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by EPA Method 624.1 and 8260C.

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1. Introduction

The Clean Water Act of 1972 created the initial pathway for regulating the discharge of pollutants in water bodies in the United States. Since then, the United States Environmental Protection Agency (US EPA) has developed several analytical methods for monitoring Volatile Organic Compounds (VOCs) in water and other environmental matrices. EPA method 8260C is suitable for the analysis of VOCs in solid waste matrices. On the other hand, EPA method 624.1 is approved for analysis of purgeable organics in municipal and industrial wastewater. The standard operating procedures for both methods are similar, but the list of targeted compounds from each method includes different analytes. Overall, method 8260C is more comprehensive than method 624.1 because of its larger list of VOCs and approved sample types. While 8260C and 624.1 methods and their use on conventional GCMS have been successful, recent improvements of instrumentation require a reevaluation of the original method on newer instruments to demonstrate that the performance requirements included in these methods are met. This poster presents results from a demonstration study to determine Method Detection Limits (MDLs) for VOCs analysis by both methods 8260C and 624.1, using the newly released Shimadzu GCMS QP2020 NX and novel BFB tuning algorithm.

2. Experimental

Tuning Conditions.

A standard autotune was done prior to loading the new BFB tuning algorithm to verify the instrument operational conditions. Unlike the traditional BFB tune, the new tune algorithm makes it easier to set target intensity ratios and keep those conditions longer. Each of the three days that this MDL study was conducted, a BFB daily spectra check was conducted with respect to EPA tuning criteria.

GC-MS and Purge and Trap Conditions.

In the study, an EST Analytical Econ Evolution purge and trap concentrator and Centurion WS autosampler were interfaced Shimadzu GCMS-QP2020 NX to the (Figure 1). The experimental parameters Evolution Purge and Trap for both GC-MS and P&T systems are Concentrator. listed in Table 1



Fig 1. Shimadzu GCMS-QP2020 NX and EST Econ

Injection port mode	Calif mode 40:4 antit ratio				
	Split mode, 40:1 split ratio				
Carrier gas	Helium				
Injection port temperature (°C)	200				
Column	SH-Rxi-624Sil MS, 30 m x 0.25 mmID x 1.4 µm				
Flow control mode	Linear volcity, 32 cm/sec				
Oven Temperature	35 °C (4.0 mins.), 14 °C/mins. to 220 °C (7.0 mins.)				
Mass Spectrometer	QP2020 NX				
Interface Temperature (°C)	180				
Ion Source Temperature (°C)	200				
Detector Voltage	Relative to Tune -0.2 kV				
Threshold	100				
Scan Range	m/z 35 to 330				
	Event time 0.18 secs.				
Purge and Trap Concentrator	EST Encon Evolution and Centurion Autosampler				
Тгар	VOCARB 3000				
Trap Ready Temp (°C)	35				
Mort ready Temp (°C)	39				
Desorb Preheat Temperature (°C)	245				
Desorb Temperature (°C)	250				
Trap Bake Temperature (°C)	260				
Mort Bake Temperature	210				
Purge Flow Rate (ml/min)	Helium, 40				
Dry Purge Flow Rate (ml/min)	Helium, 40				
Desorb time (min)	1				
Bake time (min)	8				
Dry purge time (min)	2				
Purge time (min)	11				
Purge and Trap Autosampler	EST Centurion WS				
Sample loop size (ml)	5				
Sample fill mode	Loop				
Internal standard volume (µI)	5				
Surrogate standard volume (µl)	5				
Analysis Time					
GC Run Time (min)	34				

Individual stock standard solutions of analytes were prepared by dissolving the target compound in methanol, purge and trap grade, at 100 µg/ml. Internal and surrogate standards for purging were prepared at 50 µg/L.

For the MDL study that was conducted over three days, 10 replicates of spiked blank water samples were analyzed and the MDL for each compound was estimated according to procedures described in the Code of Federal Regulations To calculate the MDL, the mathematical equation listed below was used where the standard deviation was multiplied by the Student's t value for a 99% confidence level with n-1 degree of freedom.

Performance Demonstration of a Modern GC-MS Instrument and Novel BFB Tune for Analysis of Volatile Compounds

Table 1. GCMS and P&T operating conditions

Sample Preparation

MDL= (n-1, 1-α=99) S

3. Results and Discussion

Initial Calibration

In the study, a calibration curve was prepared from 0.50 to 200 µg/L. This linear range was used to estimate MDLs at both 0.5 and 1.0 µg/L. Figure 2. shows calibration curves for selected compounds in the study.



