

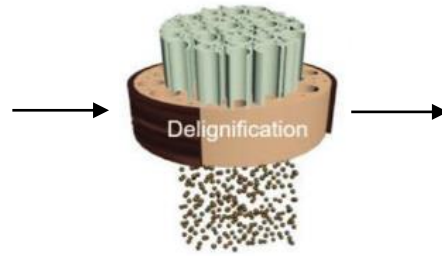
Off-line LC×SFC-HRMS/MS method for the non-target analysis of depolymerized lignin

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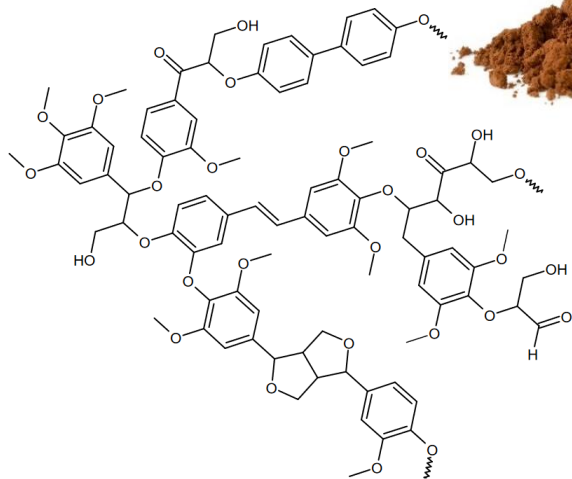
Background – what is lignin?



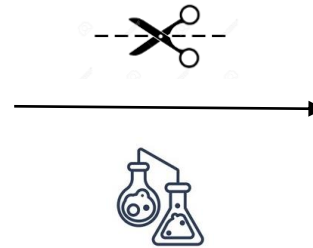
Cellulose



Isolation of chemicals



Lignin



Biofuels



Novel biomaterials



Lignin ...

- is a **by-product** of the paper industry; only 2% is commercialized
- should be **valorized more for circular economy**
- needs thorough **structural characterization** for efficient valorization



Depolymerized lignin is a **complex mixture**, consisting of:

- numerous compounds
- compounds with similar structures, including isomers
- mostly neutral compounds

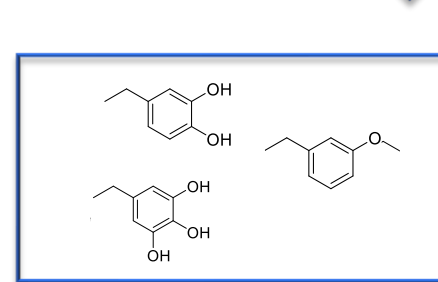
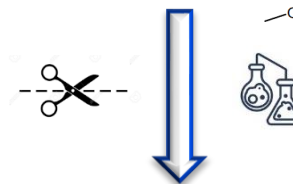
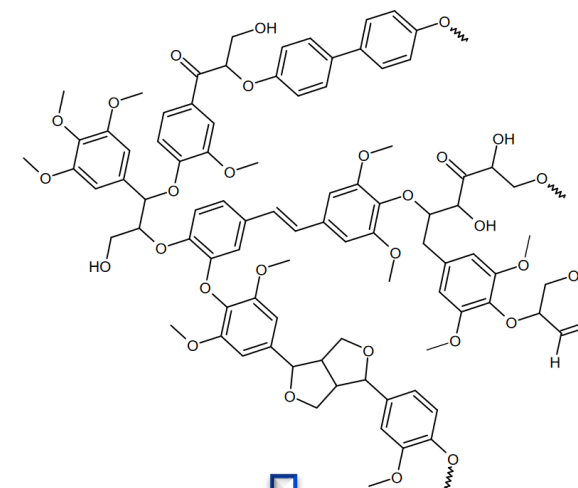
Possible solution: Two-dimensional RPLC×SFC

SFC (supercritical fluid chromatography) – can be used under normal phase conditions with supercritical fluid (here $s\text{CO}_2$) as the less polar eluent.

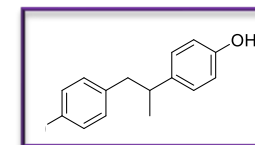
Lack of commercial standards

- monomers – quite a lot of commercial standards
- dimers and other oligomers – only a few dimers are commercially available

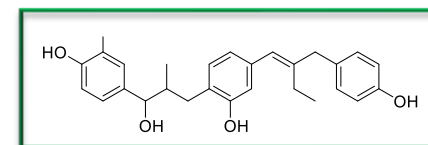
Possible solution: non-target analysis with HRMS/MS detection



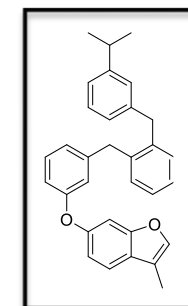
Monomers



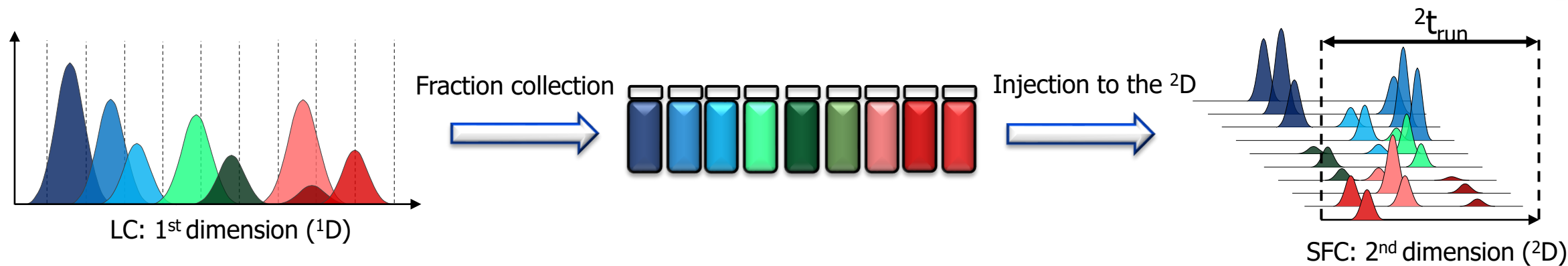
Dimers



Trimers



Tetramers



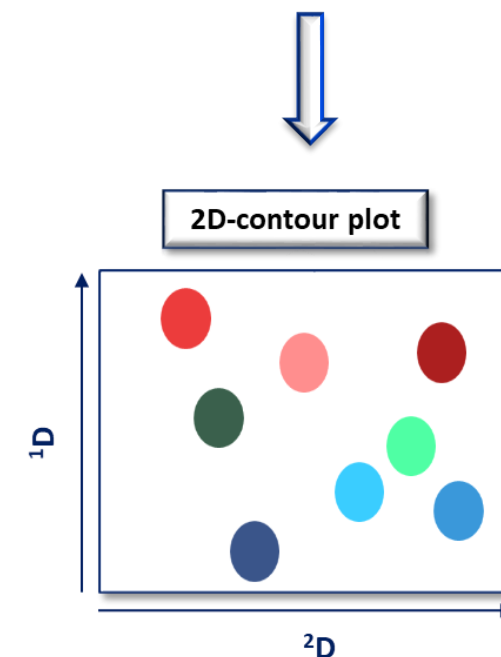
Off-line 2D method compared to on-line 2D:

Advantages

- Possibility to dissolve the fractions in a solvent suitable for 2^D
- Easier to obtain good quality MS^2 data
- No need for „complex“ on-line LC×SFC system
 - *O-19 Margaux Sanchez*

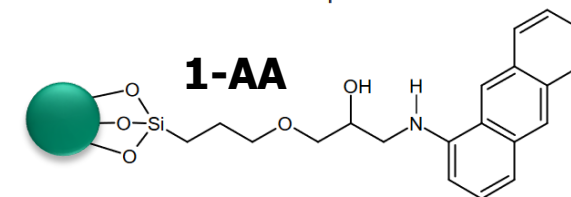
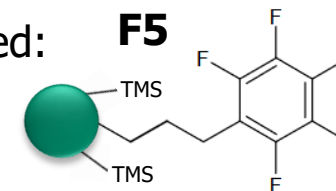
Disadvantages

