the results were similar.

The presence of another Ile/Leu residue in the remote position from the N-terminus of $z^{\cdot}\ \text{ion}$

FIGURE 3. MS3 spectrum of z_{12}^{+2} ion of brevinin 1E



The difference 21.5273 Da between the precursor and the main product ion corresponds to the loss of C_3H_7 and identifies reliably ¹³Leu. This fact demonstrates the extraordinary selectivity of the proposed method. There are no even traces of the ethyl group loss, due to the presence of ¹⁶IIe residue. Radical migration is less pronounced in ETD/HCD mode: no other product ions are formed.

Neighboring ²lle-³Leu-⁴Leu in brevinin 2Ec

FIGURE 4. MS3 spectrum of z₃₃+3 ion of brevinin 2Ec at NCE15



Due to possible radical site migration the losses of both Et and iPr radicals were expected in ETD/HCD experiment with z_{33}^{+3} ion. The application of the minimal collision energy (NCE 10) resulted in the selective fragmentation of the side chain of N-terminal ²IIe. There were no even traces of the alternative iPr losses at NCE 15 (Figure 4).

FIGURE 5. MS3 spectrum of z_{33}^{+3} ion of brevinin 2Ec at NCE 25



At NCE 25 (Figure 5) the signal of m/z 1135 representing the loss of C_3H_7 arises due to the neighboring ³Leu and ⁴Leu. Therefore, radical site migration is also possible in the proposed method. However, the process does not interfere with the correct identification of Leu/IIe pairs even in neighboring positions.

FIGURE 6. MS3 confirmation of ⁴Leu in Ranatuerin 2Ra



⁴Leu may be unequivocally identified using MS3 spectrum of z_{12}^{+2} ion.



FIGURE 7. MS3 confirmation of ³lle in Ranatuerin 2Ra



³IIe was easily and reliably identified using MS3 spectrum of z_{13}^{+2} , with ion registered at NCE 15. All the observed losses are due to IIe side chain (Figure 7). With the increase of NCE to 20 radical site migration becomes visible. MS3 spectrum of z_{13}^{+2} ion (Figure 8) represents the loss of C_3H_7 from the neighboring ⁴Leu.



The loss of Et is obvious in MS3 spectrum of z_{14}^{+2} at NCE 10 (Figure 8a). With the increase of NCE to 20 radical site migration becomes quite pronounced. MS3 spectrum (Figure 8b) besides Et elimination represents an alternative loss of C_3H_7 due to ⁴Leu. Nevertheless MS3 spectra of the targeted z ions at small NCE allowed for reliable identification of all three linked isomeric residues in Ranatuerin 2Ra: KIILNPKFRCKAAFC.

Conclusion

- ETD/HCD approach with the Orbitrap Fusion MS is the key to the successful identification of Leu/Ile in non-tryptic natural peptides with chain length up to 37 residues.
- The resulting MS3 spectra are very selective with 1-4 secondary fragment ions, with targeted w ions usually being the most abundant.
- The effects of radical site migration are significantly less pronounced than in the case of the ion trap techniques reported earlier.
- The proposed approach may be used in high throughput proteomics experiments and allows creating automated methods for the complete *de novo* sequencing exclusively by means of mass spectrometry.

References

- 1. Kjeldsen, F. et al. Anal. Chem. 2003.
- 2. Hongling, H. et al. J. Proteome Res., 2007.
- 3. Gupta, K. et al. J Proteome Res., 2012.
- 4. Leymarie, N. et al. J Am Chem Soc., 2003.

www.thermoscientific.com

©2014 Thermo Fisher Scientific Inc. All rights reserved. ISO is a trademark of the International Standards Organization. All other trademarks are the property of Thermo Fisher Scientific and its subsidiaries. This information is presented as an example of the capabilities of Thermo Fisher Scientific products. It is not intended to encourage use of these products in any manners that might infringe the intellectual property rights of others. Specifications, terms and pricing are subject to change. Not all products are available in all countries. Please consult your local sales representative for details.

Africa +43 1 333 50 34 0 Australia +61 3 9757 4300 Austria +43 810 282 206 **Belgium** +32 53 73 42 41 **Canada** +1 800 530 8447 China 800 810 5118 (free call domestic) 400 650 5118

Denmark +45 70 23 62 60 Europe-Other +43 1 333 50 34 0 Finland +358 9 3291 0200 France +33 1 60 92 48 00 Germany +49 6103 408 1014 India +91 22 6742 9494 Italy +39 02 950 591

Japan +81 45 453 9100 Latin America +1 561 688 8700 Middle East +43 1 333 50 34 0 Netherlands +31 76 579 55 55 New Zealand +64 9 980 6700 **Norway** +46 8 556 468 00 **Russia/CIS** +43 1 333 50 34 0



Singapore +65 6289 1190 Spain +34 914 845 965 **Sweden** +46 8 556 468 00 **Switzerland** +41 61 716 77 00 **UK** +44 1442 233555 **USA** +1 800 532 4752

hermo SCIENTIFIC A Thermo Fisher Scientific Brand