The calibration curve was evaluated according to EPA method 8260C criterion (RF %RSD < 20%) using the percent relative standard deviation (%RSD) of the calculated response factors (RF) for each data point in the curve. The method 8260C AVG RF criterion was chosen over method 624.1 criterion for demonstration of initial calibration because its list of target compounds is more comprehensive and covers all compound in this study. Greater than 90% of the compounds passed the EPA method 8260C RF criteria.

Continuing Calibration Verification

Continuing calibration verification (CCV) standards were used for the three consecutive days of the MDL study. A laboratory control sample (LCS) was prepared and was analyzed prior to running the batch on each day.

Method Detection Limit (MDL)

MDLs for each of the analytes met both EPA method 624.1 and 8260C detection limit criteria. For method 624.1, at 0.50 µg/L the MDLs ranged from 0.07 to 0.40, while at 1.00 µg/L MDLs ranged from 0.09 to 0.50 µg/L. Regarding method 8260C, at 0.50 µg/L the MDLs ranged from 0.07 to 0.40, while at 1.00 μ g/L MDLs ranged from 0.09 to 0.50 μ g/L. A list of all targeted compounds is shown in Table 2. Figure 3A illustrates MDL study results for compound listed in EPA method 8260C, while Figure 3B shows the %RSD for each compounds at the two individual spiking concentration.

Table 2.	Targeted
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Peak #	Compound Name	Peak #	Compound Name	Peak #	Compound Name	Peak #	Compound Name
1	Dichlorodifluoromethane	19	Chloroform	37	2-Hexanone	55	4-Chlorotoluene
2	Chloromethane	20	1,1,1-Trichloroethane	38	Dibromochloromethane	56	1,2,4-Trimethylbenzene
3	Vinyl chloride	21	Carbon tetrachloride	39	1,2-Dibromoethane	57	tert-Butylbenzene
4	Bromomethane	22	1,1-Dichloropropylene	40	Chlorobenzene	58	1,3,5-Trimethylbenzene
5	Chloroethane	23	Benzene	41	Ethylbenzene	59	sec-Butylbenzene
6	Trichlorofluoromethane	24	1,2-Dichloroethane	42	1,1,1,2-Tetrachloroethane	60	1,3-Dichlorobenzene
7	Acrolein	25	Trichloroethene	43	Xylene Total	61	4-Isopropyltoluene
8	1,1-Dichloroethene	26	1,2-Dichloropropane	44	m/p-Xylene	62	1,4-Dichlorobenzene
9	Acetone	27	Dibromomethane	45	o-Xylene	63	1,2-Dichlorobenzene
10	Iodomethane	28	Bromodichloromethane	46	Styrene	64	n-Butylbenzene
11	Methylene chloride	29	2-Chloroethylvinylether	47	Bromoform	65	1,2-Dibromo-3-chloropropane
12	trans-1,2-Dichloroethene	30	cis-1,3-Dichloropropene	48	Isopropylbenzene	66	1,2,3-Trichlorobenzene
13	1,1-Dichloroethane	31	4-Methyl-2-pentanone	49	1,1,2,2-Tetrachloroethane	67	Hexachlorobutadiene
14	Vinyl acetate	32	Toluene	50	Bromobenzene	68	Naphthalene
15	2-Butanone	33	trans-1,3-Dichloropropene	51	trans-1,4-Dichloro-2-butene	69	1,2,4-Trichlorobenzene
16	cis-1,2-dichloroethene	34	1,1,2-Trichloroethane	52	1,2,3-Trichloropropane		
17	2,2-Dichloropropane	35	Tetrachloroethene	53	n-Propylbenzene		
18	Bromochloromethane	36	1,3-Dichloropropane	54	2-Chlorotoluene		



Figure 3. 8260C Method Detection Limits (MDL) (Figure 3A, Top) and %RSD (Figure 3B, Bottom) study results for compounds listed in Table 3

4. Conclusion

The study demonstrates the satisfactory performance of the Shimadzu GCMS-QP2020 NX in the analysis of VOCs by EPA method 624.1/8260C. The suitability of the initial calibration curve was evaluated and results from most of the targeted compounds met the 8260C method's %RF RSD requirements (RF %RSD < 20 %). MDLs were calculated for both methods., but only MDLS for EPA 82060C is shown in this poster. Regarding method 8260C, at 0.50 µg/L the MDL ranged from 0.07 to 0.43, while at 1.00 μ g/L MDL ranged from 0.09 to 0.50 μ g/L.



compounds analyzed in the study.