- Long analysis time
- Higher risk of contamination



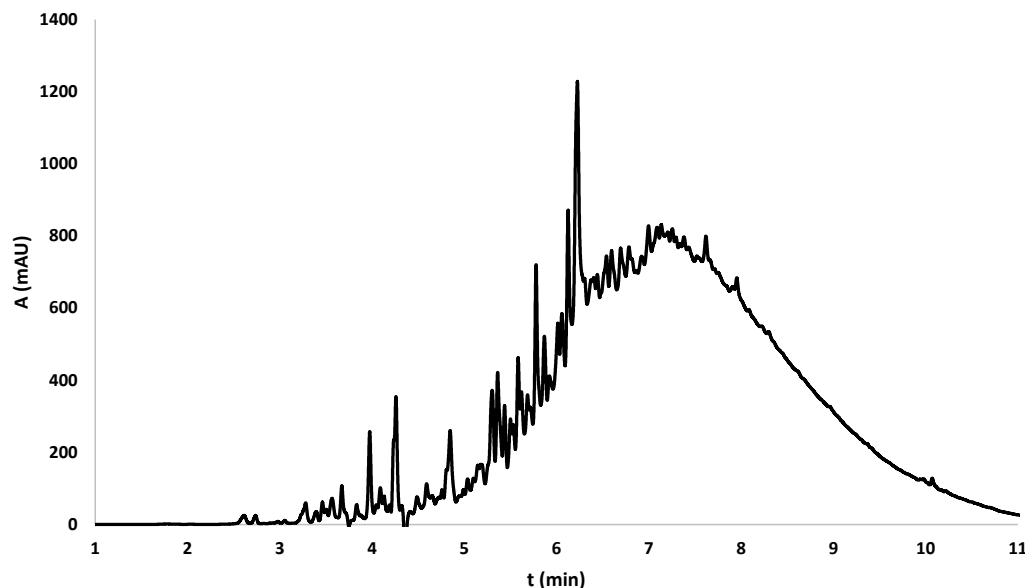
Various columns were screened using **33 monomer standards**. The following columns were selected:

- RPLC: **Kinetex F5** (pentafluorophenylpropyl)
 - *Hydrophobic, dipole-dipole, aromatic, steric, and hydrogen bonding interactions*
- SFC: **Torus 1-AA** (1-aminoanthracene)
 - *Aromatic, steric, and hydrogen bonding interactions*



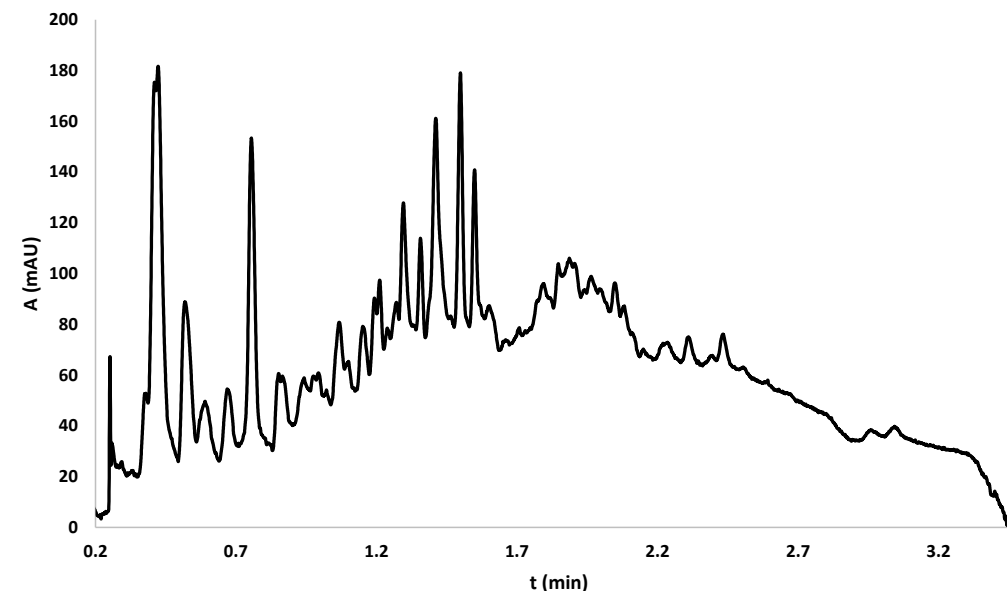
The two dimensions were separately optimized using the **depolymerized lignin** sample:

LC-UV with Kinetex F5



Kinetex F5, 150 mm x 4.6 mm, 5 μ m; depolymerized lignin 30 mg/mL in MeOH; A: H₂O + 0.1% FA; B: ACN + 0.1% FA, T = 30 °C, gradient: 1-90 % B in 9 min, F = 2.1 mL/min, V_{inj} = 10 μ L, UV (254 nm)

SFC-UV with Torus 1-AA

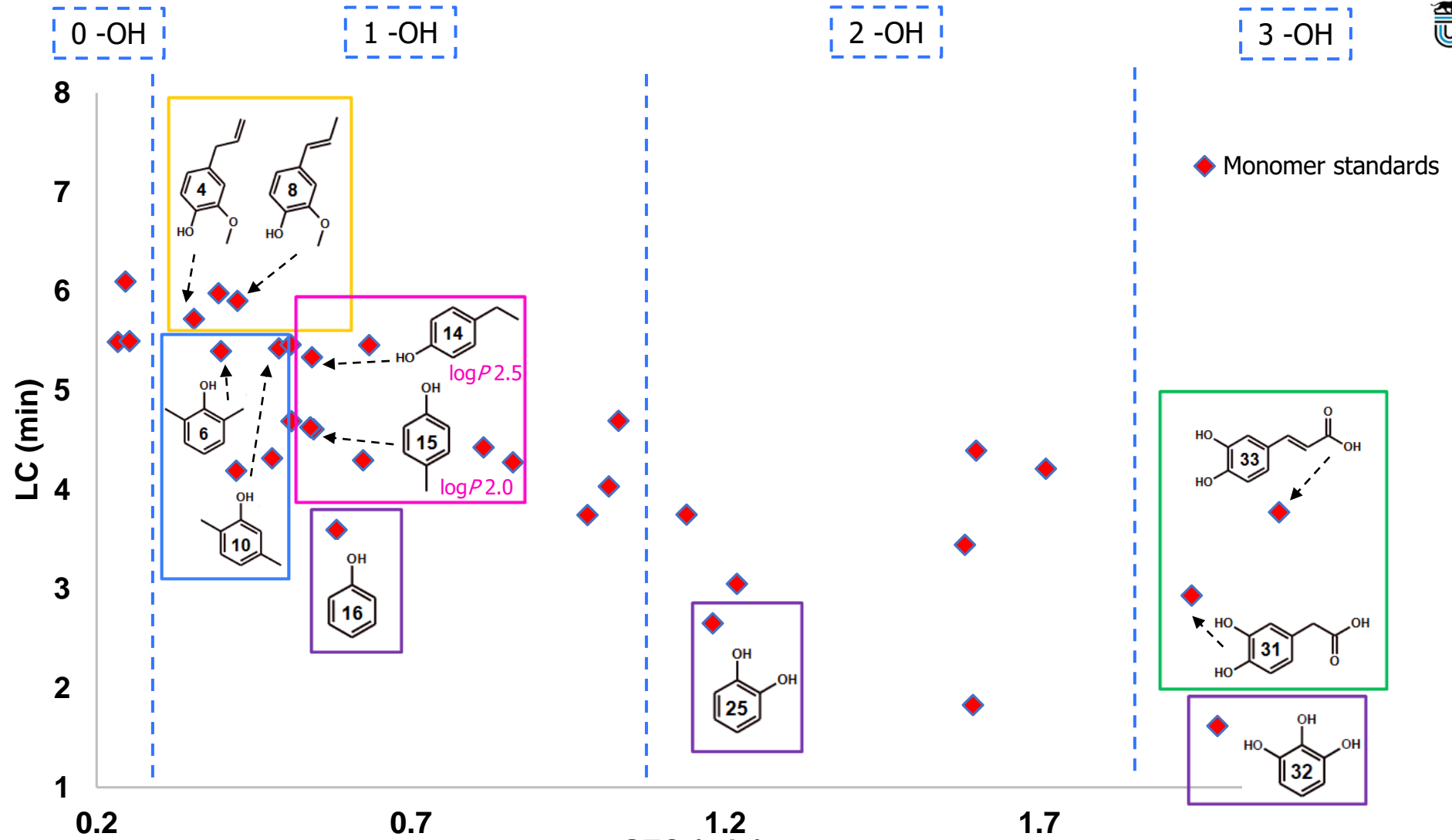


Torus 1-AA, 50 mm x 3 mm, 1.7 μ m; depolymerized lignin 5 mg/mL in MtBE, T = 40 °C, BPR = 140 bar, gradient: 5-50% MeOH in 2.95 min, F = 1.7 mL/min (= F_{max}), V_{inj} = 7 μ L, UV (210 nm)



