

# Analysis of produced water contaminants by ion chromatography

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## Summary

The analytical challenge treated by the present work consists in detecting sub-ppm quantities of bromide, sulfate, aliphatic monocarboxylic acids and several alkaline earth metals in the presence of very high concentrations of sodium and chloride. Bromide, sulfate, acetate and butyrate can reliably be determined by suppressed conductivity detection. Due to matrix effects, propionate can only be detected qualitatively. This drawback can be overcome by coupling the ion chromatograph (IC) to a mass spectrometric (MS) detector. This results in reduced matrix interferences and significantly enhanced sensitivities.

The cations magnesium, barium and strontium are determined by non-suppressed conductivity detection.

## Introduction

Produced water is water trapped in underground reservoir rocks that is brought to the surface along with the crude oil and gas. Besides dispersed oil droplets and dissolved organic compounds, produced water mainly contains significant amounts of inorganic cations such as calcium, magnesium, barium and strontium, and anions such as carbonate, bromide and sulfate. Precipitation of the corresponding salts often leads to the formation of scale deposits that can clog pipes.

For these reasons it is of paramount importance to determine the inorganic constituents of produced water, not least for determining the correct dosing of scale inhibitors. Additionally, the detection of dissolved organic acids is of particular interest as they indicate the presence of oil.

Of the numerous analytical techniques for the determination of ionic species, ion chromatography (IC) is the best suited in terms of selectivity, flexibility, automation and ruggedness. This poster presents an overview of the different IC-based methods used for determining ionic components in produced water.

### IC/MS

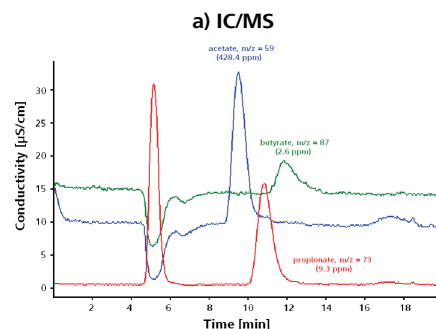
- 818 Advanced IC Pump
- 819 Advanced IC Detector
- 820 IC Separation Center
- 830 Advanced IC Interface
- 833 Advanced IC Liquid Handling Pump Unit
- 838 Advanced IC Sample Processor
- 1100 MSD SL, Agilent Technologies

### Conductivity detection

- 850 Professional IC Anion – MCS
  - 858 Professional Sample Processor – Pump
- Alternatively, all analyses can be performed with the 861 Advanced Compact IC.



## Aliphatic monocarboxylic acids



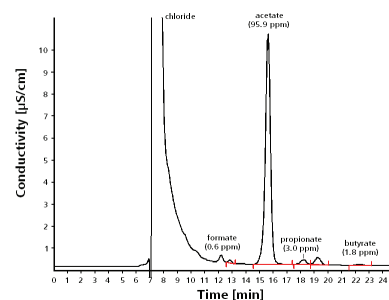
### Anion exclusion chromatography with MS detection after post-column (PC) addition of ammonia

**Sample prep.:** 1:10 dilution, dialysis  
**Column:** Metrosep Organic Acids – 100  
**Eluent:** 0.25 mmol/L oxalic acid  
**Flow:** 0.4 mL/min  
**Loop:** 10 µL

**PC solution:** 300 mmol/L ammonia  
**ESI:** 13 L/min N<sub>2</sub>, 350 °C, 60 psig, 3 kV  
**MSD:** SIM neg.  
**Fragmentor:** 70 V

Matrix interferences are reduced to an absolute minimum and carboxylic acids can be sensitively determined even in the presence of high salt concentrations.

## b) Conductivity detection

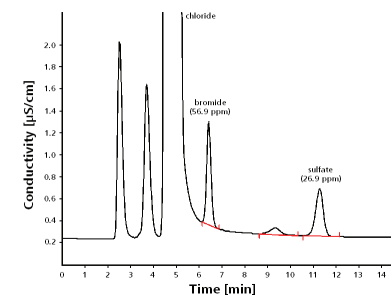


### Anion exclusion chromatography with suppressed conductivity detection

**Sample prep.:** 1:5 dilution, acidification with 0.05 mmol/L sulfuric acid  
**Eluent:** 0.5 mmol/L sulfuric acid  
**Flow:** 0.5 mL/min  
**Column:** Metrosep Organic Acids – 250  
**Precolumn:** Metrosep RP Guard  
**Loop:** 100 µL

Target compounds are well separated from the chloride peak.

## Bromide and sulfate



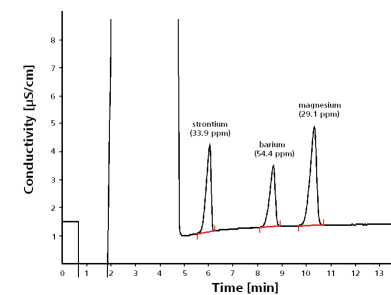
### Anion exchange chromatography with suppressed conductivity detection

**Sample prep.:** Ultrasonic treatment for 10 minutes, filtration through a 0.45 µm filter, 1:20 dilution  
**Eluent:** 1.8/1.7 mmol/L Na<sub>2</sub>CO<sub>3</sub>/NaHCO<sub>3</sub>  
**Flow:** 1.0 mL/min  
**Loop:** 20 µL

**Column:** Metrosep A Supp 4 – 250  
**Precolumn:** Metrosep RP Guard

Despite the high chloride concentration in process water samples, bromide and sulfate can be accurately determined.

## Magnesium, barium and strontium



### Cation exchange chromatography with direct conductivity detection

**Sample prep.:** Ultrasonic treatment for 10 minutes, filtration through a 0.45 µm filter, acidification with nitric acid to a pH of 2.5  
**Eluent:** 4.0 mmol/L tartaric acid, 3.0 mmol/L ethylene diamine, 0.5 mmol/L dipicolinic acid + 5% acetone  
**Flow:** 1.5 mL/min  
**Loop:** 20 µL

**Column:** Nucleosil 5SA  
**Precolumn:** Metrosep RP Guard

Since the monovalent cations elute with the injection peak, even high sodium concentrations do not interfere with the determination of low concentrated alkaline earth metal concentrations.