LC×SFC separation mechanism – 33 monomer standards

RPLC: hydrophobicity



SFC: π - π interactions, conjugation

SFC: hydrogen bonding interaction, steric hindrance

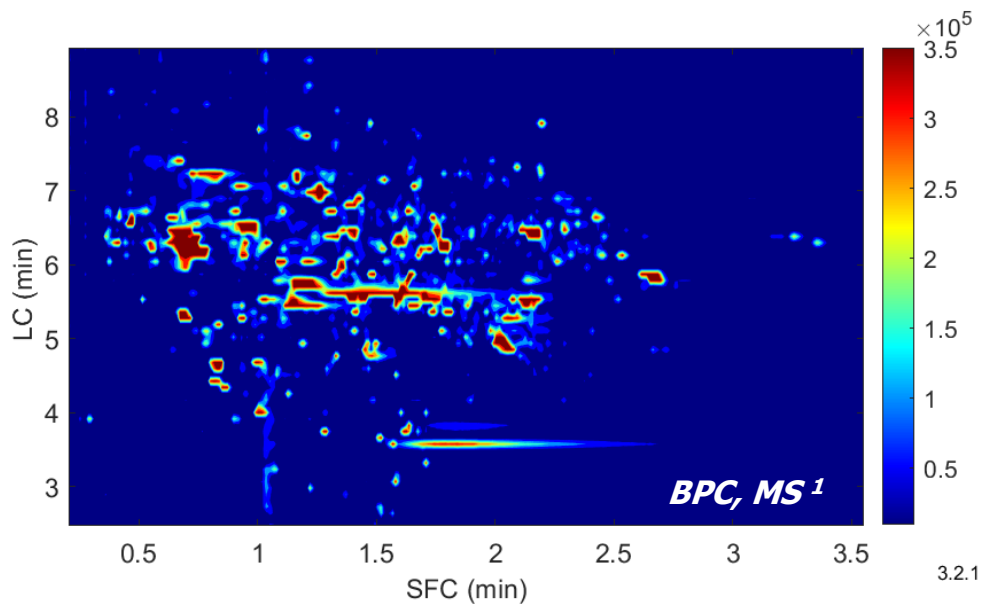
SFC: hydrogen bonding interaction, number of -OH groups

SFC: π - π interactions



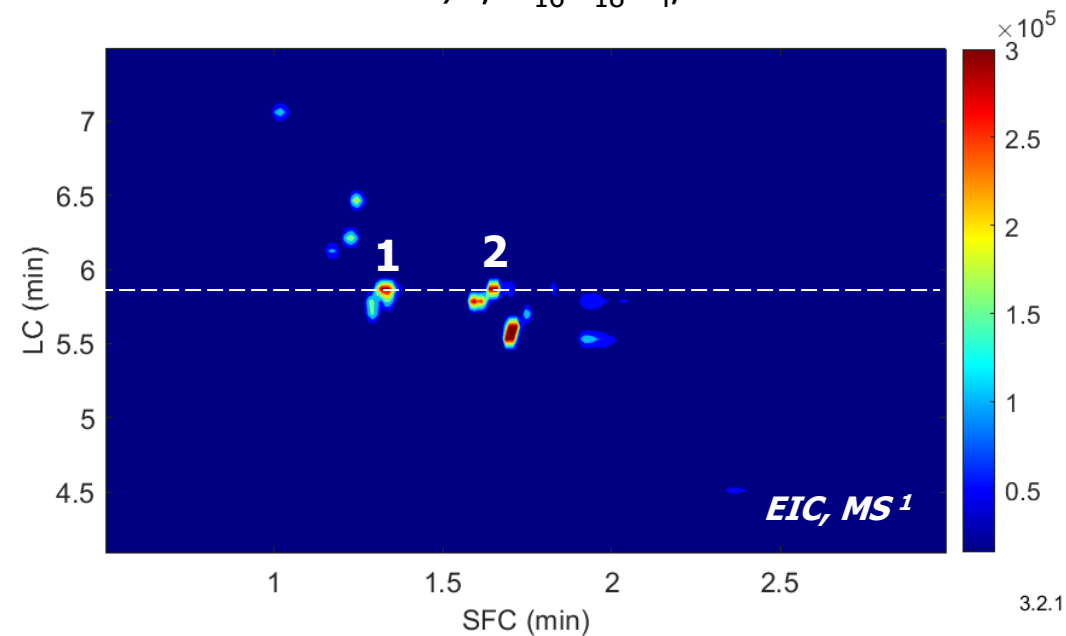
Corrected peak capacity: **1995**

Time of analysis: 430 min



Good separation of depolymerized lignin compounds

273.11279 m/z , $C_{16}H_{18}O_4$, DBE 8

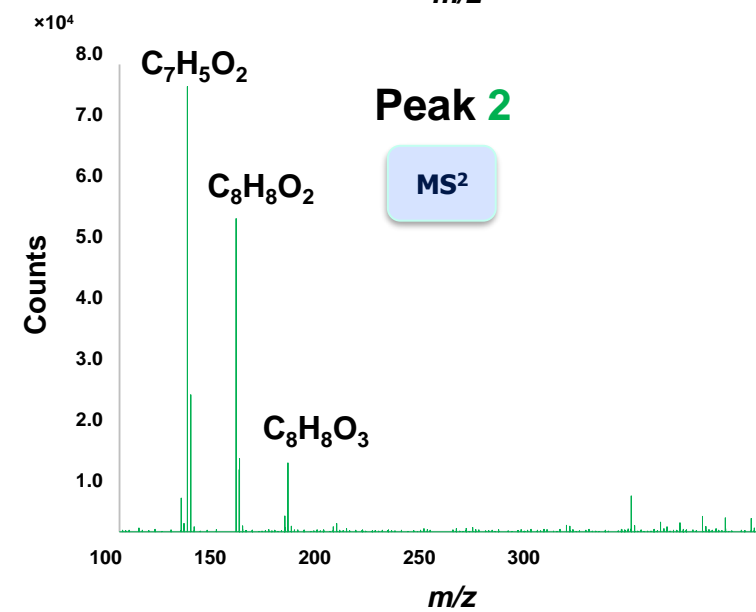
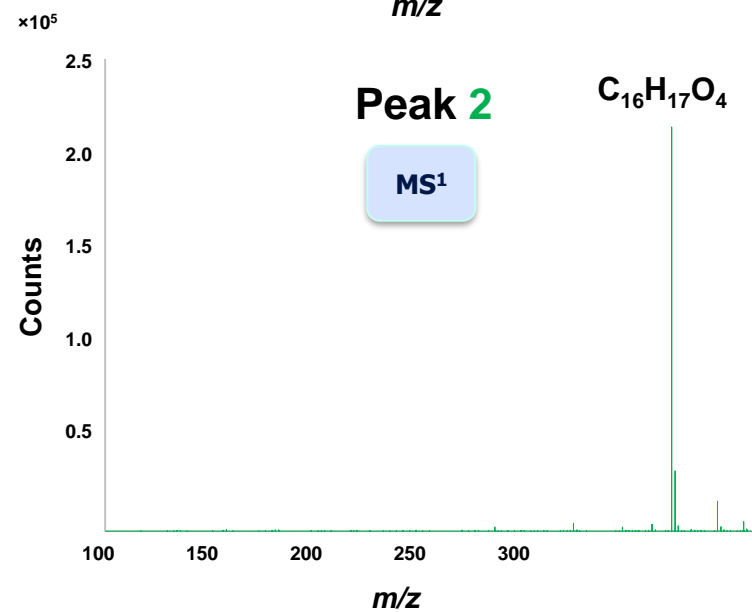
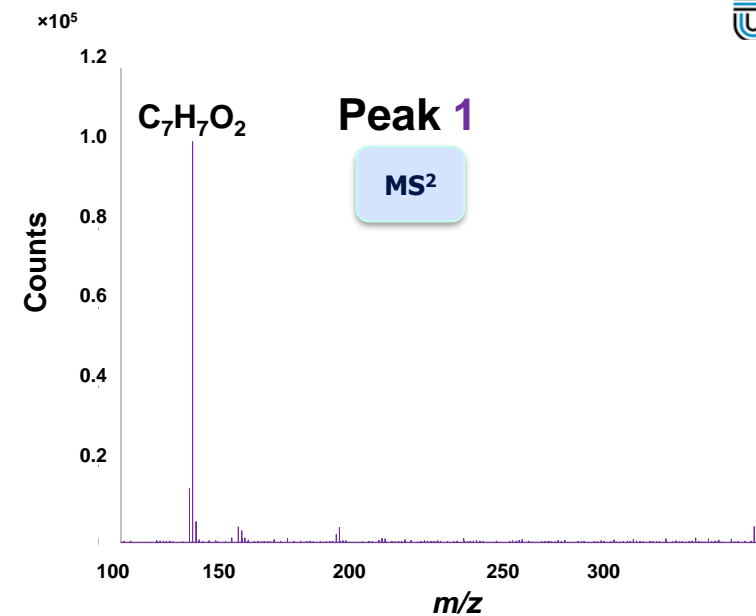
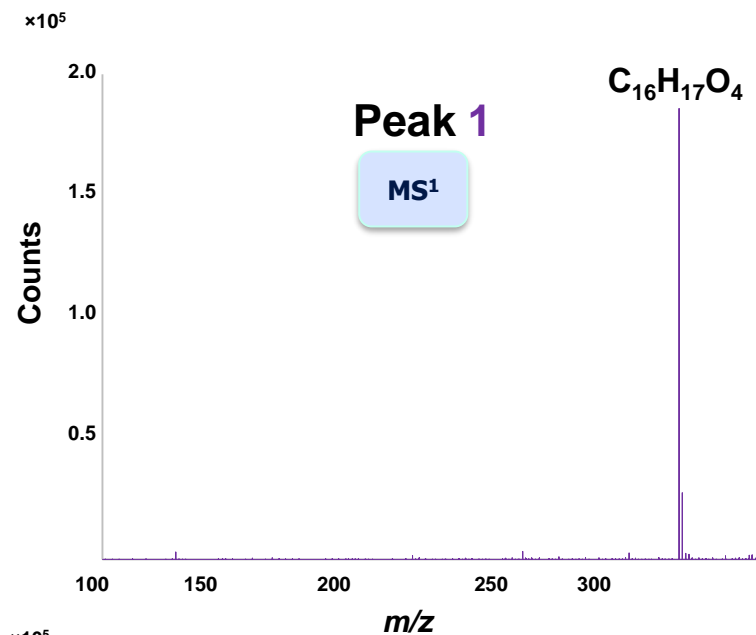
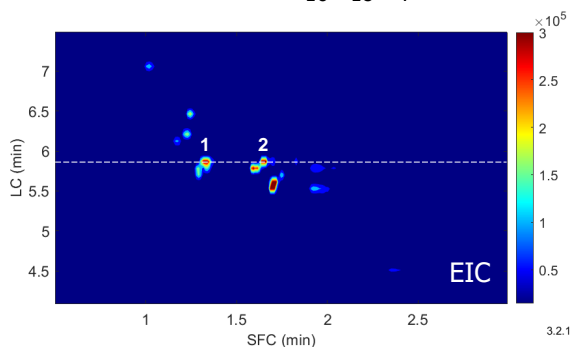


Good separation of isomers



Good quality MS¹ and MS² spectra

273.11279 m/z , C₁₆H₁₈O₄, DBE 8



✓ Clean MS¹ and MS² spectra

✓ Isomers of a dimer



Grouping of analytes based on their DBE values



- MS-DIAL software
 - peak picking
 - MS/MS deconvolution



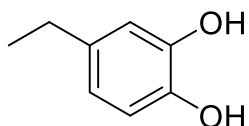
MS-DIAL

DBE: double bond equivalent

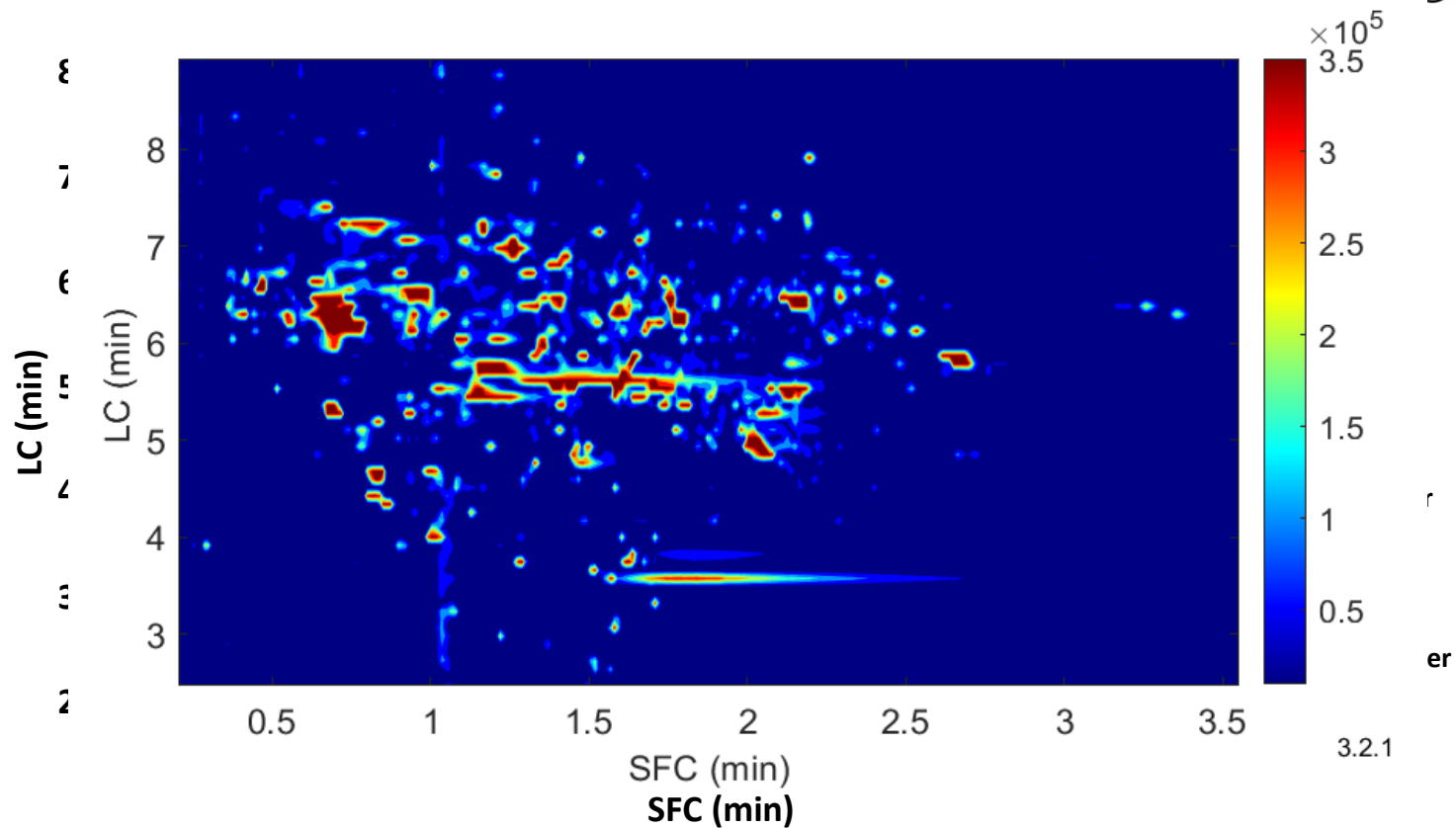
$$DBE = C + 1 - \frac{H}{2} - \frac{X}{2} + \frac{N}{2}$$

For $C_xH_yO_z$ compounds:

$$DBE = C + 1 - \frac{H}{2}$$

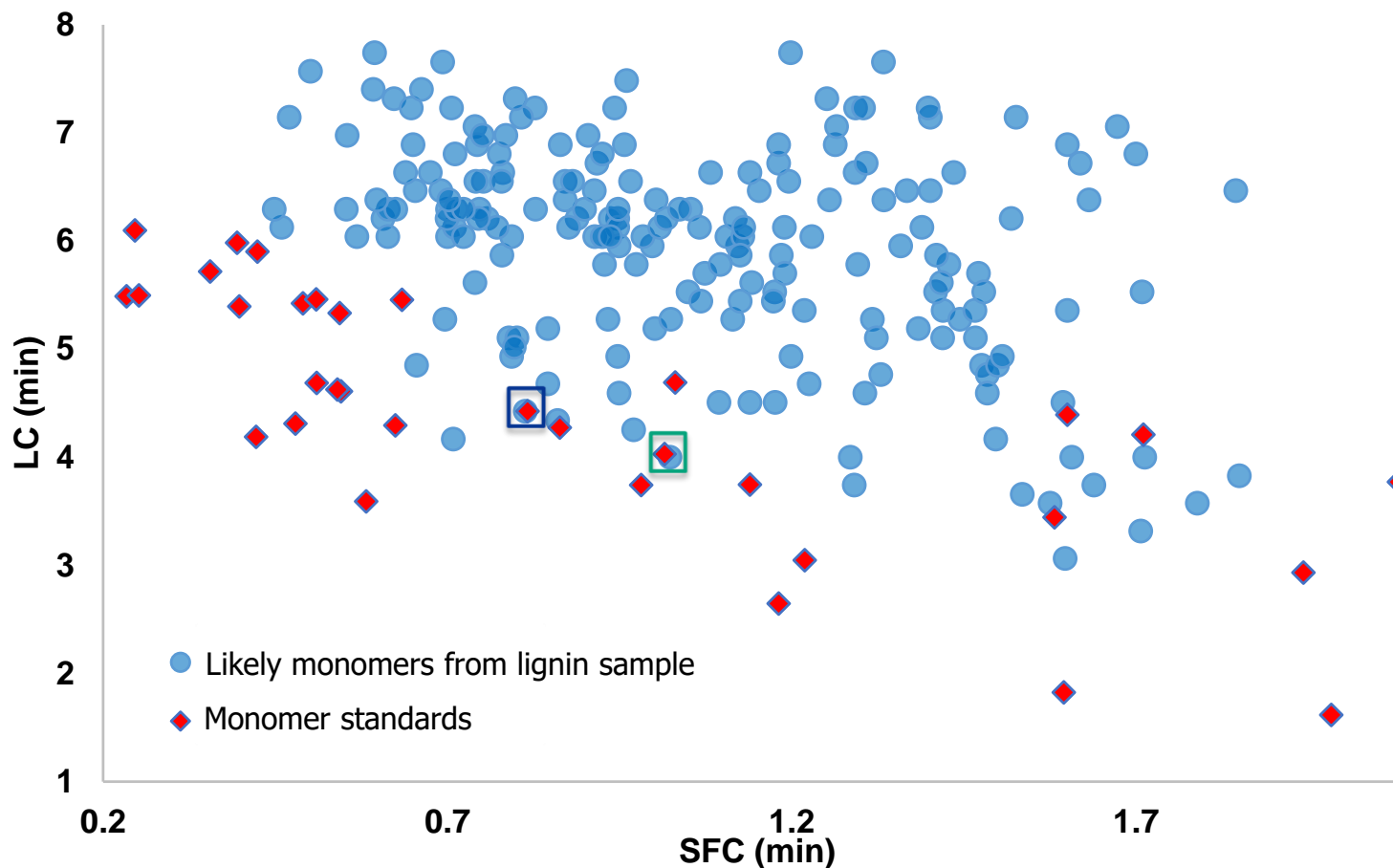


$$DBE = 8 + 1 - 10/2 = 4$$



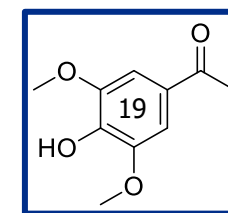
471 $C_xH_yO_z$ compounds with $DBE \geq 4$

$C_xH_yO_z$ compounds with DBE 4-7

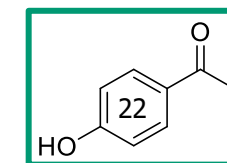


Identification based on 4 criteria:

- I. t_R in LC (± 0.085 min)
- II. t_R in SFC (± 0.012 min)
- III. Accurate mass (± 3 ppm)
- IV. MS/MS (match above 90%)



Acetosyringone



p-Hydroxyacetophenone

For the structural characterization of the other detected compounds MS/MS is needed



Molecular Network (MN)

Connects together features with similar MS² pattern

- The **more similar the MS²** (= structurally similar), the **more close the features** are on the Molecular Network

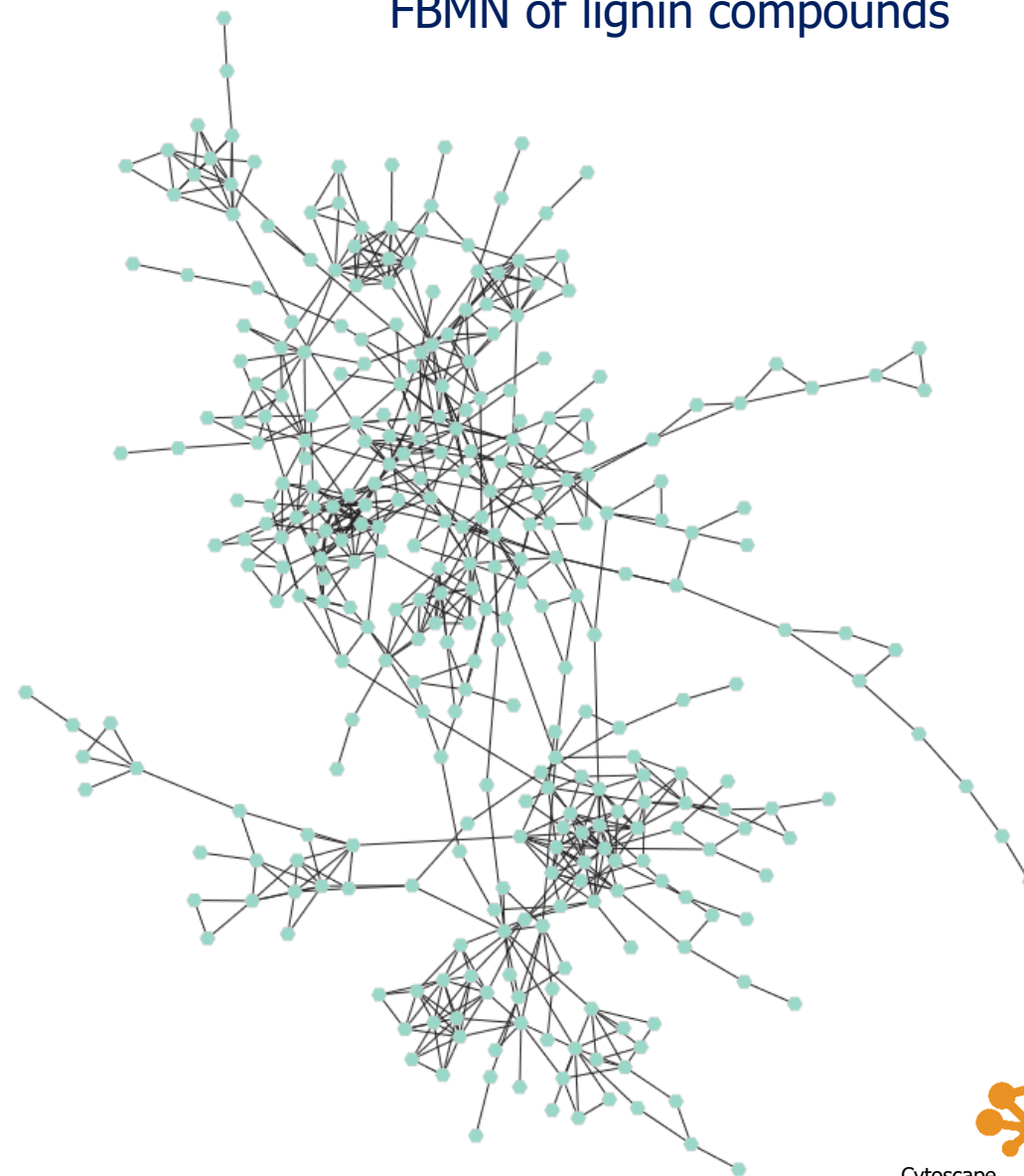
Feature-Based Molecular Network (FBMN)

MN that enables the visualization of features with:

- the **same monoisotopic mass**
- **different retention times**

Sample with a high number of isomers –
77% of compounds had at least one isomer

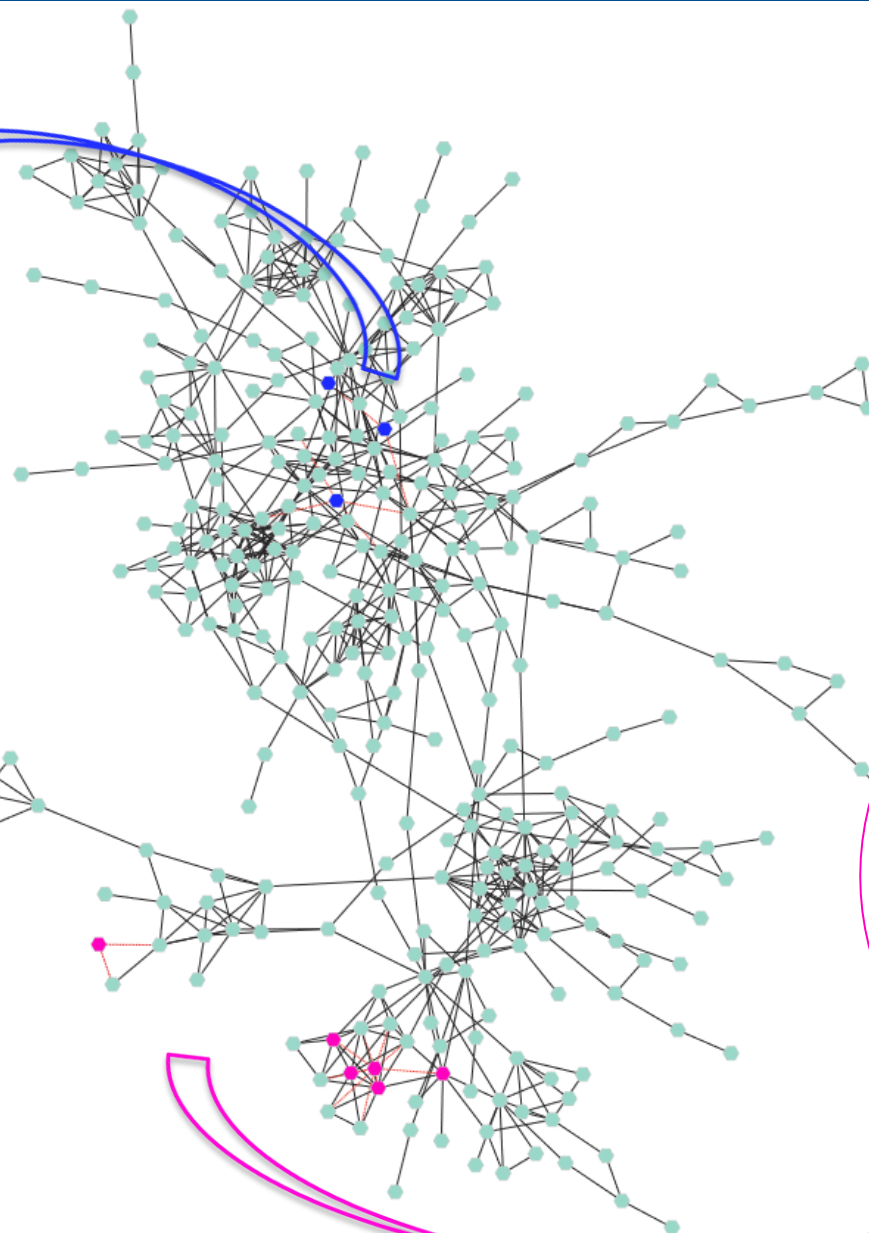
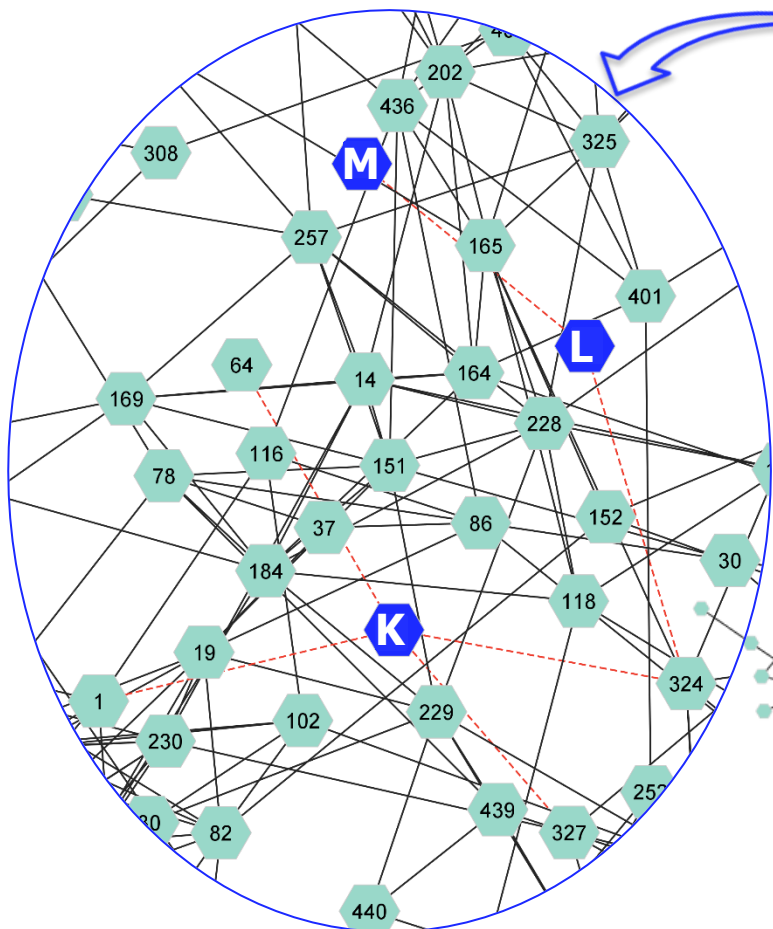
FBMN of lignin compounds



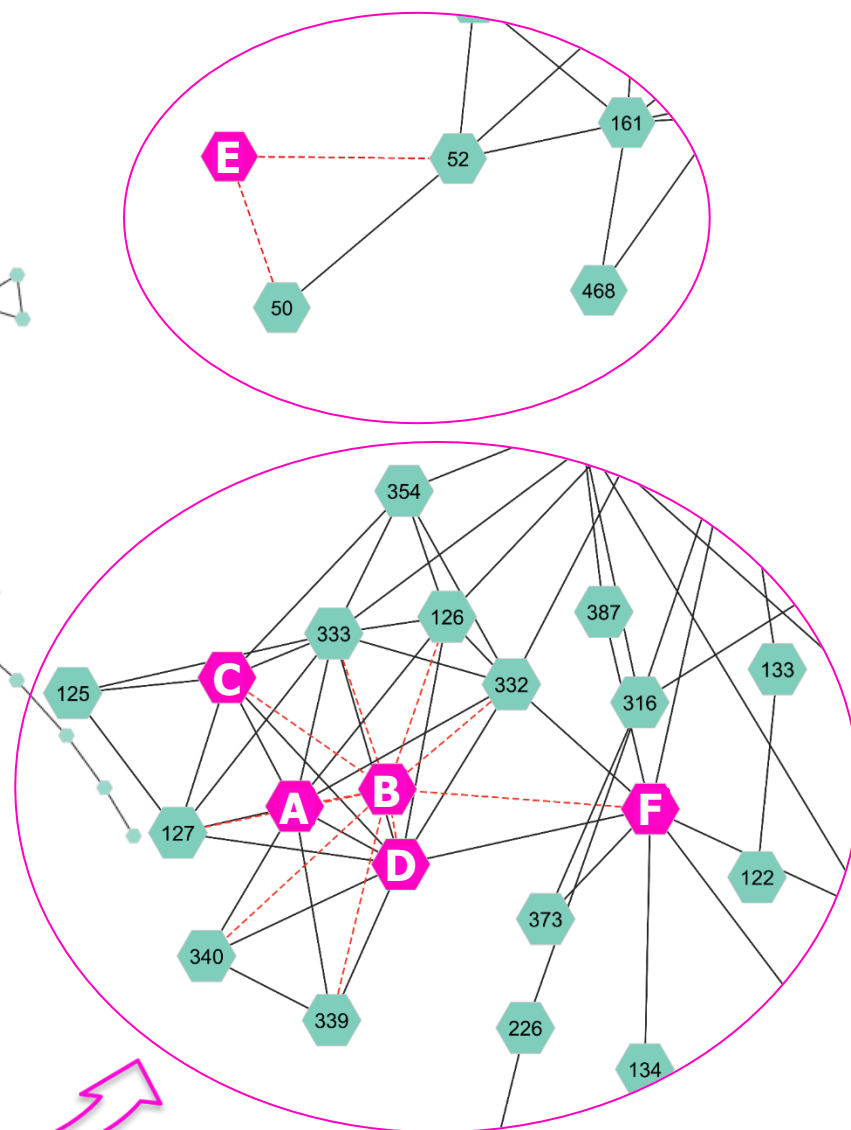


Feature-Based Molecular Network – FBMN

Isomers of $C_9H_{10}O_4$

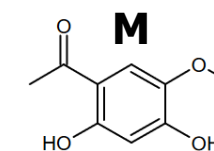
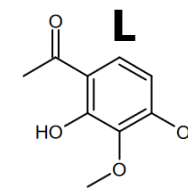
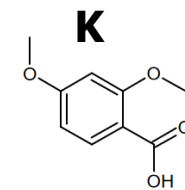
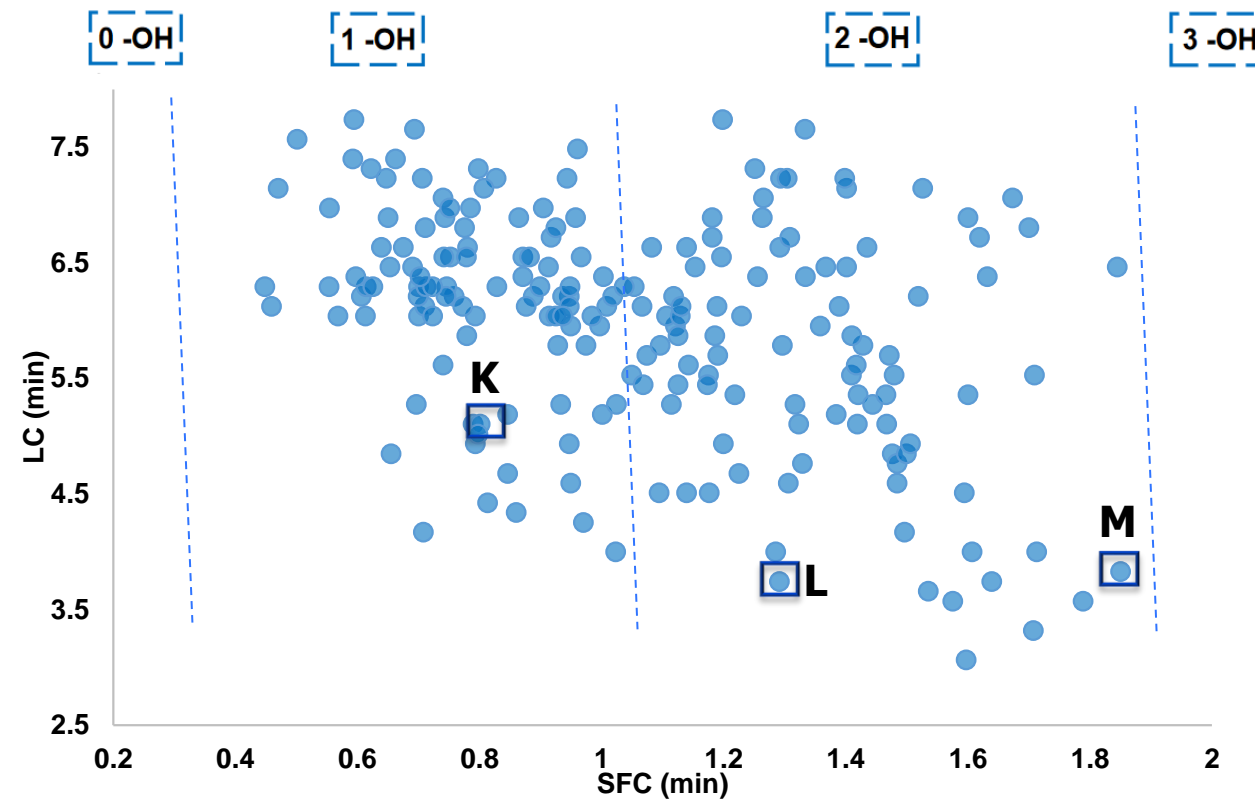
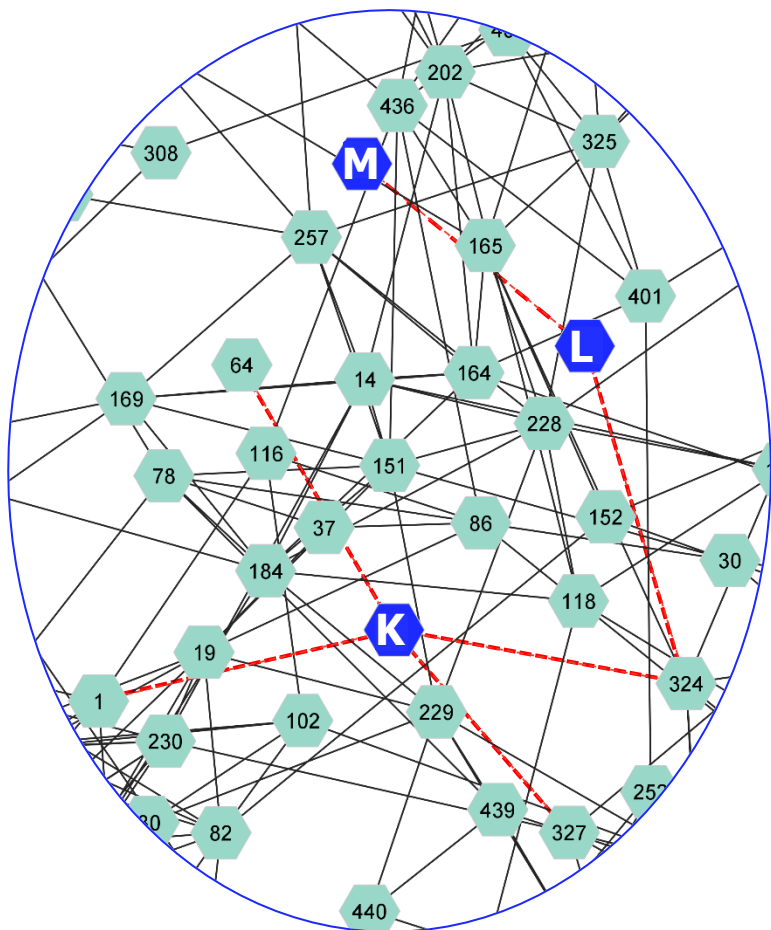


Isomers of $C_{19}H_{18}O_6$





Feature-Based Molecular Network – FBMN

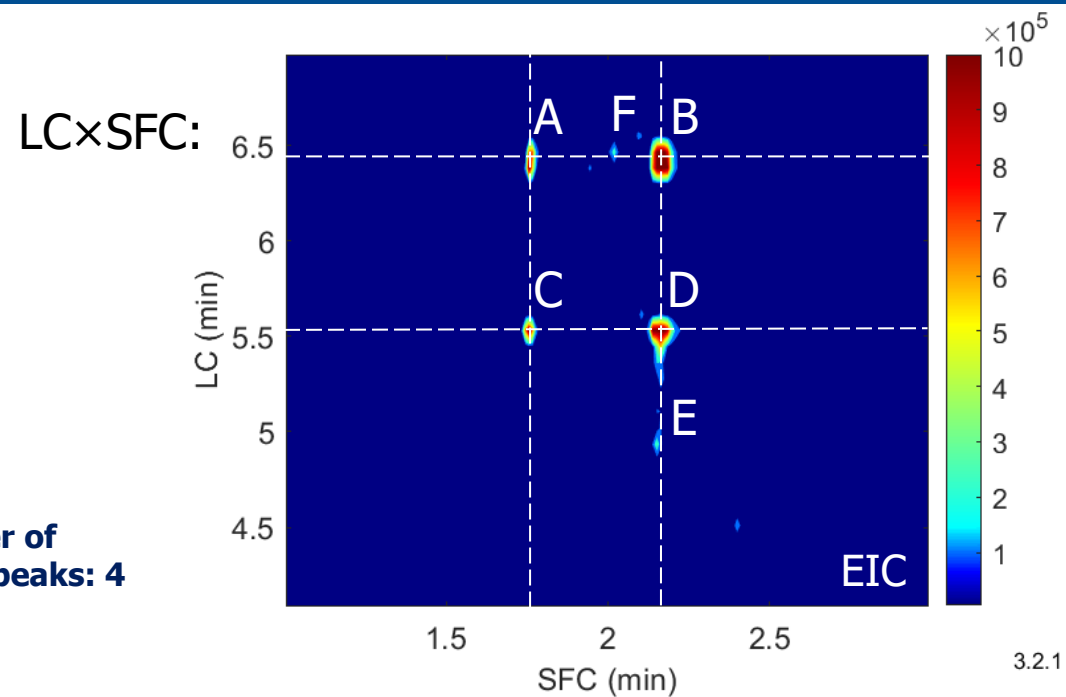


181.05009 m/z , $C_9H_{10}O_4$, DBE 5

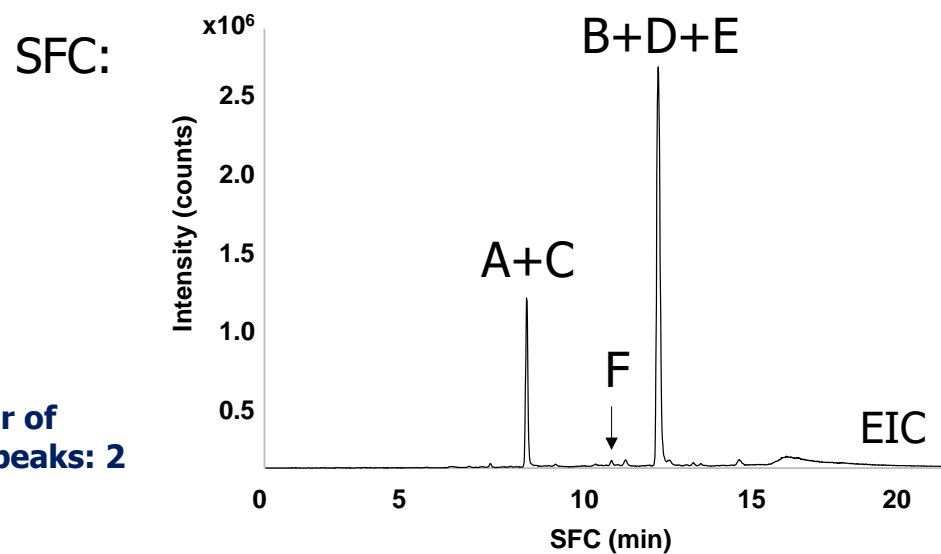
Potential structural candidates using SIRIUS annotation software



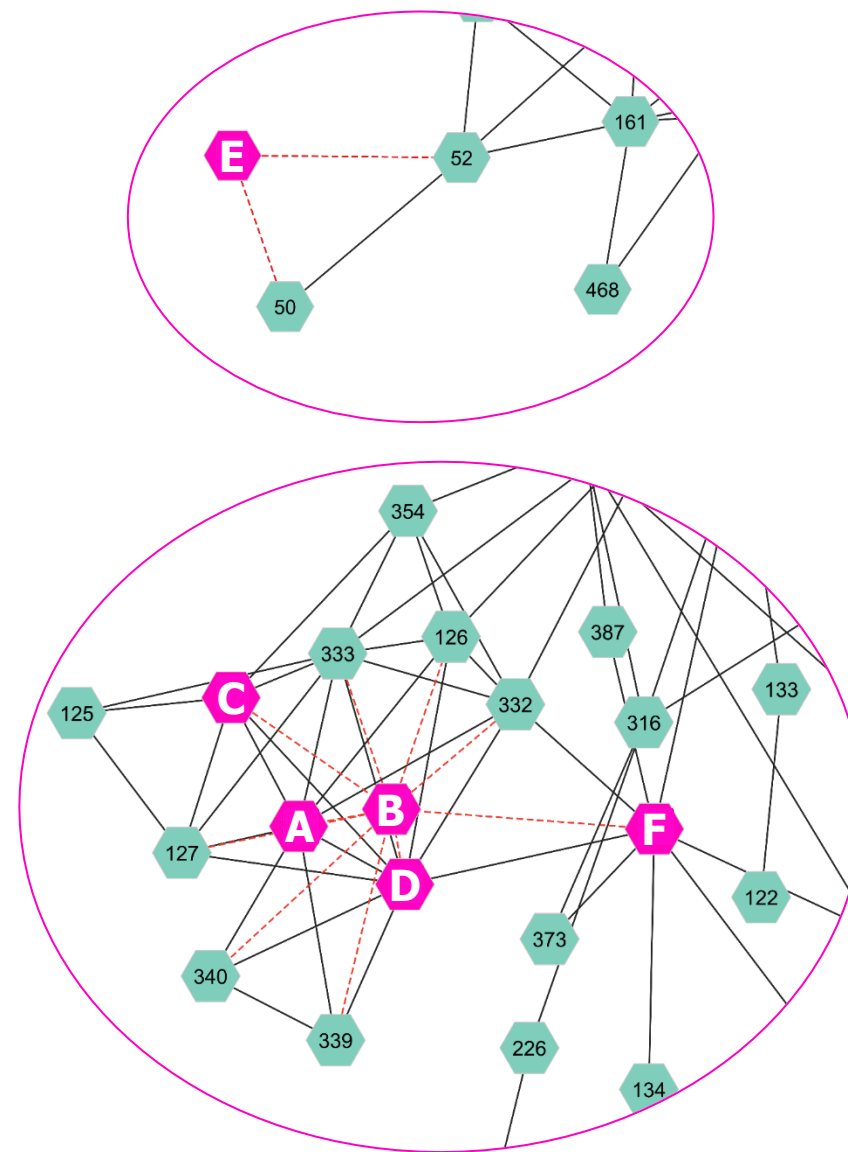
Feature-Based Molecular Network – FBMN



Number of major peaks: 4



Number of major peaks: 2



341.10263 m/z , $C_{19}H_{18}O_6$, DBE 11



✓ Compound **annotation** based on **4 criteria**

I. t_R in LC

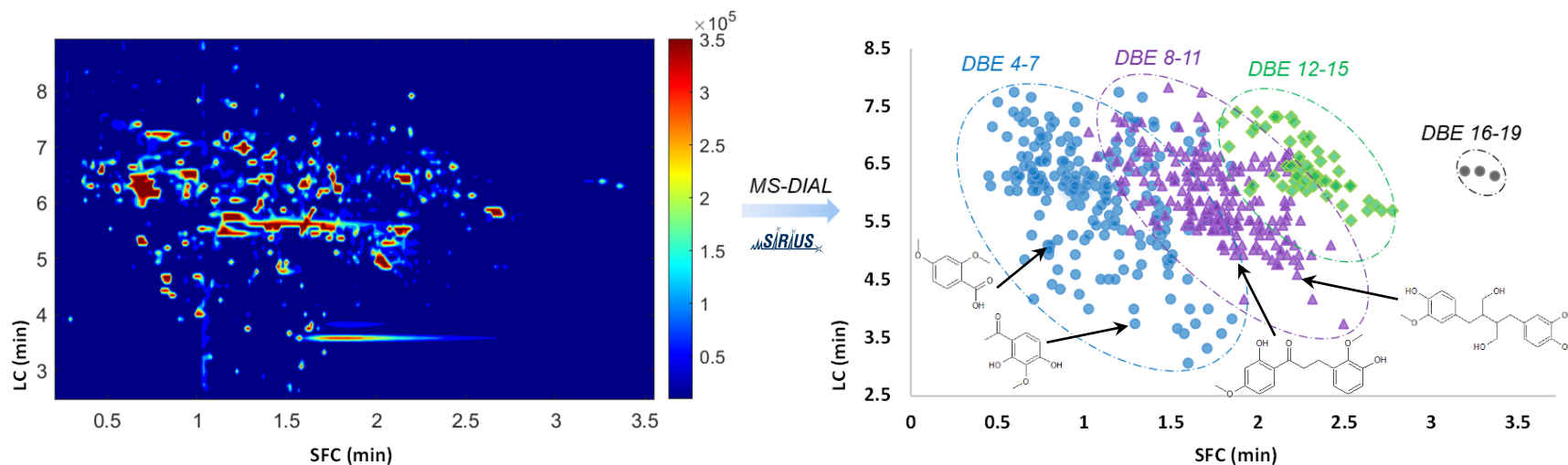
II. t_R in SFC

III. Accurate mass

IV. Fragmentation pattern

✓ Good **separation of isomers**

✓ **Grouping of lignin compounds** on the **2D plots** based on their **DBE value**



Thank you for your attention!



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Clémence Giffard
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